(19) Europäisches Patentamt
European Patent Office
Office européen des brevets

(11) Publication number: **0 648 740 A1**

(12) **EUROPEAN PATENT APPLICATION published in accordance with Art. 158(3) EPC**

(21) Application number: **94914561.9**

(22) Date of filing: **27.04.94**

(86) International application number: **PCT/JP94/00708**

(87) International publication number: **WO 94/25432 (10.11.94 94/25)**

(51) Int. Cl.[6]: **C07C 271/22**, C07C 271/32, C07C 271/54, C07C 317/14, C07C 327/20, C07D 213/28, C07D 239/32, C07D 307/04, C07D 307/40, C07D 307/77, C07D 333/14

(30) Priority: **28.04.93 JP 125455/93**

(43) Date of publication of application: **19.04.95 Bulletin 95/16**

(84) Designated Contracting States: **DE ES FR GB IT NL PT**

(71) Applicant: **KUMIAI CHEMICAL INDUSTRY CO., LTD.**
4-26, Ikenohata 1-chome
Taitoh-ku
Tokyo 110-91 (JP)
Applicant: **IHARA CHEMICAL INDUSTRY Co., Ltd.**
4-26, Ikenohata 1-chome
Taitoh-ku,
Tokyo 110 (JP)

(72) Inventor: **SHIBATA, Masaru, K. I Chem. Res. Inst. Co., Ltd**
408-1, Shioshinden,
Fukude-cho
Iwata-gun,
Shizuoka 437-12 (JP)
Inventor: **SUGIYAMA, Kazuhiko, K. I Chem. Res. Inst. Co.,Ltd**
408-01, Shioshinden,
Fukude-cho
Iwata-gun,
Shizuoka 437-12 (JP)
Inventor: **YONEKURA, Norihisa, K. I Chem. Res. Inst. Co.,Ltd**
408-01, Shioshinden,
Fukude-cho
Iwata-gun,
Shizuoka 437-12 (JP)
Inventor: **SAKAI, Junetsu**
1809, Kamo,
Kikugawa-cho
Ogasa-gun,
Shizuoka 439 (JP)
Inventor: **KOJIMA, Yoshiyuki**
69, Kohgosho
Kakegawa-shi,
Shizuoka 436 (JP)
Inventor: **HAYASHI, Shigeru**
970-1, Niino,
Hamaoka-cho
Ogasa-gun,
Shizuoka 437-16 (JP)

(74) Representative: **Hansen, Bernd, Dr. Dipl.-Chem. et al**
Hoffmann, Eitle & Partner,
Patentanwälte,
Arabellastrasse 4
D-81925 München (DE)

(54) **AMINO ACID AMIDE DERIVATIVE, AGROHORTICULTURAL BACTERICIDE, AND PRODUCTION PROCESS.**

(57) An amino acid amide derivative represented by general formula (I) and an agrohorticultural bactericide

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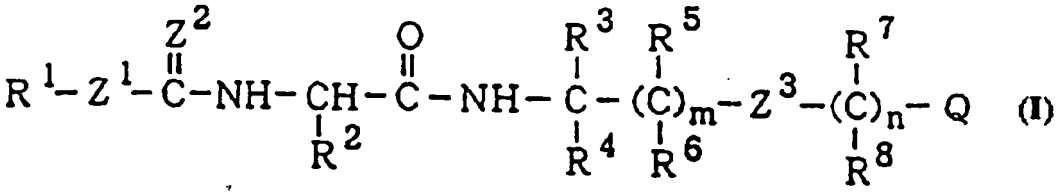
containing the same as the active ingredient. In said formula (I) R^1 represents lower alkyl which may be substituted by one or more substituents selected from the same or different halogen atoms, alkoxy groups and a cyano group; R^2 represents ethyl or n-propyl; R^3 represents hydrogen or lower alkyl; R^4 represents hydrogen; R^5, R^6 and R^7 represent each independently hydrogen or lower alkyl; R^8 represents hydrogen or lower alkyl; Z^1 and Z^2 represent each independently oxygen or sulfur; Z^3 represents oxygen or sulfur; Q represents phenyl; m represents an integer of 0 to 2; and n represents an integer of 0 or 1. This derivative has a high control effect on downy mildew of cucumber, diseases of tomato and downy mildew of grape. Further it is excellent in the capability of penetration and migration, residual effect and rainfall resistance without causing crop injury.



(I)

[Field of the Invention]

The present invention relates to an amino-acid amide derivative as well as to an agricultural or horticultural fungicide containing the same as an active ingredient. The present invention also relates to a process for preparing the same.

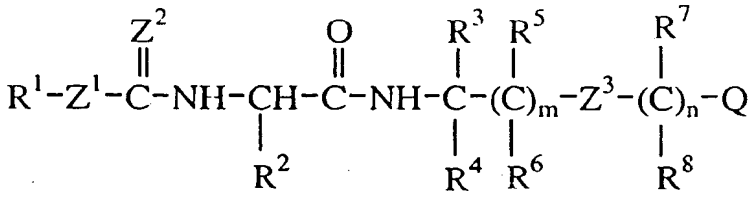
[Background of the Art]

Amino-acid amide derivatives have been disclosed as intermediates for medicines in Japanese Patent Application, First Publication Nos. Sho 56-8352 and Sho 62-89696. However, these documents fail to disclose the utility of the amino-acid amide derivatives. Although Japanese Patent Application First Publication Nos. Hei 3-5451, Hei 3-153657, Hei 4-230652, Hei 4-230653, Hei 4-283554, Hei 4-308507, and Hei 4-338372 disclose that some amino-acid amides are useful for biocides, the compounds disclosed in these documents are different from the amino-acid amide derivatives according to the present invention.

[Disclosure of the Invention]

The present inventors have synthesized various amino-acid amide derivatives and have carried out extensive research in connection with their effects on the physiological activities of fungi. As a result, we have found that the compounds according to the present invention exhibit a broad spectrum of anti-fungal activity especially against cucumber downy mildew, grape downy mildew, and tomato late blight, and at the same time they do not hinder desirable plant growth.

According to an aspect of the present invention, there is provided an amino-acid amide derivative represented by the formula:



[I]

wherein R^1 represents

a lower alkyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom, an alkoxy group, and a cyano group),

a lower alkenyl group,

a lower alkynyl group,

a cycloalkyl group (optionally having at least one same or different substituent selected from the group consisting of methyl group and a halogen atom),

a cycloalkylalkyl group,

a cycloalkenyl group,

an alkylene oxide group,

an aralkyl group (optionally having at least one same or different substituent selected from the group consisting of a methyl group, a cyano group, and a nitro group),

a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom,

a lower alkyl group which may be substituted with a same or different halogen atom,

a lower alkoxy group which may be substituted with a same or different halogen atom,

a cyano group, and

a nitro group), or

a heterocyclic group,

R^2 represents an ethyl group, an *n*-propyl group, an isopropyl group, an isobutyl group, a *sec*-butyl group, a *tert*-butyl group, an alkenyl group, a cycloalkyl group, a phenyl group (optionally having at least one substituent of halogen atom),

R^3 represents a hydrogen atom or a lower alkyl group,

R^4 represents a hydrogen atom, a lower alkyl group, or a cyano group,

R^5, R^6, and R^7 independently represent a hydrogen atom or a lower alkyl group,

R^8 represents a hydrogen atom, a lower alkyl group, an aralkyl group, a phenyl group, an alkoxycarbonyl group, or a cyano group,

Z^1 and Z^2 independently represent an oxygen atom or a sulfur atom,

Z^3 represents an oxygen atom, a sulfur atom, a group N-R^{10} (wherein R^{10} represents a hydrogen atom, a methyl group, a methylcarbonyl group, a phenylcarbonyl group, a methoxycarbonyl group, or a methoxymethyl group), a sulfinyl group, a sulfonyl group, a group COO, a group $CONR^{11}$ (wherein R^{11} represents a hydrogen atom or a lower alkyl group),

Q represents

a phenyl group [optionally having at least one same or different substituent selected from the group consisting of

a halogen atom,

a lower alkyl group which may be substituted with at least one same or different halogen atom,

a lower alkoxy group which may be substituted with a same or different halogen atom,

a cyano group,

a nitro group,

a lower alkoxycarbonyl group,

a methylsulfonyl group,

a methylsulfinyl group,

a methylthio group which may be substituted with a halogen atom,

a dimethylamino group,

a phenylsulfonyl group,

an acyl group, and a phenyl group],

an alkylene oxide group,

a heterocyclic group (optionally having a substituent selected from the group consisting of a halogen atom, an alkyl group, a trifluoromethyl group, and a nitro group), or

a condensed heterocyclic group optionally having a substituent selected from the group consisting of a halogen atom and a nitro group,

m represents an integer from 0 to 2, and

n represents 0 or 1,

and an agricultural or horticultural fungicide containing the same as the active ingredient.

The terms employed in the present invention are defined as follows. The term "alkyl group" is used herein to mean a straight or branched alkyl group possessing 1 to 6 carbon atoms including, but not limited to, a methyl group, ethyl group, *n*-propyl group, isopropyl group, *n*-butyl group, isobutyl group, *sec*-butyl group, *tert*-butyl group, *n*-pentyl group, 1-methylbutyl group, 2-methylbutyl group, 3-methylbutyl group, 2,2-dimethylpropyl group, 1,1-dimethylpropyl group, 1-ethylpropyl group, *n*-hexyl group, or the like.

The term "halogen atom" is used herein to mean a fluorine atom, chlorine atom, bromine atom, or iodine atom.

The term "lower alkenyl group" is used herein to mean a straight or branched alkenyl group possessing 2 to 6 carbon atoms and including, but not limited to, a vinyl group, 1-propenyl group, 2-propenyl group, isopropenyl group, 1-butenyl group, 2-butenyl group, 3-butenyl group, 1-methyl-1-propenyl group, 2-methylpropenyl group, 1-ethylvinyl group, or the like.

The term "lower alkynyl group" is used herein to mean a straight or branched alkynyl group possessing 2 to 6 carbon atoms and including, for example, an ethynyl group, propynyl group, butynyl group, 1-methyl-2-propynyl group, or the like.

The term "cycloalkyl group" is used herein to mean a cycloalkyl group possessing 3 to 8 carbon atoms and including, but not limited to, a cyclopropyl group, cyclobutyl group, cyclopentyl group, cyclohexyl group, cycloheptyl group, or the like.

The term "cycloalkenyl group" is used herein to mean a cycloalkenyl group possessing 4 to 8 carbon atoms and including, for example, a cyclobutenyl group, cyclopentenyl group, cyclohexenyl group, cycloheptenyl group, or the like.

The term "aralkyl group" is used herein to mean an aralkyl group possessing 7 to 8 carbon atoms and including, but not limited to, a benzyl group, phenethyl group, or the like.

The term "alkylene oxide group" is used herein to mean an alkylene oxide group possessing 2 to 6 carbon atoms and including, for example, an oxiranyl group, oxetanyl group, tetrahydrofuranyl group, tetrahydropyranyl group, or the like.

The preferred compounds of the present invention are represented by formula [I], wherein R^1 represents a straight or branched alkyl group possessing 2 to 6 carbon atoms, a straight or branched alkenyl group possessing 3 carbon atoms, a cycloalkyl group possessing 5 to 6 carbon atoms, or a phenyl group optionally having a substituent; R^2 represents an ethyl group, an *n*-propyl group, an isopropyl group, or a *sec*-butyl group; R^3 represents a hydrogen atom or a methyl group; R^4 represents a hydrogen atom or a methyl group; R^5 represents a hydrogen atom or a methyl group; R^6 represents a hydrogen atom or a methyl group; Q represents a phenyl group optionally having a substituent; m represents an integer of 0 or 1; n represents 0; Z^1, Z^2, and Z^3 represent an oxygen atom or a sulfur atom; and the amino acid is an L-isomer.

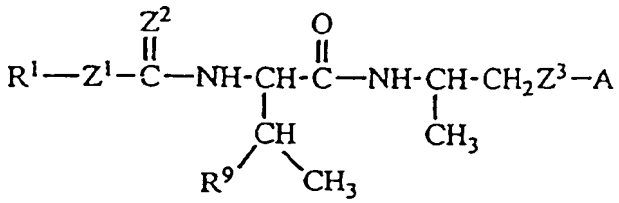
The compounds represented by formula [I] according to the present invention can exist in stereoisomers by virtue of the presence of two or more chiral centers. The present invention relates to all such stereoisomers, including diastereomers, enantiomers, and mixtures thereof, which can be separated by appropriate methods.

Next, the compounds represented by formula [I] according to the present invention are listed in Tables 1 to 12. However, it should be understood that the invention is not limited to these compounds. The compound Numbers. given in Tables 1 to 12 will be referred to in the subsequent description.

In Tables 1 to 12, Compound Nos. 108, 433, 456, 459, 460, 461, 462, 464, 467, 470, 471, 472, and 475 possess D,L-configurational amino acid moieties; Compound No. 109 possesses a D-configurational amino acid moiety; Compound Nos. 233, 234, 235, 236, 237, 238, 425, 426, 427 possess (2S)-butyric acid moieties; and the compounds other than the compounds described above possess L-configurational amino acid moieties. Compound Nos. 33, 345, and 346; Compound Nos. 107. 116, and 117; Compound Nos. 135, 395, and 396; Compound Nos. 228, 414, and 415; and Compound Nos. 452, 453, and 454 are mixtures of diastereomers, and are also individual diastereomers. In addition, Compound Nos. 26 and 27; Compound Nos. 45 and 356; Compound Nos. 335 and 336; Compound Nos. 397 and 401; and Compound Nos. 409 and 410 are mixtures of diastereomers, and are also one of the individual diastereomers, respectively. Compound No. 108 is a mixture of four isomers and Compound No. 433 is a mixture of two isomers. Compound Nos. 483 to 501, 504, 505, 510 to 518, 521, and 522 are L-Val-DL-Ala; Compound Nos. 502, 503, 508, 509, 519, and 525 are L-Val-D-Ala; Compound No. 520 is L-Val-L-Ala; Compound Nos. 506 and 523 are L-Ile-D-Ala; Compound No. 526 is L-Val-Gly; and Compound Nos. 507 and 524 are (2S)-butylyl-D-Ala.

In the tables of the present specification, the expressions "C_3H_7-i", "C_4H_9-t", "C_4H_9-s", and "C_4H_9-i" are used to indicate an isopropyl group, a *tert*-butyl group, a *sec*-butyl group, and an isobutyl group, respectively.

Table 1



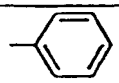
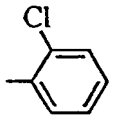
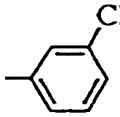
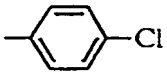
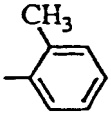
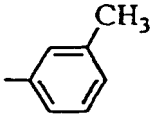
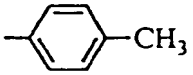
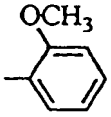
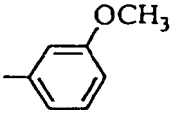
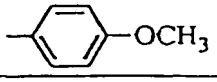
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
1	C_4H_9-t	CH_3	O	O	O		88-92
2	C_4H_9-t	CH_3	O	O	O		98-100
3	C_4H_9-t	CH_3	O	O	O		1.5051
4	C_4H_9-t	CH_3	O	O	O		97-98
5	C_4H_9-t	CH_3	O	O	O		77-80
6	C_4H_9-t	CH_3	O	O	O		1.5051
7	C_4H_9-t	CH_3	O	O	O		99-101
8	C_4H_9-t	CH_3	O	O	O		86-89
9	C_4H_9-t	CH_3	O	O	O		1.4899
10	C_4H_9-t	CH_3	O	O	O		86-89

Table 1 (continued)

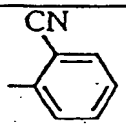
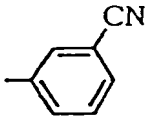
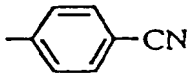
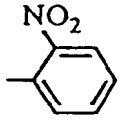
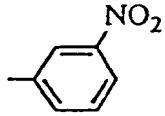
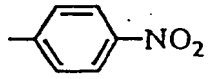
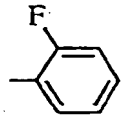
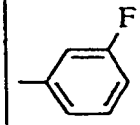
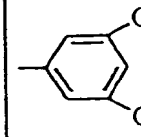
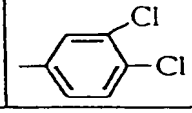
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
11	C_4H_9-t	CH_3	O	O	O		
12	C_4H_9-t	CH_3	O	O	O		83-87
13	C_4H_9-t	CH_3	O	O	O		53-56
14	C_4H_9-t	CH_3	O	O	O		1.5081
15	C_4H_9-t	CH_3	O	O	O		112-114
16	C_4H_9-t	CH_3	O	O	O		105-107
17	C_4H_9-t	CH_3	O	O	O		95-97
18	C_4H_9-t	CH_3	O	O	O		89-92
19	C_4H_9-t	CH_3	O	O	O		85-89
20	C_4H_9-t	CH_3	O	O	O		99-100
21	C_4H_9-t	CH_3	O	O	O		102-104

Table 1 (continued)

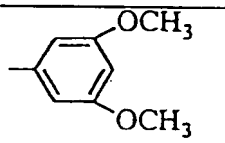
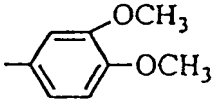
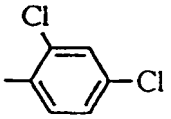
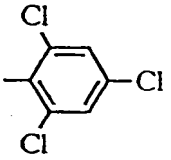
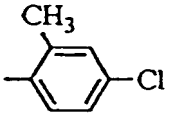
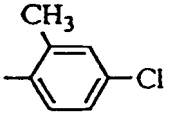
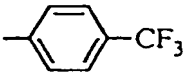
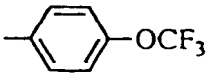
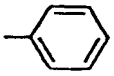
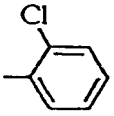
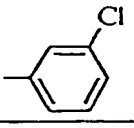
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
22	C_4H_9-t	CH_3	O	O	O		87-91
23	C_4H_9-t	CH_3	O	O	O		88-90
24	C_4H_9-t	CH_3	O	O	O		98-103
25	C_4H_9-t	CH_3	O	O	O		120-125
26	C_4H_9-t	CH_3	O	O	O		108-110
27	C_4H_9-t	CH_3	O	O	O		143-146
28	C_4H_9-t	CH_3	O	O	O		115-117
29	C_4H_9-t	CH_3	O	O	O		94-98
30	C_3H_7-i	CH_3	O	O	O		
31	C_3H_7-i	CH_3	O	O	O		
32	C_3H_7-i	CH_3	O	O	O		

Table 1 (continued)

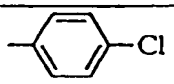
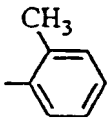
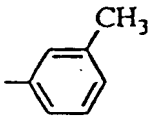
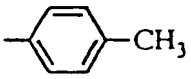
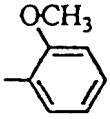
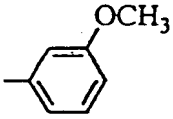
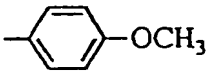
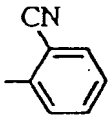
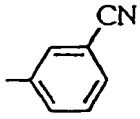
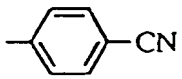
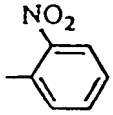
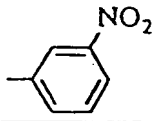
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
33	C_3H_7-i	CH_3	O	O	O		149-152
34	C_3H_7-i	CH_3	O	O	O		
35	C_3H_7-i	CH_3	O	O	O		
36	C_3H_7-i	CH_3	O	O	O		
37	C_3H_7-i	CH_3	O	O	O		
38	C_3H_7-i	CH_3	O	O	O		
39	C_3H_7-i	CH_3	O	O	O		
40	C_3H_7-i	CH_3	O	O	O		
41	C_3H_7-i	CH_3	O	O	O		
42	C_3H_7-i	CH_3	O	O	O		149-152
43	C_3H_7-i	CH_3	O	O	O		
44	C_3H_7-i	CH_3	O	O	O		

Table 1 (continued)

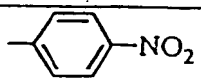
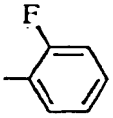
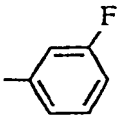
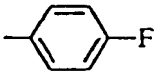
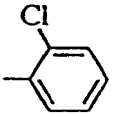
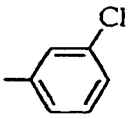
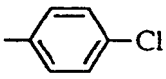
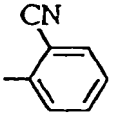
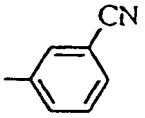
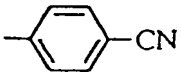
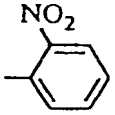
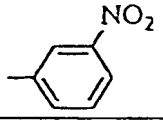
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
45	C_3H_7-i	CH_3	O	O	O		not determined
46	C_3H_7-i	CH_3	O	O	O		
47	C_3H_7-i	CH_3	O	O	O		
48	C_3H_7-i	CH_3	O	O	O		
49	C_2H_5	CH_3	O	O	O		
50	C_2H_5	CH_3	O	O	O		
51	C_2H_5	CH_3	O	O	O		
52	C_2H_5	CH_3	O	O	O		
53	C_2H_5	CH_3	O	O	O		
54	C_2H_5	CH_3	O	O	O		112-115
55	C_2H_5	CH_3	O	O	O		
56	C_2H_5	CH_3	O	O	O		

Table 1 (continued)

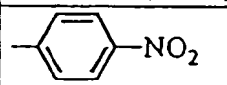
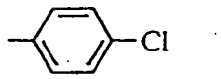
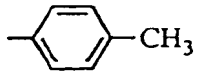
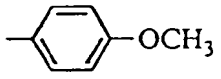
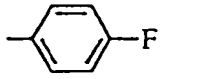
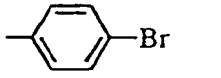
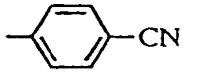
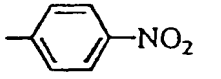
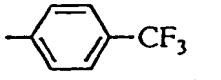
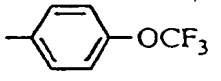
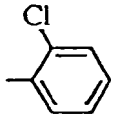
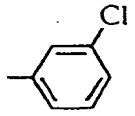
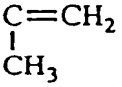
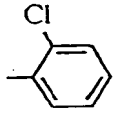
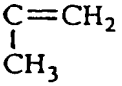
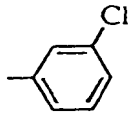
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
57	C_2H_5	CH_3	O	O	O		
58	C_4H_9-s	CH_3	O	O	O		
59	C_4H_9-s	CH_3	O	O	O		
60	C_4H_9-s	CH_3	O	O	O		
61	C_4H_9-s	CH_3	O	O	O		
62	C_4H_9-s	CH_3	O	O	O		
63	C_4H_9-s	CH_3	O	O	O		140-143
64	C_4H_9-s	CH_3	O	O	O		
65	C_4H_9-s	CH_3	O	O	O		
66	C_4H_9-s	CH_3	O	O	O		
67	C_4H_9-s	CH_3	O	O	O		
68	C_4H_9-s	CH_3	O	O	O		
69		CH_3	O	O	O		
70		CH_3	O	O	O		

Table 1 (continued)

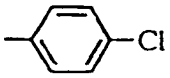
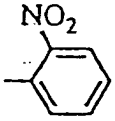
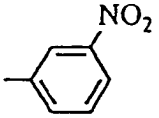
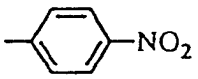
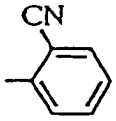
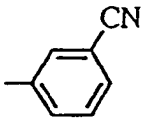
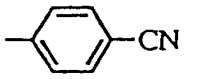
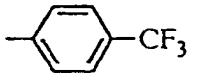
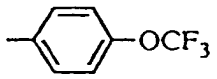
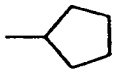
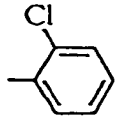
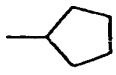
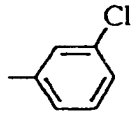
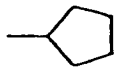
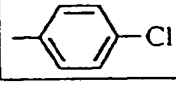
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
71	$-C(CH_3)=CH_2$	CH_3	O	O	O		
72	$-C(CH_3)=CH_2$	CH_3	O	O	O		
73	$-C(CH_3)=CH_2$	CH_3	O	O	O		
74	$-C(CH_3)=CH_2$	CH_3	O	O	O		
75	$-C(CH_3)=CH_2$	CH_3	O	O	O		
76	$-C(CH_3)=CH_2$	CH_3	O	O	O		
77	$-C(CH_3)=CH_2$	CH_3	O	O	O		82-86
78	$-C(CH_3)=CH_2$	CH_3	O	O	O		
79	$-C(CH_3)=CH_2$	CH_3	O	O	O		
80		CH_3	O	O	O		
81		CH_3	O	O	O		
82		CH_3	O	O	O		

Table 1 (continued)

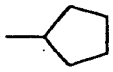
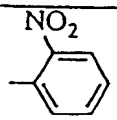
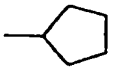
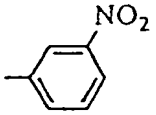
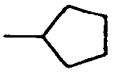
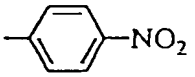
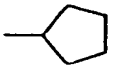
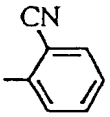
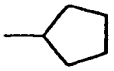
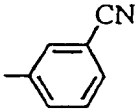
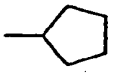
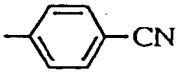
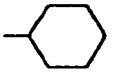
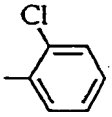
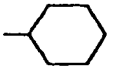
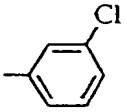
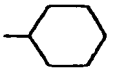
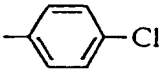
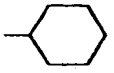
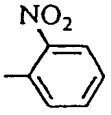
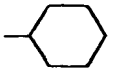
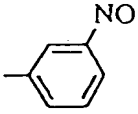
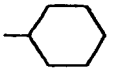
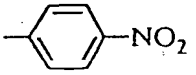
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
83		CH_3	O	O	O		
84		CH_3	O	O	O		
85		CH_3	O	O	O		
86		CH_3	O	O	O		
87		CH_3	O	O	O		
88		CH_3	O	O	O		145-148
89		CH_3	O	O	O		
90		CH_3	O	O	O		
91		CH_3	O	O	O		
92		CH_3	O	O	O		
93		CH_3	O	O	O		
94		CH_3	O	O	O		

Table 1 (continued)

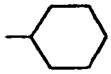
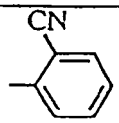
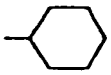
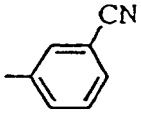
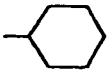
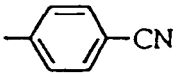
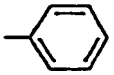
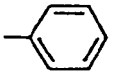
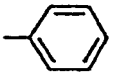
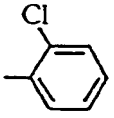
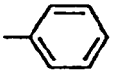
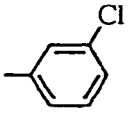
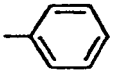
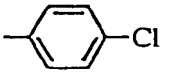
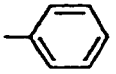
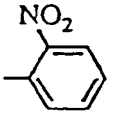
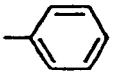
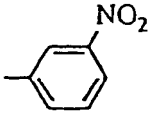
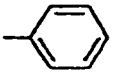
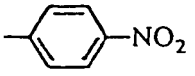
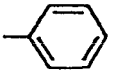
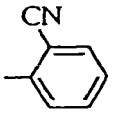
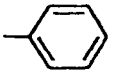
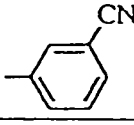
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Refractive Index (n_D^{20})
95		CH_3	O	O	O		
96		CH_3	O	O	O		
97		CH_3	O	O	O		158-162
98		CH_3	O	O	O		123-126
99		CH_3	O	O	O		
100		CH_3	O	O	O		
101		CH_3	O	O	O		165-170
102		CH_3	O	O	O		
103		CH_3	O	O	O		
104		CH_3	O	O	O		166-169
105		CH_3	O	O	O		
106		CH_3	O	O	O		

Table 1 (continued)

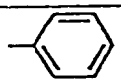
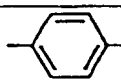
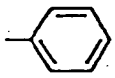
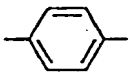
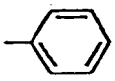
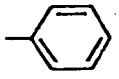
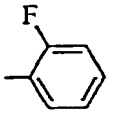
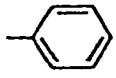
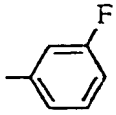
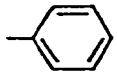
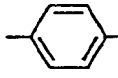
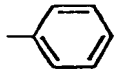
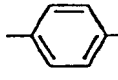
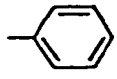
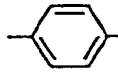
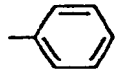
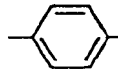
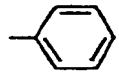
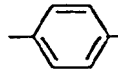
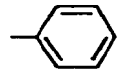
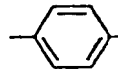
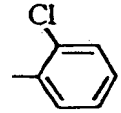
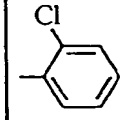
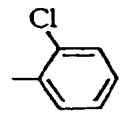
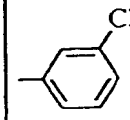
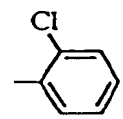
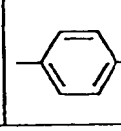
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
107		CH_3	O	O	O		142-146
108		CH_3	O	O	O		158-162
109		CH_3	O	O	O		128-133
110		CH_3	O	O	O		
111		CH_3	O	O	O		
112		CH_3	O	O	O		137-142
113		CH_3	O	O	O		
114		CH_3	O	O	O		151-155
115		CH_3	O	O	O		144-147
116		CH_3	O	O	O		145-147
117		CH_3	O	O	O		166-170
118		CH_3	O	O	O		
119		CH_3	O	O	O		
120		CH_3	O	O	O		

Table 1 (continued)

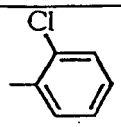
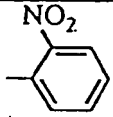
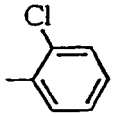
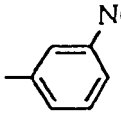
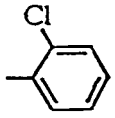
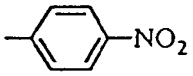
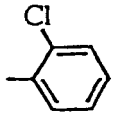
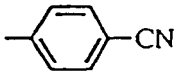
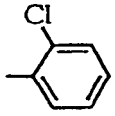
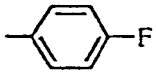
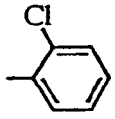
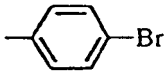
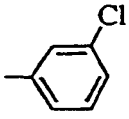
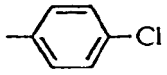
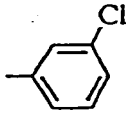
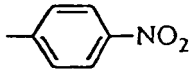
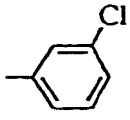
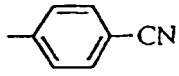
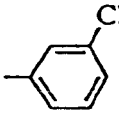
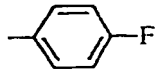
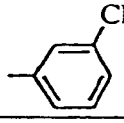
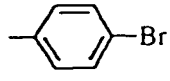
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
121		CH_3	O	O	O		
122		CH_3	O	O	O		
123		CH_3	O	O	O		
124		CH_3	O	O	O		137-142
125		CH_3	O	O	O		
126		CH_3	O	O	O		
127		CH_3	O	O	O		
128		CH_3	O	O	O		
129		CH_3	O	O	O		114-117
130		CH_3	O	O	O		
131		CH_3	O	O	O		

Table 1 (continued)

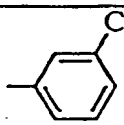
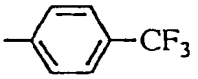
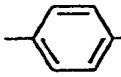
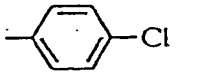
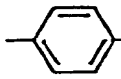
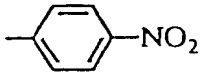
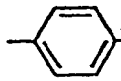
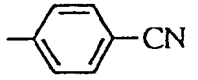
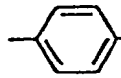
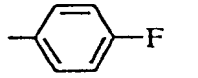
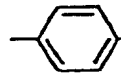
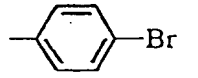
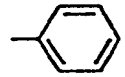
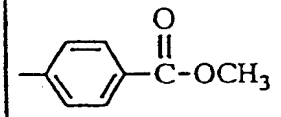
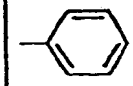
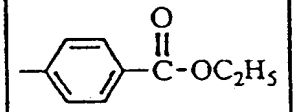
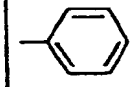
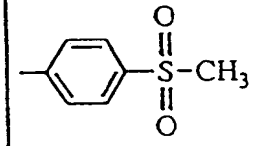
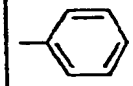
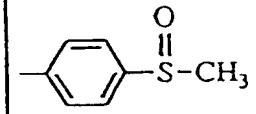
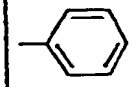
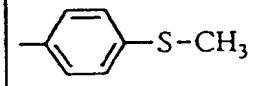
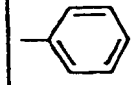
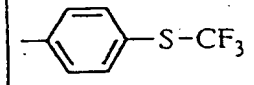
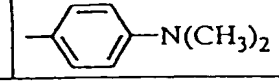
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Refractive Index (n_D^{20})
132		CH_3	O	O	O		
133		CH_3	O	O	O		
134		CH_3	O	O	O		133-138
135		CH_3	O	O	O		156-160
136		CH_3	O	O	O		
137		CH_3	O	O	O		
138		CH_3	O	O	O		
139		CH_3	O	O	O		
140		CH_3	O	O	O		
141		CH_3	O	O	O		
142		CH_3	O	O	O		
143		CH_3	O	O	O		
144		CH_3	O	O	O		

Table 1 (continued)

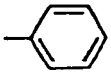
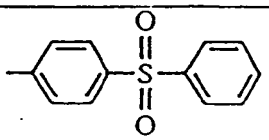
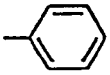
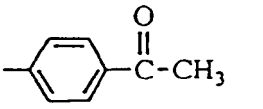
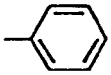
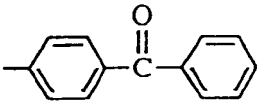
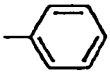
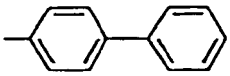
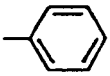
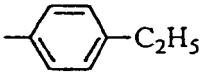
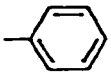
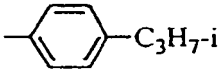
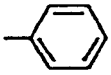
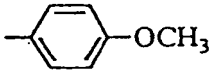
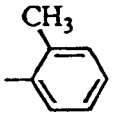
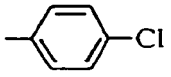
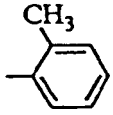
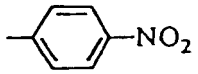
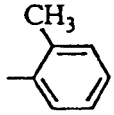
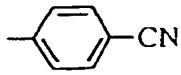
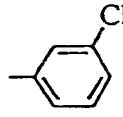
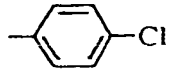
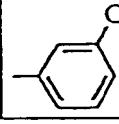
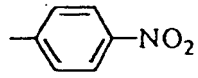
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
145		CH_3	O	O	O		
146		CH_3	O	O	O		
147		CH_3	O	O	O		
148		CH_3	O	O	O		
149		CH_3	O	O	O		
150		CH_3	O	O	O		
151		CH_3	O	O	O		
152		CH_3	O	O	O		
153		CH_3	O	O	O		
154		CH_3	O	O	O		146-150
155		CH_3	O	O	O		
156		CH_3	O	O	O		

Table 1 (continued)

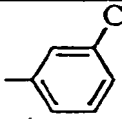
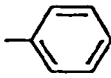
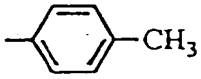
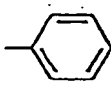
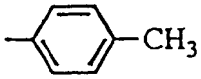
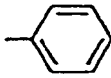
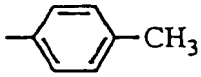
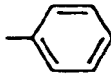
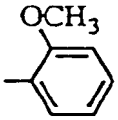
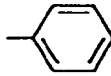
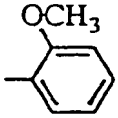
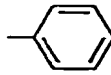
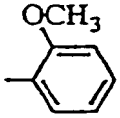
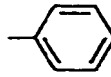
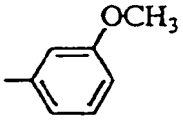
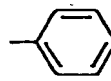
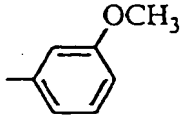
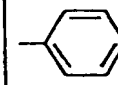
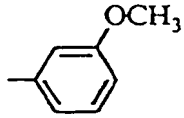
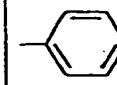
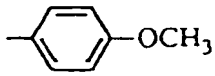
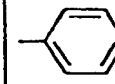
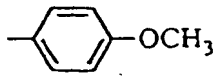
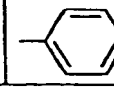
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Refractive Index (n_D^{20})
157		CH_3	O	O	O		97-100
158		CH_3	O	O	O		
159		CH_3	O	O	O		
160		CH_3	O	O	O		152-155
161		CH_3	O	O	O		
162		CH_3	O	O	O		
163		CH_3	O	O	O		137-140
164		CH_3	O	O	O		
165		CH_3	O	O	O		
166		CH_3	O	O	O		134-137
167		CH_3	O	O	O		
168		CH_3	O	O	O		

Table 1 (continued)

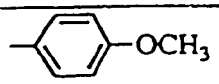
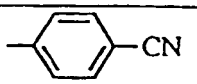
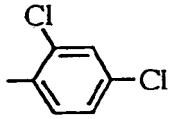
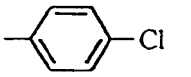
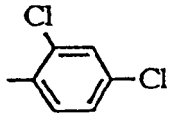
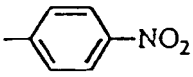
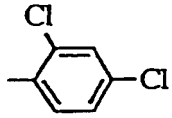
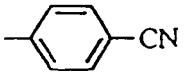
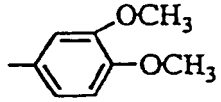
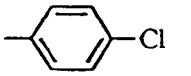
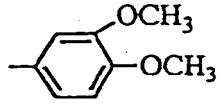
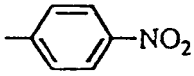
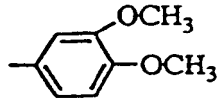
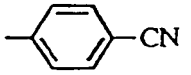
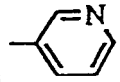
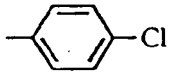
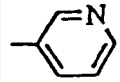
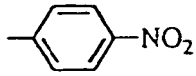
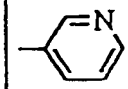
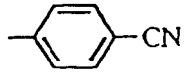
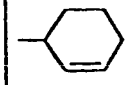
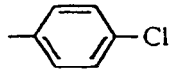
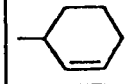
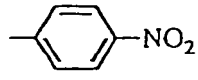
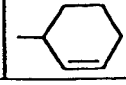
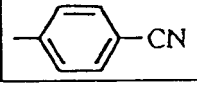
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
169		CH_3	O	O	O		139-145
170		CH_3	O	O	O		
171		CH_3	O	O	O		
172		CH_3	O	O	O		
173		CH_3	O	O	O		
174		CH_3	O	O	O		
175		CH_3	O	O	O		
176		CH_3	O	O	O		
177		CH_3	O	O	O		
178		CH_3	O	O	O		
179		CH_3	O	O	O		
180		CH_3	O	O	O		
181		CH_3	O	O	O		

Table 1 (continued)

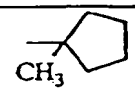
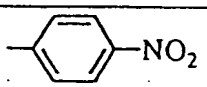
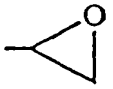
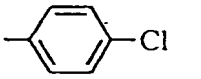
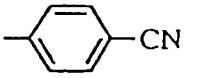
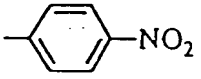
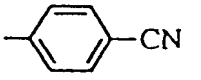
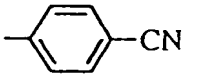
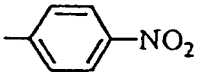
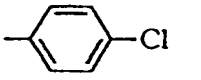
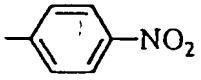
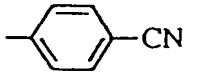
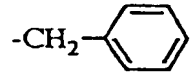
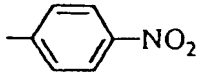
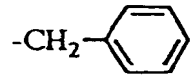
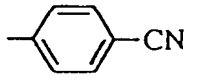
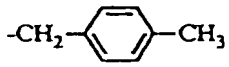
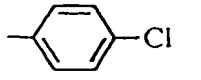
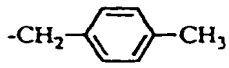
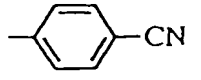
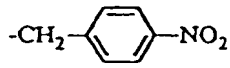
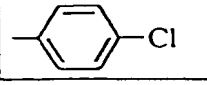
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
182		CH_3	O	O	O		
183		CH_3	O	O	O		
184	CH_2CH_2Cl	CH_3	O	O	O		170-175
185	CH_2Cl	CH_3	O	O	O		
186	$CH(Cl)CH_3$	CH_3	O	O	O		
187	CH_2CF_3	CH_3	O	O	O		
188	$CH_2-C\equiv CH$	CH_3	O	O	O		
189	$CH_2CH_2OCH_3$	CH_3	O	O	O		
190	$CH_2CH_2OCH_3$	CH_3	O	O	O		
191	$CH_2CH_2OCH_3$	CH_3	O	O	O		
192		CH_3	O	O	O		
193		CH_3	O	O	O		125-128
194		CH_3	O	O	O		
195		CH_3	O	O	O		98-101
196		CH_3	O	O	O		

Table 1 (continued)

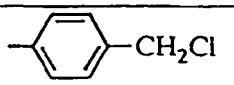
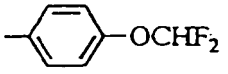
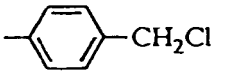
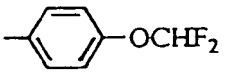
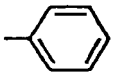
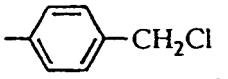
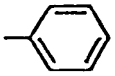
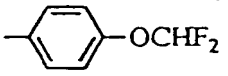
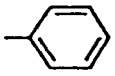
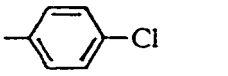
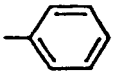
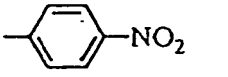
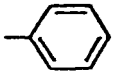
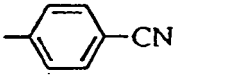
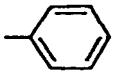
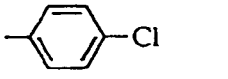
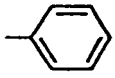
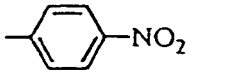
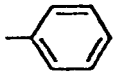
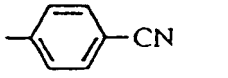
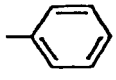
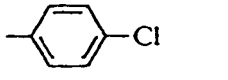
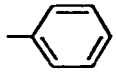
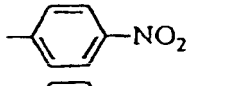
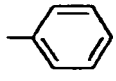
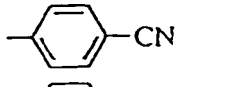
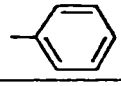
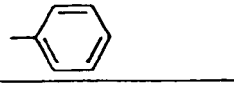
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
197	C_4H_9-t	CH_3	O	O	O		
198	C_4H_9-t	CH_3	O	O	O		
199	C_3H_7-i	CH_3	O	O	O		
200	C_3H_7-i	CH_3	O	O	O		
201		CH_3	O	O	O		
202		CH_3	O	O	O		
203		CH_3	S	O	O		111-113
204		CH_3	S	O	O		149-152
205		CH_3	S	O	O		146-149
206		CH_3	O	S	O		
207		CH_3	O	S	O		
208		CH_3	O	S	O		not determined
209		CH_3	S	S	O		
210		CH_3	S	S	O		
211		CH_3	S	S	O		not determined
212		CH_3	O	O	S		140-144

Table 1 (continued)

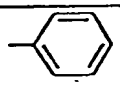
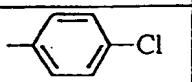
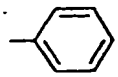
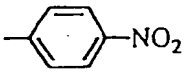
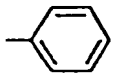
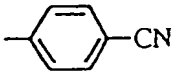
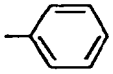
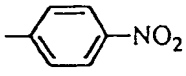
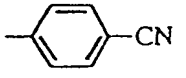
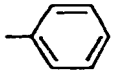
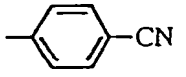
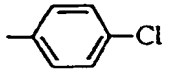
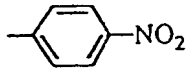
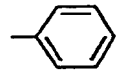
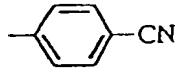
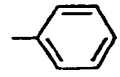
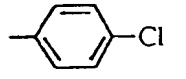
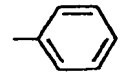
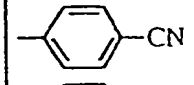
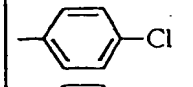
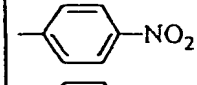
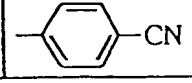
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
213		CH_3	O	O	S		136-140
214		CH_3	O	O	S		123-126
215		CH_3	O	O	S		144-146
216	C_4H_9-t	CH_3	O	O	S		74-78
217	C_4H_9-t	CH_3	O	O	S		109-112
218	C_4H_9-t	CH_3	O	O	S		
219	C_3H_7-i	CH_3	O	O	S		122-126
220	C_3H_7-i	CH_3	O	O	S		165-169
221	C_3H_7-i	CH_3	O	O	NH		158-160
222	C_4H_9-t	CH_3	O	O	NCH_3		
223		CH_3	O	O	$NCOCH_3$		
224		CH_3	O	O	NCO- 		
225		CH_3	O	O	NCO_2CH_3		
226	C_4H_9-t	C_2H_5	O	O	O		101-104
227	C_4H_9-t	C_2H_5	O	O	O		128-130
228	C_4H_9-t	C_2H_5	O	O	O		100-106

Table 1 (continued)

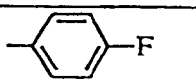
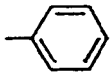
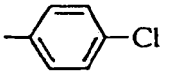
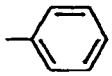
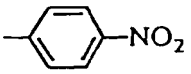
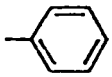
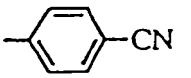
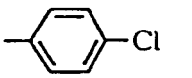
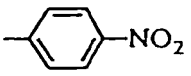
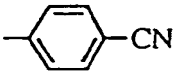
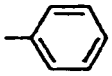
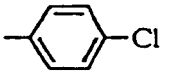
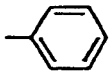
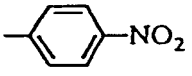
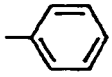
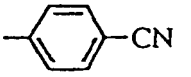
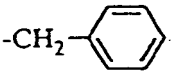
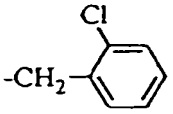
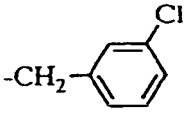
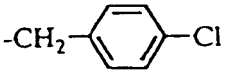
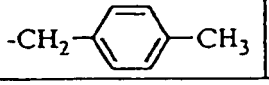
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
229	C_4H_9-t	C_2H_5	O	O	O		
230		C_2H_5	O	O	O		149-154
231		C_2H_5	O	O	O		152-154
232		C_2H_5	O	O	O		108-112
233	C_4H_9-t	H	O	O	O		1.5081
234	C_4H_9-t	H	O	O	O		
235	C_4H_9-t	H	O	O	O		not determined
236		H	O	O	O		125-130
237		H	O	O	O		
238		H	O	O	O		43-46
239	C_4H_9-t	CH_3	O	O	O		
240	C_4H_9-t	CH_3	O	O	O		
241	C_4H_9-t	CH_3	O	O	O		
242	C_4H_9-t	CH_3	O	O	O		
243	C_4H_9-t	CH_3	O	O	O		

Table 1 (continued)

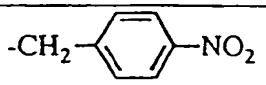
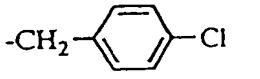
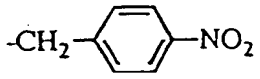
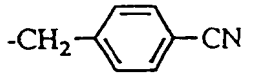
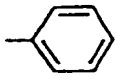
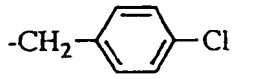
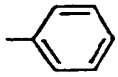
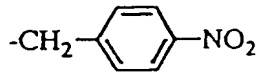
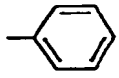
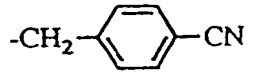
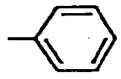
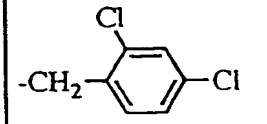
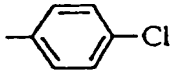
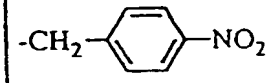
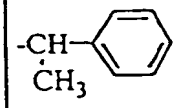
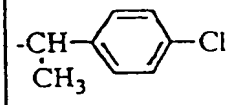
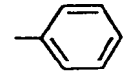
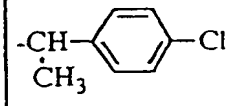
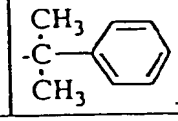
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
244	C_4H_9-t	CH_3	O	O	O		
245	C_4H_9-t	CH_3	O	O	O		
246	C_3H_7-i	CH_3	O	O	O		104-109
247	C_3H_7-i	CH_3	O	O	O		
248	C_3H_7-i	CH_3	O	O	O		
249		CH_3	O	O	O		
250		CH_3	O	O	O		
251		CH_3	O	O	O		
252		CH_3	O	O	O		
253		CH_3	O	O	O		
254	C_3H_7-i	CH_3	O	O	O		
255	C_3H_7-i	CH_3	O	O	O		
256		CH_3	O	O	O		
257	C_3H_7-i	CH_3	O	O	O		

Table 1 (continued)

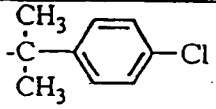
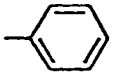
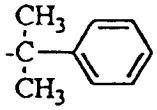
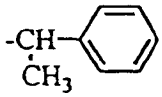
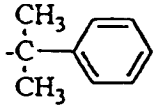
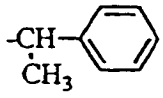
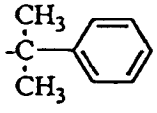
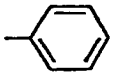
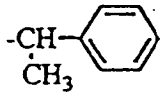
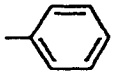
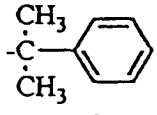
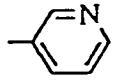
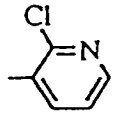
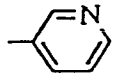
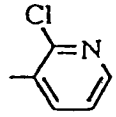
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
258	C_3H_7-i	CH_3	O	O	O		
259		CH_3	O	O	O		
260	C_4H_9-t	CH_3	O	O	NH		
261	C_4H_9-t	CH_3	O	O	NH		
262	C_3H_7-i	CH_3	O	O	NH		
263	C_3H_7-i	CH_3	O	O	NH		
264		CH_3	O	O	NH		
265		CH_3	O	O	NH		
266	C_4H_9-t	CH_3	O	O	O		
267	C_4H_9-t	CH_3	O	O	O		
268	C_3H_7-i	CH_3	O	O	O		
269	C_3H_7-i	CH_3	O	O	O		

Table 1 (continued)

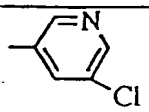
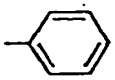
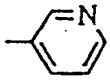
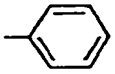
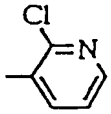
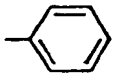
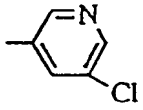
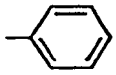
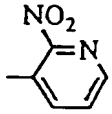
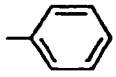
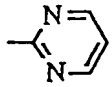
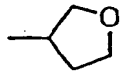
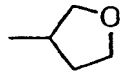
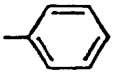
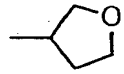
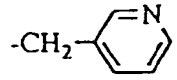
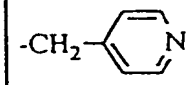
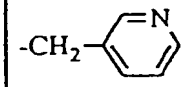
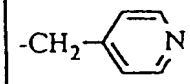
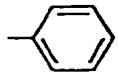
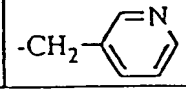
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
270	C_3H_7-i	CH_3	O	O	O		
271		CH_3	O	O	O		
272		CH_3	O	O	O		
273		CH_3	O	O	O		
274		CH_3	O	O	O		
275		CH_3	O	O	O		60-65
276	C_4H_9-t	CH_3	O	O	O		
277	C_3H_7-i	CH_3	O	O	O		
278		CH_3	O	O	O		
279	C_4H_9-t	CH_3	O	O	O		
280	C_4H_9-t	CH_3	O	O	O		
281	C_3H_7-i	CH_3	O	O	O		
282	C_3H_7-i	CH_3	O	O	O		
283		CH_3	O	O	O		

Table 1 (continued)

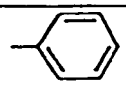
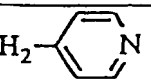
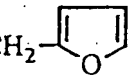
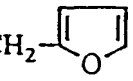
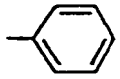
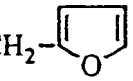
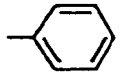
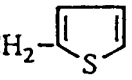
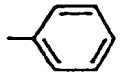
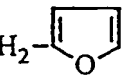
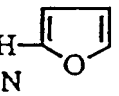
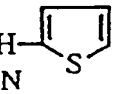
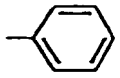
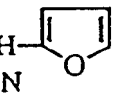
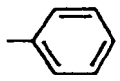
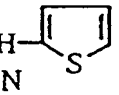
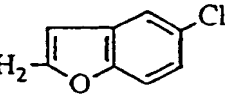
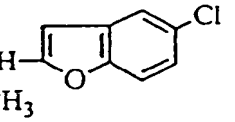
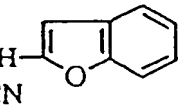
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
284		CH_3	O	O	O		
285	C_3H_7-i	CH_3	O	O	O		
286	C_3H_7-i	CH_3	O	O	S		
287		CH_3	O	O	O		
288		CH_3	O	O	S		
289		CH_3	O	O	NH		
290	C_3H_7-i	CH_3	O	O	NH		
291	C_3H_7-i	CH_3	O	O	NH		
292		CH_3	O	O	NH		
293		CH_3	O	O	NH		
294	C_3H_7-i	CH_3	O	O	O		
295	C_3H_7-i	CH_3	O	O	O		
296	C_3H_7-i	CH_3	O	O	O		

Table 1 (continued)

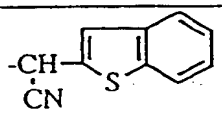
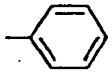
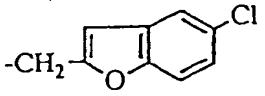
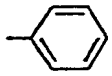
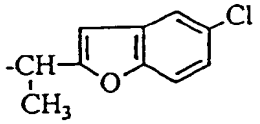
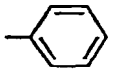
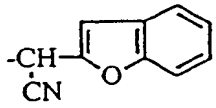
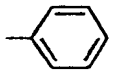
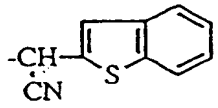
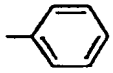
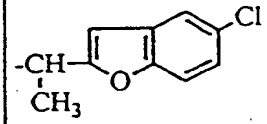
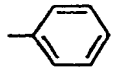
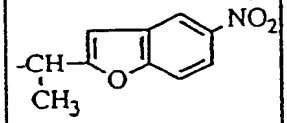
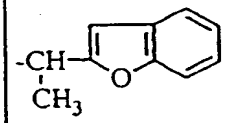
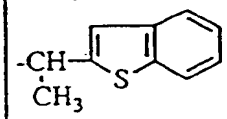
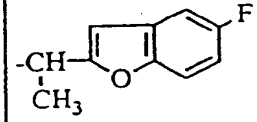
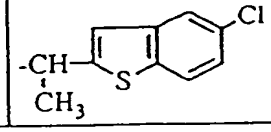
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
297	C_3H_7-i	CH_3	O	O	NH		
298		CH_3	O	O	O		
299		CH_3	O	O	O		
300		CH_3	O	O	O		
301		CH_3	O	O	NH		
302		CH_3	O	O	NH		
303		CH_3	O	O	NH		
304	C_3H_7-i	CH_3	O	O	NH		
305	C_3H_7-i	CH_3	O	O	NH		
306	C_3H_7-i	CH_3	O	O	NH		
307	C_3H_7-i	CH_3	O	O	NH		

Table 1 (continued)

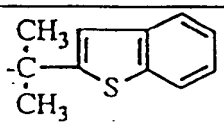
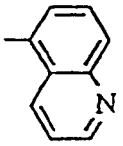
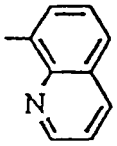
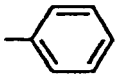
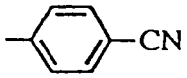
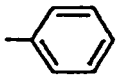
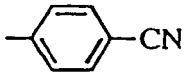
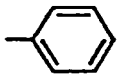
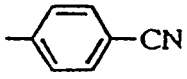
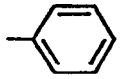
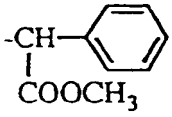
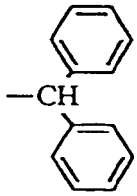
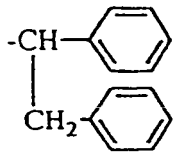
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
308	C_3H_7-i	CH_3	O	O	NH		
309	C_3H_7-i	CH_3	O	O	O		
310	C_3H_7-i	CH_3	O	O	O		
311		CH_3	O	O	NCH_2OCH_3		
312		CH_3	O	O	SO		
313		CH_3	O	O	SO_2		
314		CH_3	O	O	NH		
315	C_3H_7-i	CH_3	O	O	NH		
316	C_3H_7-i	CH_3	O	O	O		

Table 1 (continued)

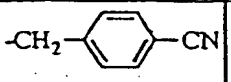
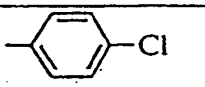
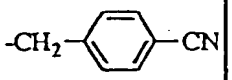
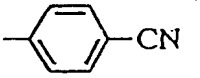
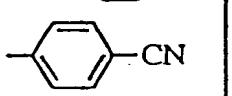
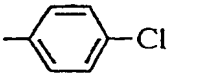
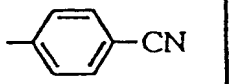
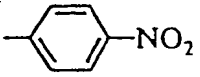
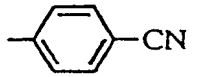
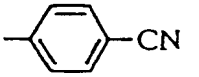
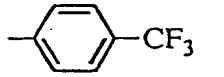
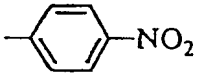
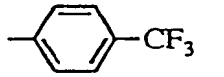
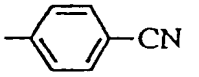
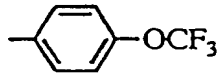
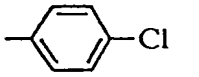
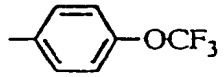
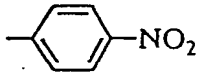
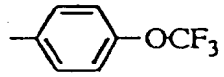
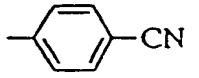
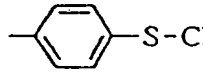
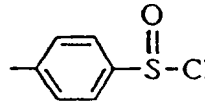
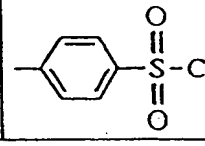
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
317		CH_3	O	O	O		
318		CH_3	O	O	O		
319		CH_3	O	O	O		
320		CH_3	O	O	O		
321		CH_3	O	O	O		
322		CH_3	O	O	O		
323		CH_3	O	O	O		115-117
324		CH_3	O	O	O		
325		CH_3	O	O	O		
326		CH_3	O	O	O		127-129
327	C_4H_9-t	CH_3	O	O	O		93-96
328	C_4H_9-t	CH_3	O	O	O		48-51
329	C_4H_9-t	CH_3	O	O	O		122-125

Table 1 (continued)

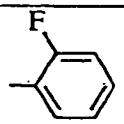
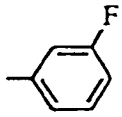
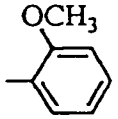
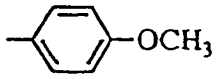
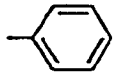
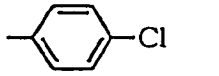
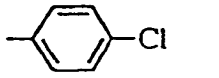
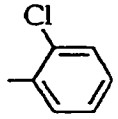
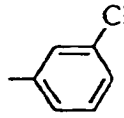
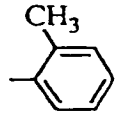
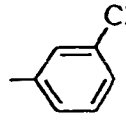
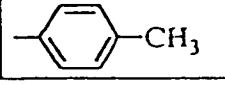
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
330	C_4H_9-t	CH_3	O	O	S		74-77
331	C_4H_9-t	CH_3	O	O	S		1.5164
332	C_4H_9-t	CH_3	O	O	S		1.5319
333	C_4H_9-t	CH_3	O	O	S		1.5361
334	C_4H_9-t	CH_3	O	O	NH		102-104
335	C_4H_9-t	CH_3	O	O	S		80-84
336	C_4H_9-t	CH_3	O	O	S		133-137
337	C_4H_9-t	CH_3	O	O	S		1.5360
338	C_4H_9-t	CH_3	O	O	S		1.5361
339	C_4H_9-t	CH_3	O	O	S		1.5274
340	C_4H_9-t	CH_3	O	O	S		1.5245
341	C_4H_9-t	CH_3	O	O	S		1.5269

Table 1 (continued)

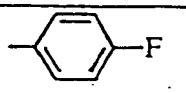
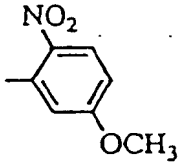
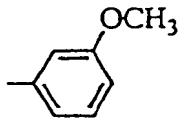
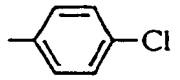
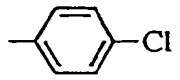
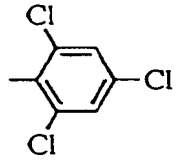
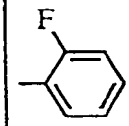
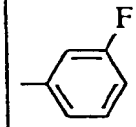
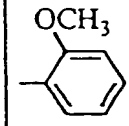
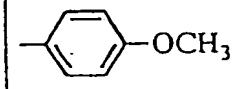
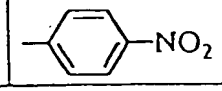
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
342	C_4H_9-t	CH_3	O	O	S		66-69
343	C_4H_9-t	CH_3	O	O	O		71-74
344	C_4H_9-t	CH_3	O	O	S		1.5312
345	C_3H_7-i	CH_3	O	O	O		161-163
346	C_3H_7-i	CH_3	O	O	O		167-171
347	C_3H_7-i	CH_3	O	O	O		166-172
348	C_3H_7-i	CH_3	O	O	S		121-123
349	C_3H_7-i	CH_3	O	O	S		125-129
350	C_3H_7-i	CH_3	O	O	S		103-106
351	C_3H_7-i	CH_3	O	O	S		122-125
352	C_3H_7-i	CH_3	O	O	S		155-158

Table 1 (continued)

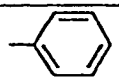
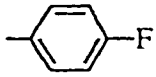
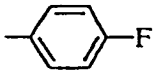
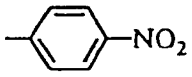
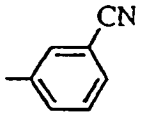
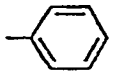
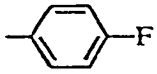
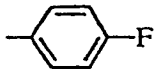
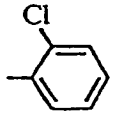
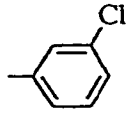
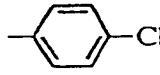
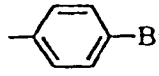
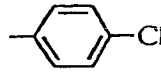
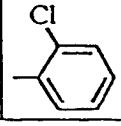
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
353	C_3H_7-i	CH_3	O	O	NH		130-134
354	C_3H_7-i	CH_3	O	O	SO		119-123
355	C_3H_7-i	CH_3	O	O	SO_2		151-153
356	C_3H_7-i	CH_3	O	O	O		177-180
357	C_3H_7-i	CH_3	O	O	S		137-140
358	C_3H_7-i	CH_3	O	O	NCH_3		145-148
359	C_3H_7-i	CH_3	O	O	NH		155-156
360	C_3H_7-i	CH_3	O	O	NCH_3		141-143
361	C_3H_7-i	CH_3	O	O	NH		85-90
362	C_3H_7-i	CH_3	O	O	NH		143-145
363	C_3H_7-i	CH_3	O	O	NCH_3		65-67
364	C_3H_7-i	CH_3	O	O	NH		146-149
365	C_3H_7-i	CH_3	O	O	S		115-118
366	C_3H_7-i	CH_3	O	O	S		124-127

Table 1 (continued)

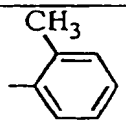
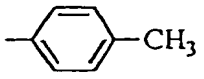
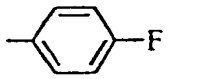
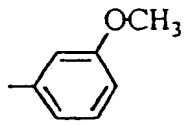
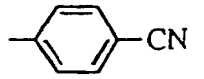
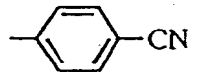
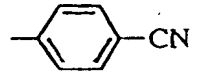
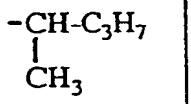
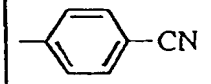
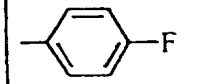
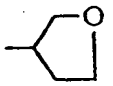
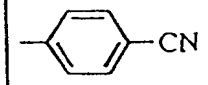
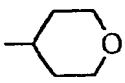
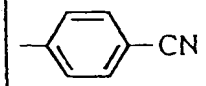
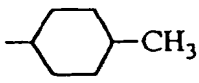
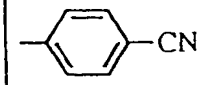
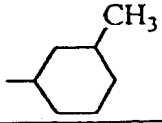
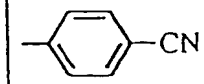
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
367	C_3H_7-i	CH_3	O	O	S		119-121
368	C_3H_7-i	CH_3	O	O	S		107-110
369	C_3H_7-i	CH_3	O	O	S		111-115
370	C_3H_7-i	CH_3	O	O	S		109-112
371	C_4H_9-i	CH_3	O	O	O		125-130
372	C_5H_{11}	CH_3	O	O	O		109-111
373	C_6H_{13}	CH_3	O	O	O		107-110
374		CH_3	O	O	O		122-125
375	C_3H_7-i	CH_3	O	O	$NCOCH_3$		56-60
376		CH_3	O	O	O		181-184
377		CH_3	O	O	O		201-204
378		CH_3	O	O	O		111-116
379		CH_3	O	O	O		141-142

Table 1 (continued)

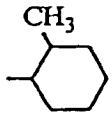
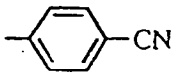
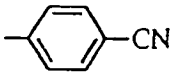
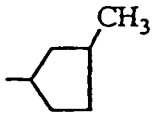
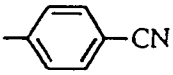
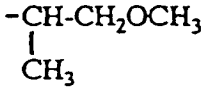
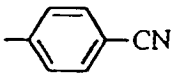
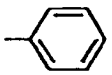
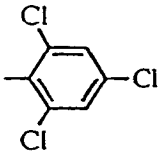
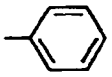
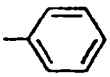
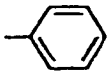
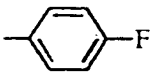
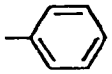
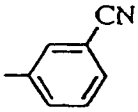
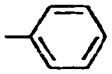
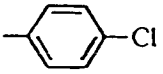
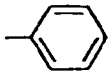
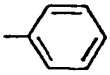
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
380		CH_3	O	O	O		133-136
381	CH_2-C≡CH	CH_3	O	O	O		148-151
382		CH_3	O	O	O		161-164
383		CH_3	O	O	O		102-107
384		CH_3	O	O	O		159-162
385		CH_3	O	O	NH		130-134
386		CH_3	O	O	S		127-130
387		CH_3	O	O	S		108-110
388		CH_3	O	O	NH		154-156
389		CH_3	O	O	NCH_3		125-130

Table 1 (continued)

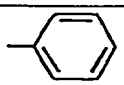
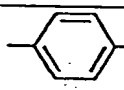
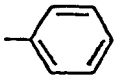
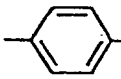
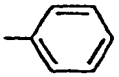
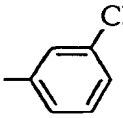
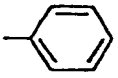
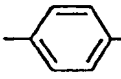
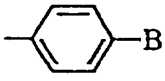
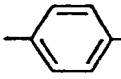
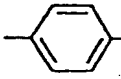
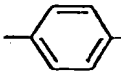
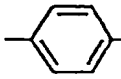
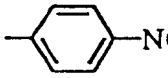
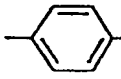
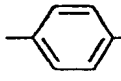
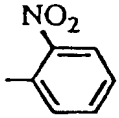
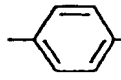
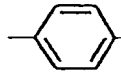
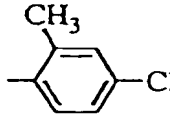
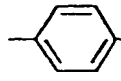
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
390		CH_3	O	O	NH		147-149
391		CH_3	O	O	NCH_3		64-70
392		CH_3	O	O	NH		117-119
393		CH_3	O	O	NH		156-160
394		CH_3	O	O	O		156-162
395		CH_3	O	O	O		137-140
396		CH_3	O	O	O		174-179
397		CH_3	O	O	O		153-156
398		CH_3	O	O	O		130-134
399		CH_3	O	O	O		156-161
400		CH_3	O	O	O		125-129
401		CH_3	O	O	O		155-158
402		CH_3	O	O	O		141-144

Table 1 (continued)

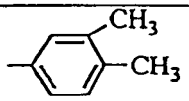
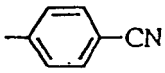
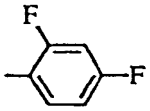
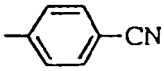
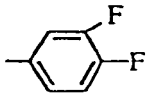
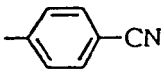
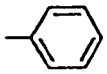
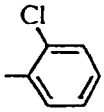
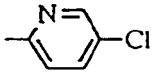
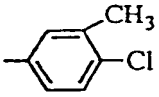
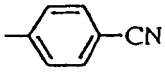
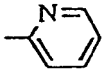
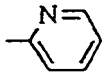
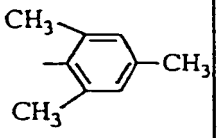
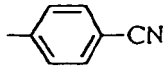
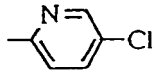
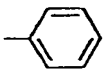
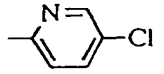
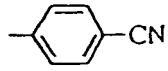
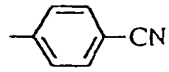
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
403		CH_3	O	O	O		153-154
404		CH_3	O	O	O		144-148
405		CH_3	O	O	O		129-133
406		CH_3	O	O	NH		60-62
407	C_4H_9-t	CH_3	O	O	O		90-93
408		CH_3	O	O	O		129-130
409	C_4H_9-t	CH_3	O	O	O		111-112
410	C_4H_9-t	CH_3	O	O	O		129-131
411		CH_3	O	O	O		163-164
412	C_3H_7-i	CH_3	O	O	O		118-120
413		CH_3	O	O	O		123-124
414	C_4H_9-t	C_2H_5	O	O	O		122-125
415	C_4H_9-t	C_2H_5	O	O	O		135-137

Table 1 (continued)

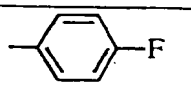
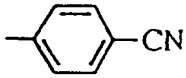
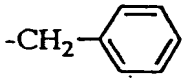
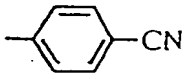
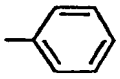
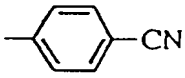
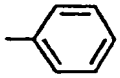
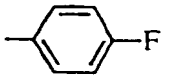
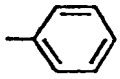
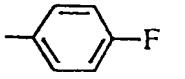
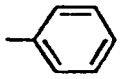
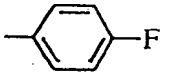
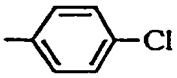
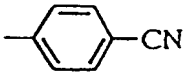
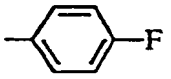
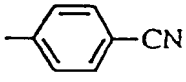
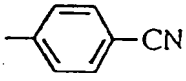
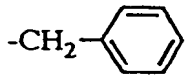
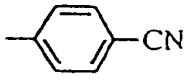
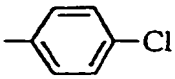
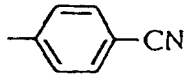
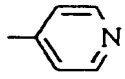
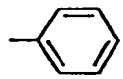
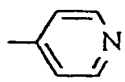
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
416	C_3H_7-i	C_2H_5	O	O	S		85-86
417	C_3H_7-i	C_2H_5	O	O	O		145-148
418		C_2H_5	O	O	O		139-141
419		C_2H_5	O	O	S		105-107
420		C_2H_5	O	O	S		130-133
421		C_2H_5	O	O	NH		137-139
422		C_2H_5	O	O	NCH_3		53-56
423		C_2H_5	O	O	O		159-163
424		C_2H_5	O	O	O		150-153
425	C_3H_7-i	H	O	O	O		118-121
426		H	O	O	O		127-132
427		H	O	O	O		141-145
428	C_3H_7-i	CH_3	O	O	O		217-220
429		CH_3	O	O	O		65-68

Table 1 (continued)

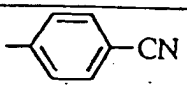
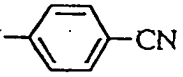
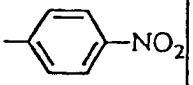
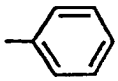
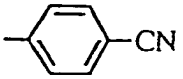
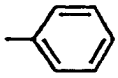
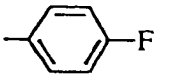
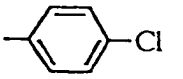
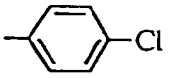
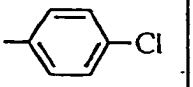
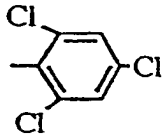
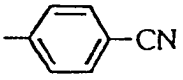
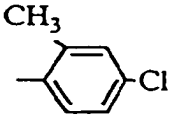
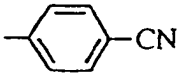
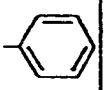
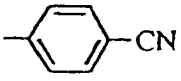
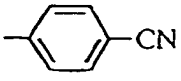
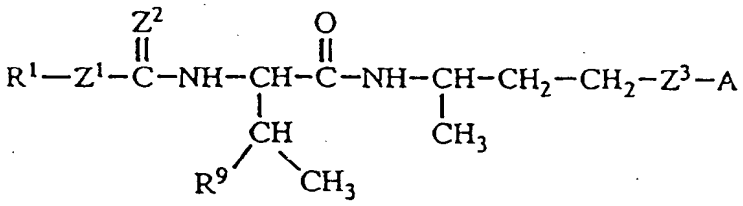
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
430	C_3H_7-i	CH_3	S	O	O		161-163
431	C_2H_5	CH_3	S	O	O		152-154
432	C_2H_5	CH_3	S	O	O		164-166
433		CH_3	O	O	O		118-120
434		CH_3	O	O	$NCOCH_3$		
435	C_3H_7-i	CH_3	O	O	NCO_2CH_3		
436	C_3H_7-i	CH_3	O	O	NCO- 		71-73
437	C_3H_7-i	CH_3	O	O	NCH_2OCH_3		
438		CH_3	O	O	O		
439		CH_3	O	O	O		135-138
440	-CH_2CH_2- 	CH_3	O	O	O		112-114
441	C_3H_7-i	CH_3	O	O	NH		

Table 2



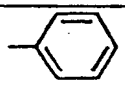
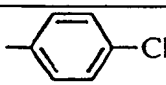
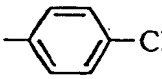
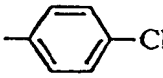
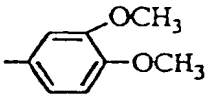
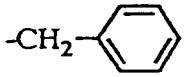
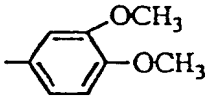
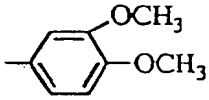
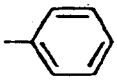
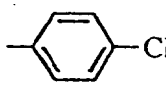
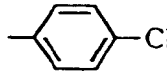
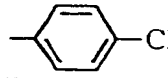
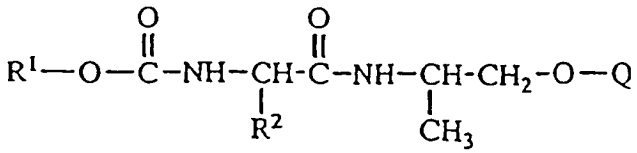
Compound No.	R^1	R^9	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
442		CH_3	O	O	O		
443	C_4H_9-t	CH_3	O	O	O		
444	C_3H_7-i	CH_3	O	O	O		
445		CH_3	O	O	O		
446		CH_3	O	O	O		
447	C_3H_7-i	CH_3	O	O	O		
448		CH_3	O	O	O		
449	C_4H_9-t	CH_3	O	O	O		
450	C_3H_7-i	CH_3	O	O	O		

Table 3



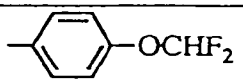
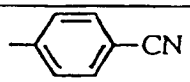
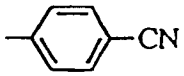
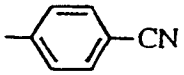
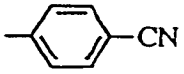
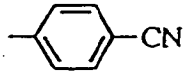
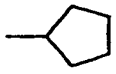
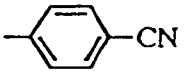
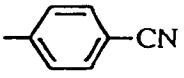
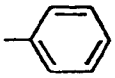
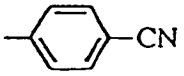
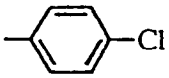
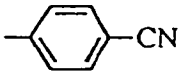
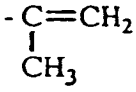
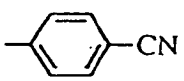
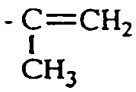
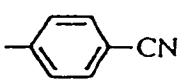
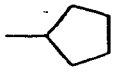
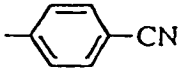
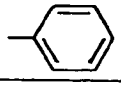
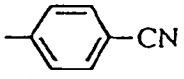
Compound No.	R^1	R^2	Q	Melting Point (°C) or Refractive Index (n_D^{20})
451		C_3H_7		117-119
452	C_4H_9-t	C_3H_7		78-80
453	C_4H_9-t	C_3H_7		105-107
454	C_4H_9-t	C_3H_7		93-95
455	C_4H_9-t	C_4H_9-i		not determined
456	C_4H_9-t			140-142
457	C_4H_9-t	C_4H_9-t		68-71
458	C_4H_9-t			61-64
459	C_4H_9-t			124-126
460	C_4H_9-t			1.5132
461	C_4H_9-t			107-109
462	C_3H_7-i			155-158
463	C_3H_7-i			149-151

Table 3 (continued)

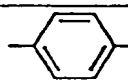
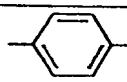
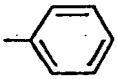
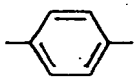
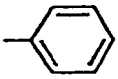
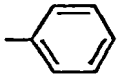
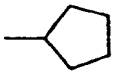
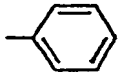
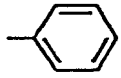
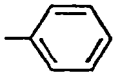
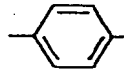
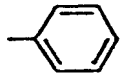
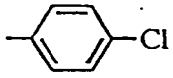
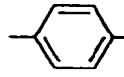
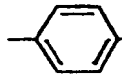
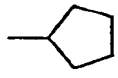
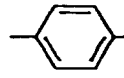
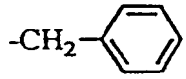
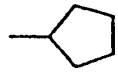
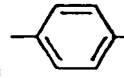
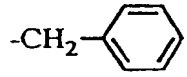
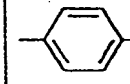
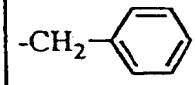
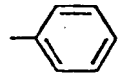
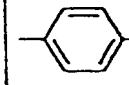
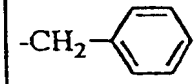
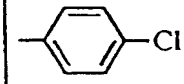
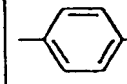
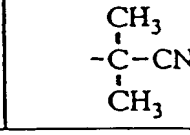
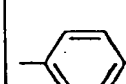
Compound No.	R^1	R^2	Q	Melting Point (°C) or Reflactive Index (n_D^{20})
464	C_3H_7-i			158-161
465		C_3H_7		88-91
466		C_4H_9-i		43-47
467				153-156
468		C_4H_9-t		75-78
469				68-71
470				152-155
471				141-145
472				170-174
473		C_4H_9-t		46-49
474				155-157
475				128-129
476		C_3H_7-i		127-129

Table 3 (continued)

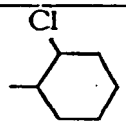
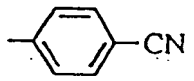
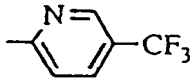
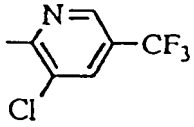
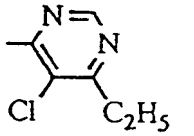
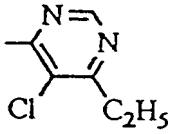
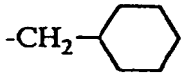
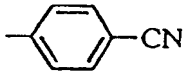
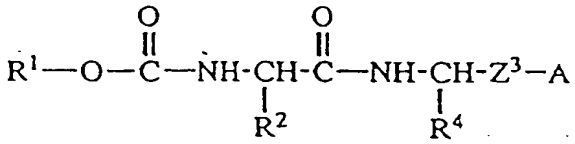
Compound No.	R^1	R^2	Q	Melting Point (°C) or Reflactive Index (n_D^{20})
477		C_3H_7-i		152-154
478	C_4H_9-t	C_3H_7-i		100-103
479	C_4H_9-t	C_3H_7-i		105-106
480	C_4H_9-t	C_3H_7-i		109-112
481	C_3H_7-i	C_3H_7-i		173-175
482		C_3H_7-i		128-129

Table 4



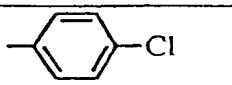
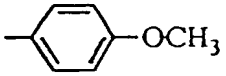
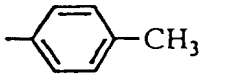
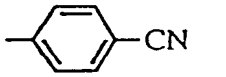
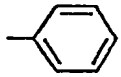
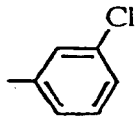
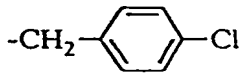
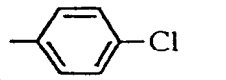
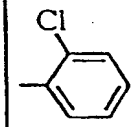
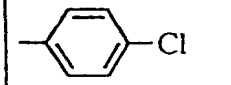
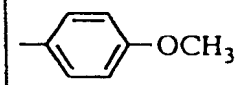
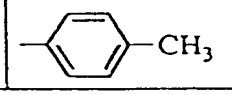
Compound No.	R^1	R^2	R^4	Z^3	A	Melting Point (°C) or Reflactive Index (np^{20})
483	C_4H_9-t	C_3H_7-i	CH_3			82-87
484	C_4H_9-t	C_3H_7-i	CH_3			156-159
485	C_4H_9-t	C_3H_7-i	CH_3			145-149
486	C_4H_9-t	C_3H_7-i	CH_3			96-100
487	C_4H_9-t	C_3H_7-i	CH_3			157-158
488	C_4H_9-t	C_3H_7-i	CH_3			83-86
489	C_4H_9-t	C_3H_7-i	CH_3			144-146
490	C_4H_9-t	C_3H_7-i	CH_3			70-73
491	C_4H_9-t	C_3H_7-i	CH_3			140-143
492	C_3H_7-i	C_3H_7-i	CH_3			179-182
493	C_3H_7-i	C_3H_7-i	CH_3			251-255
494	C_3H_7-i	C_3H_7-i	CH_3			219-222

Table 4 (continued)

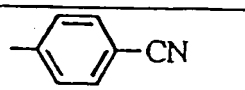
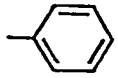
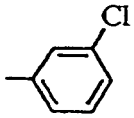
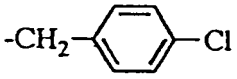
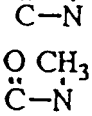
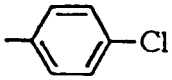
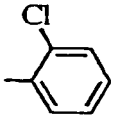
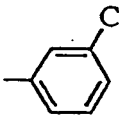
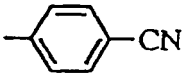
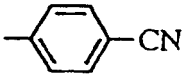
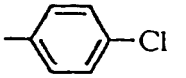
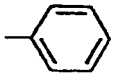
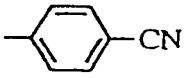
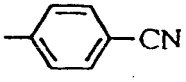
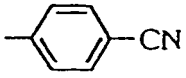
Compound No.	R^1	R^2	R^4	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
495	C_3H_7-i	C_3H_7-i	CH_3			88-92
496	C_3H_7-i	C_3H_7-i	CH_3			211-212
497	C_3H_7-i	C_3H_7-i	CH_3			210-213
498	C_3H_7-i	C_3H_7-i	CH_3			200-203
499	C_3H_7-i	C_3H_7-i	CH_3			68-72
500	C_3H_7-i	C_3H_7-i	CH_3			205-210
501	C_3H_7-i	C_3H_7-i	CH_3			113-115
502	C_3H_7-i	C_3H_7-i	CH_3			184-186
503	C_3H_7-i	C_3H_7-i	CH_3			73-75
504	C_3H_7-i	C_3H_7-i	CH_3	COO		184-185
505	C_3H_7-i	C_3H_7-i	CH_3	COO		151-153
506	C_3H_7-i	C_4H_9-s	CH_3			197-198
507	C_3H_7-i	C_2H_5	CH_3			84-87
508	C_4H_9-s	C_3H_7-i	CH_3			165-167

Table 4 (continued)

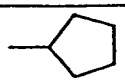
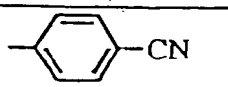
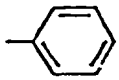
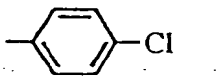
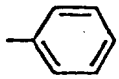
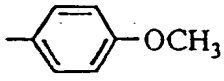
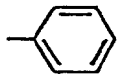
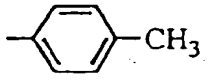
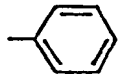
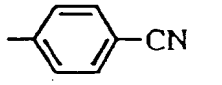
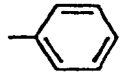
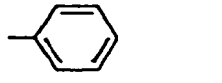
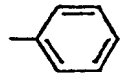
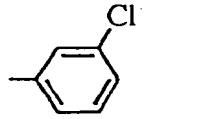
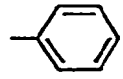
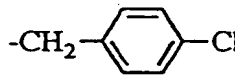
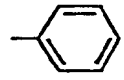
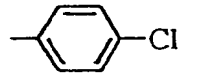
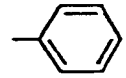
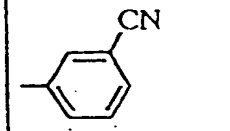
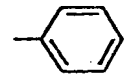
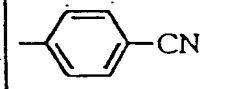
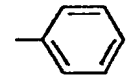
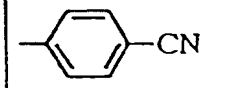
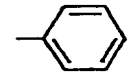
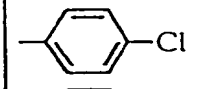
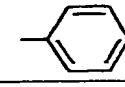
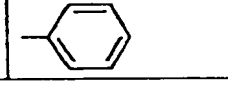
Compound No.	R^1	R^2	R^4	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
509		C_3H_7-i	CH_3			197-199
510		C_3H_7-i	CH_3			201-204
511		C_3H_7-i	CH_3			219-221
512		C_3H_7-i	CH_3			245-250
513		C_3H_7-i	CH_3			225-230
514		C_3H_7-i	CH_3			199-202
515		C_3H_7-i	CH_3			194-197
516		C_3H_7-i	CH_3			173-175
517		C_3H_7-i	CH_3			69-71
518		C_3H_7-i	CH_3			149-153
519		C_3H_7-i	CH_3			158-161
520		C_3H_7-i	CH_3			202-203
521		C_3H_7-i	CH_3	COO		168-170
522		C_3H_7-i	CH_3	COO		175-178

Table 4 (continued)

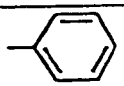
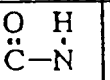
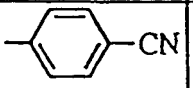
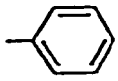
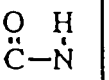
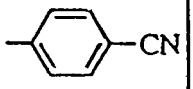
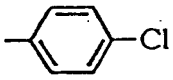
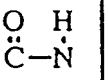
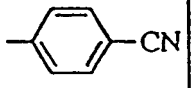
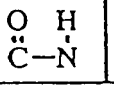
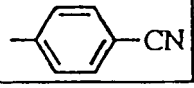
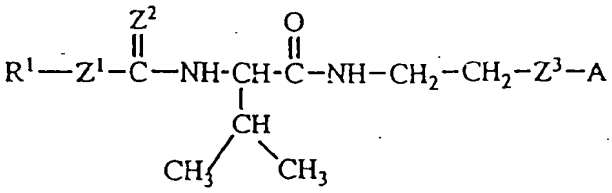
Compound No.	R^1	R^2	R^4	Z^3	A	Melting Point (°C) or Refractive Index (nD^{20})
523		C_4H_9-s	CH_3			157-159
524		C_2H_5	CH_3			156-158
525		C_3H_7-i	CH_3			182-184
526	C_3H_7-i	C_3H_7-i	H			181-185

Table 5



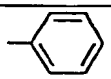
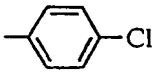
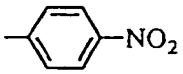
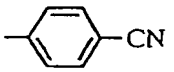
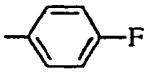
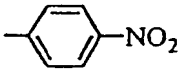
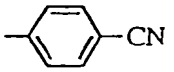
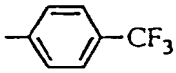
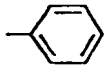
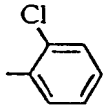
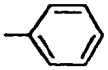
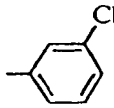
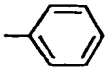
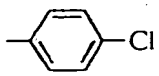
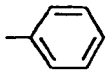
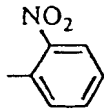
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
527	C_4H_9-t	O	O	O		158-160
528	C_4H_9-t	O	O	O		
529	C_4H_9-t	O	O	O		
530	C_4H_9-t	O	O	O		
531	C_4H_9-t	O	O	O		
532	C_3H_7-i	O	O	O		
533	C_3H_7-i	O	O	O		
534	C_3H_7-i	O	O	O		
535		O	O	O		
536		O	O	O		
537		O	O	O		
538		O	O	O		

Table 5 (continued)

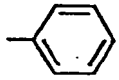
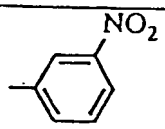
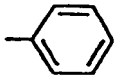
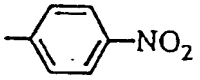
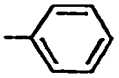
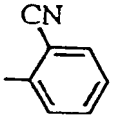
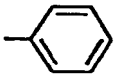
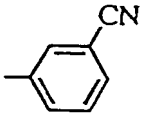
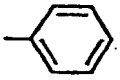
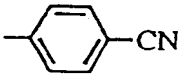
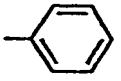
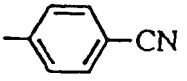
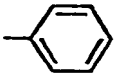
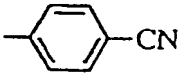
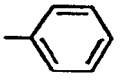
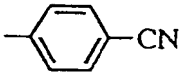
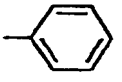
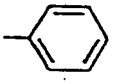
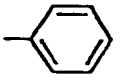
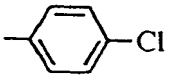
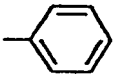
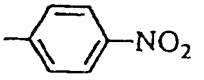
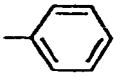
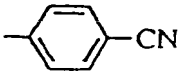
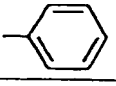
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
539		O	O	O		
540		O	O	O		
541		O	O	O		
542		O	O	O		
543		O	O	O		
544		S	O	O		
545		O	S	O		
546		S	S	O		
547		O	O	S		
548		O	O	S		
549		O	O	S		
550		O	O	S		
551	C_4H_9-t	O	O	S		75-77

Table 5 (continued)

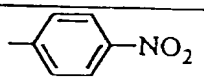
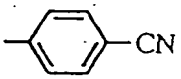
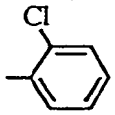
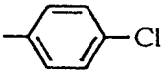
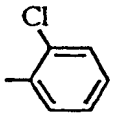
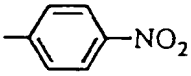
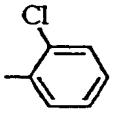
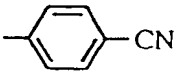
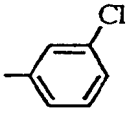
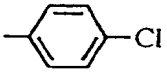
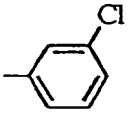
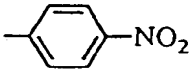
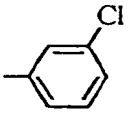
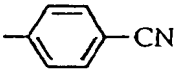
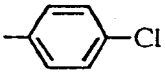
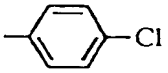
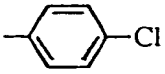
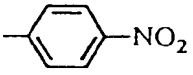
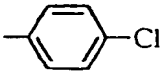
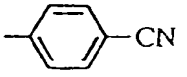
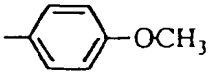
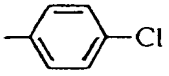
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
552	C_4H_9-t	O	O	S		
553	C_4H_9-t	O	O	S		
554		O	O	O		
555		O	O	O		
556		O	O	O		
557		O	O	O		
558		O	O	O		
559		O	O	O		
560		O	O	O		
561		O	O	O		
562		O	O	O		
563		O	O	O		

Table 5 (continued)

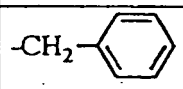
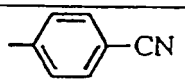
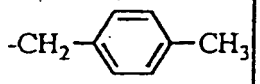
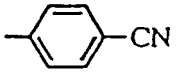
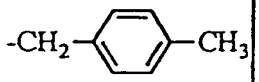
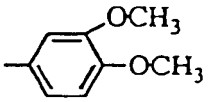
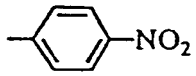
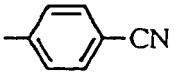
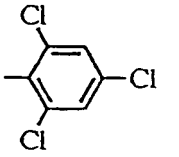
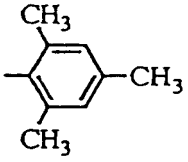
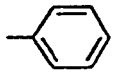
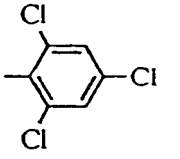
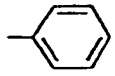
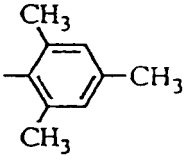
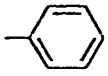
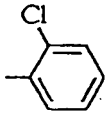
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
564		O	O	O		
565		O	O	O		
566		O	O	O		
567		O	O	O		
568	C_3H_7-i	O	O	O		
569	C_3H_7-i	O	O	O		
570		O	O	O		
571		O	O	O		
572	C_3H_7-i	O	O	NH		
573	C_3H_7-i	O	O	NH		

Table 5 (continued)

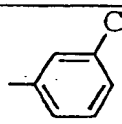
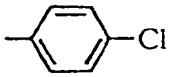
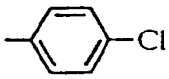
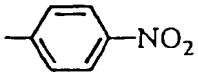
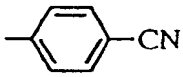
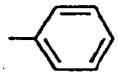
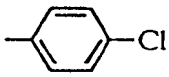
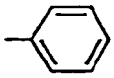
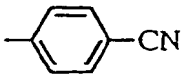
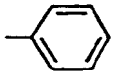
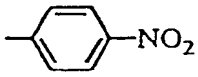
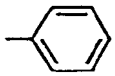
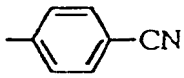
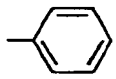
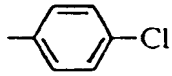
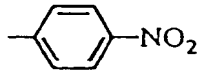
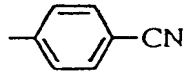
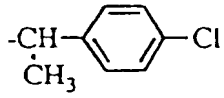
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
574	C_3H_7-i	O	O	NH		
575	C_3H_7-i	O	O	NH		
576	C_3H_7-i	O	O	NCH_3		
577	C_3H_7-i	O	O	NCH_3		
578	C_3H_7-i	O	O	NCH_3		
579		O	O	NH		
580		O	O	NH		
581		O	O	NCH_3		
582		O	O	NCH_3		
583	C_3H_7-i	O	O	S		
584	C_3H_7-i	O	O	S		
585	C_3H_7-i	O	O	S		
586	C_3H_7-i	O	O	S		
587	C_3H_7-i	O	O	O		

Table 5 (continued)

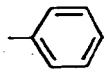
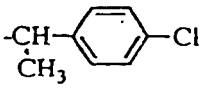
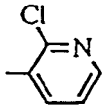
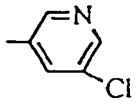
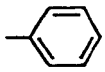
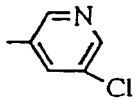
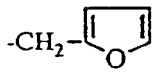
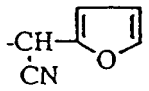
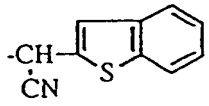
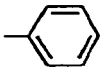
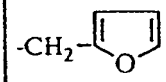
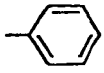
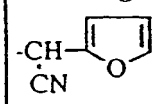
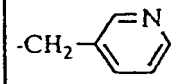
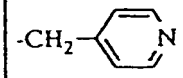
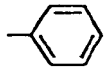
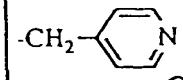
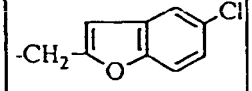
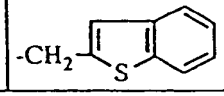
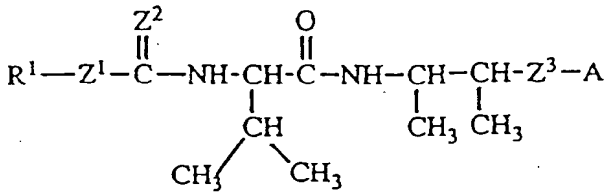
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Refractive Index (n_D^{20})
588		O	O	O		
589	C_3H_7-i	O	O	O		
590	C_3H_7-i	O	O	O		
591		O	O	O		
592	C_3H_7-i	O	O	O		
593	C_3H_7-i	O	O	O		
594	C_3H_7-i	O	O	O		
595		O	O	O		
596		O	O	O		
597	C_3H_7-i	O	O	O		
598	C_3H_7-i	O	O	O		
599		O	O	O		
600	C_3H_7-i	O	O	O		
601	C_3H_7-i	O	O	O		

Table 6



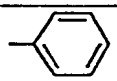
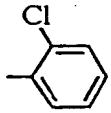
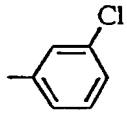
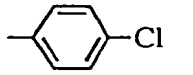
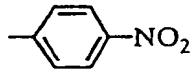
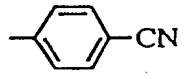
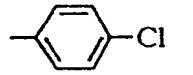
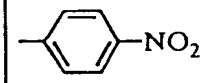
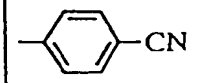
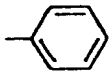
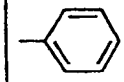
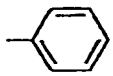
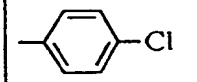
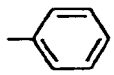
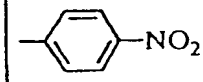
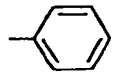
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
602	C_4H_9-t	O	O	O		not determined
603	C_4H_9-t	O	O	O		
604	C_4H_9-t	O	O	O		
605	C_4H_9-t	O	O	O		1.4784
606	C_4H_9-t	O	O	O		1.5109
607	C_4H_9-t	O	O	O		not determined
608	C_3H_7-i	O	O	O		
609	C_3H_7-i	O	O	O		
610	C_3H_7-i	O	O	O		
611		O	O	O		
612		O	O	O		
613		O	O	O		
614		O	O	O		

Table 6 (continued)

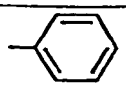
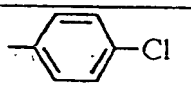
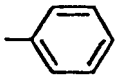
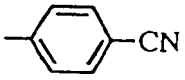
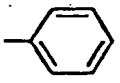
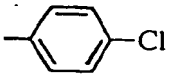
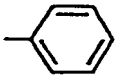
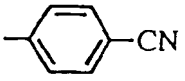
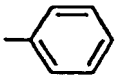
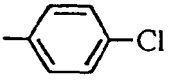
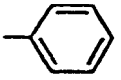
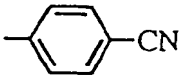
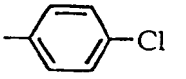
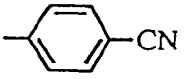
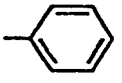
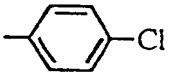
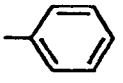
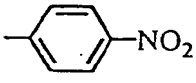
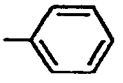
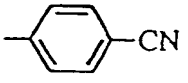
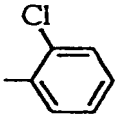
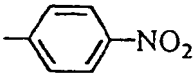
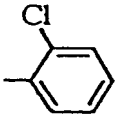
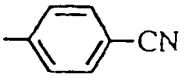
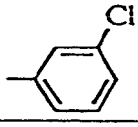
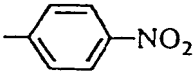
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
615		S	O	O		
616		S	O	O		
617		O	S	O		
618		O	S	O		
619		S	S	O		
620		S	S	O		
621	C_3H_7-i	O	O	S		
622	C_3H_7-i	O	O	S		
623		O	O	S		
624		O	O	S		
625		O	O	S		
626		O	O	O		
627		O	O	O		
628		O	O	O		

Table 6 (continued)

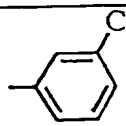
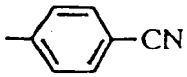
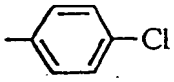
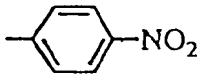
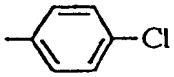
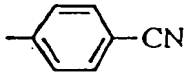
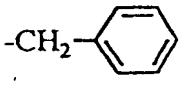
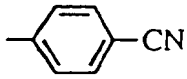
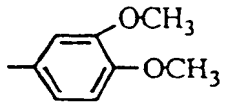
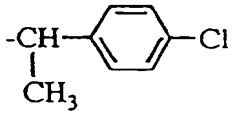
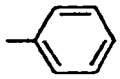
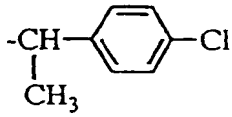
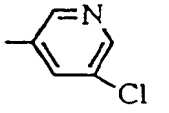
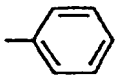
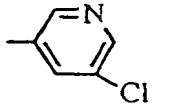
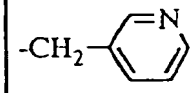
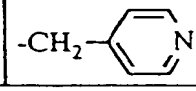
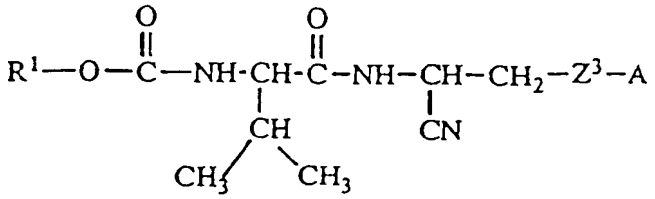
Compound No.	R^1	Z^1	Z^2	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
629		O	O	O		
630		O	O	O		
631		O	O	O		
632		O	O	O		
633		O	O	O		
634	C_3H_7-i	O	O	O		
635		O	O	O		
636	C_3H_7-i	O	O	O		
637		O	O	O		
638	C_3H_7-i	O	O	O		
639	C_3H_7-i	O	O	O		

Table 7



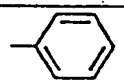
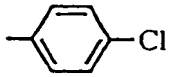
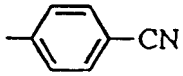
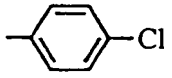
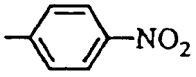
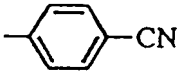
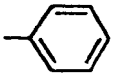
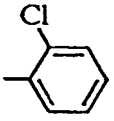
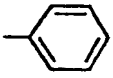
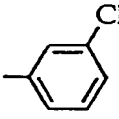
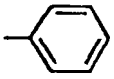
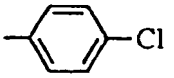
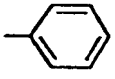
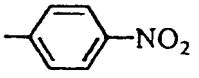
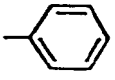
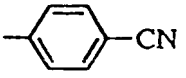
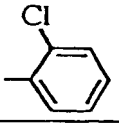
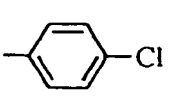
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
640	C_4H_9-t	O		
641	C_4H_9-t	O		
642	C_4H_9-t	O		
643	C_3H_7-i	O		153-155
644	C_3H_7-i	O		
645	C_3H_7-i	O		
646		O		
647		O		
648		O		157-160
649		O		
650		O		
651		O		

Table 7 (continued)

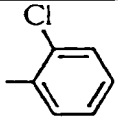
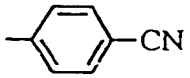
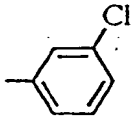
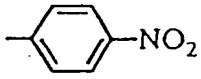
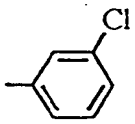
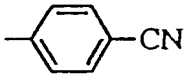
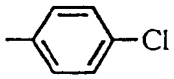
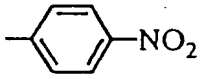
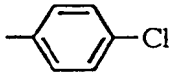
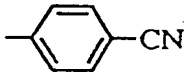
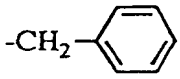
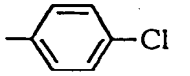
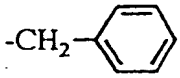
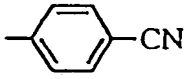
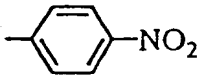
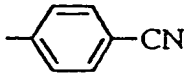
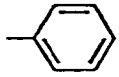
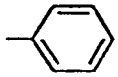
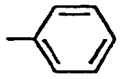
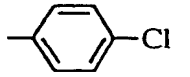
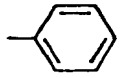
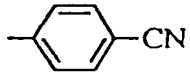
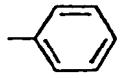
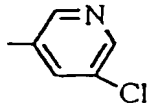
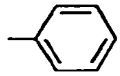
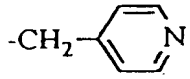
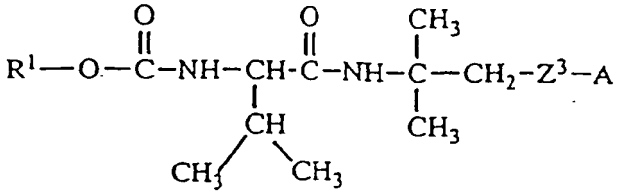
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
652		O		
653		O		
654		O		
655		O		
656		O		
657		O		
658		O		
659		O		
660		O		
661		S		
662		S		
663		S		
664		O		
665		O		

Table 8



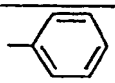
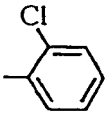
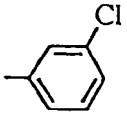
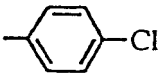
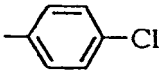
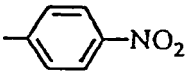
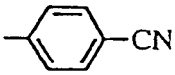
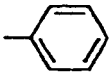
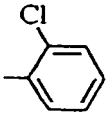
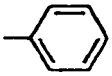
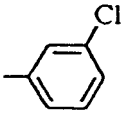
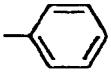
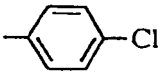
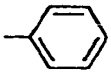
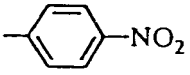
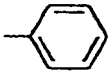
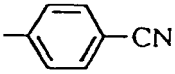
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (nD^{20})
666	C_4H_9-t	O		
667	C_4H_9-t	O		
668	C_4H_9-t	O		
669	C_4H_9-t	O		
670	C_3H_7-i	O		
671	C_3H_7-i	O		
672	C_3H_7-i	O		1.5111
673		O		
674		O		
675		O		
676		O		
677		O		

Table 8 (continued)

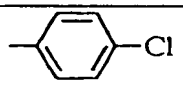
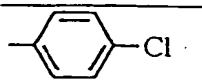
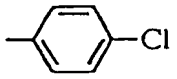
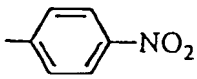
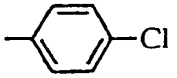
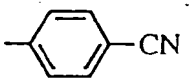
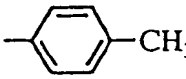
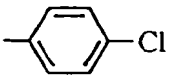
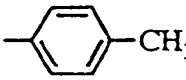
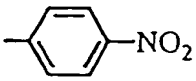
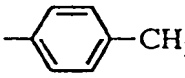
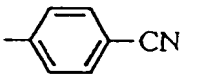
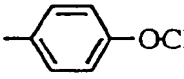
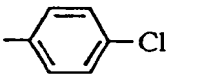
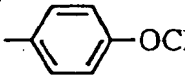
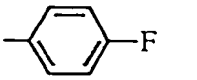
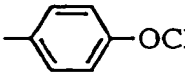
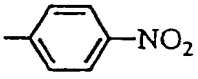
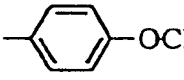
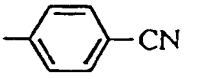
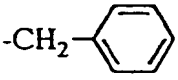
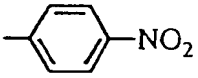
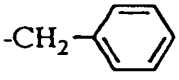
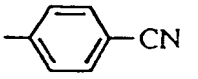
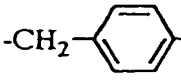
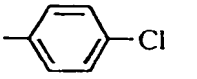
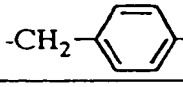
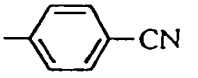
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
678		O		
679		O		
680		O		
681		O		
682		O		
683		O		
684		O		
685		O		
686		O		
687		O		
688		O		
689		O		
690		O		
691		O		

Table 8 (continued)

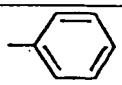
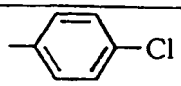
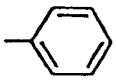
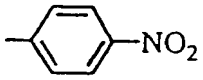
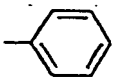
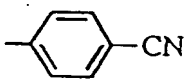
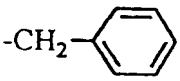
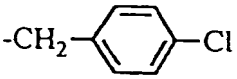
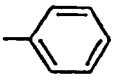
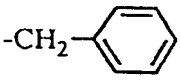
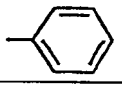
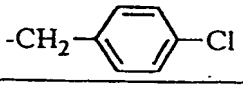
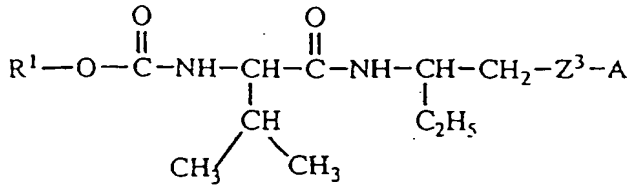
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
692		S		
693		S		
694		S		
695	C_3H_7-i	O		
696	C_3H_7-i	O		
697		O		
698		O		

Table 9



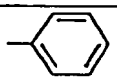
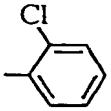
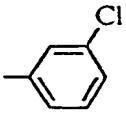
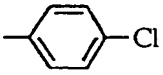
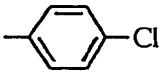
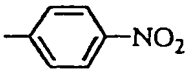
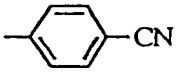
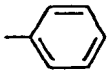
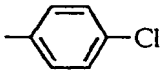
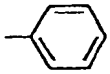
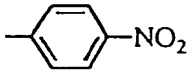
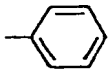
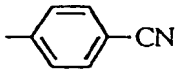
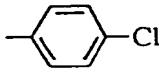
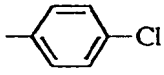
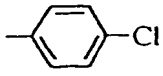
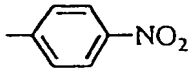
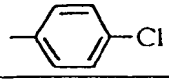
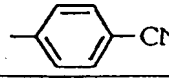
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
699	C_4H_9-t	O		
700	C_4H_9-t	O		
701	C_4H_9-t	O		
702	C_4H_9-t	O		
703	C_3H_7-i	O		
704	C_3H_7-i	O		
705	C_3H_7-i	O		
706		O		
707		O		
708		O		128-130
709		O		
710		O		
711		O		

Table 9 (continued)

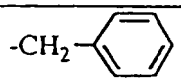
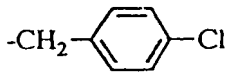
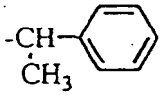
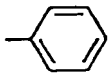
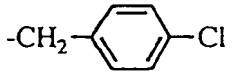
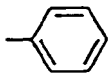
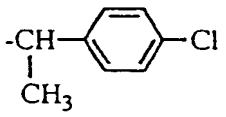
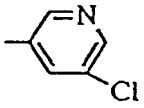
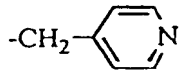
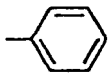
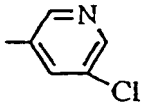
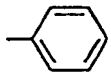
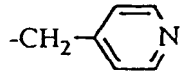
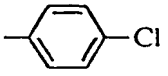
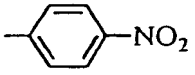
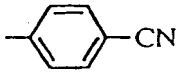
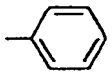
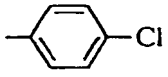
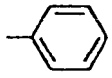
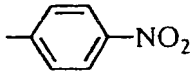
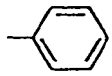
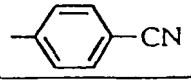
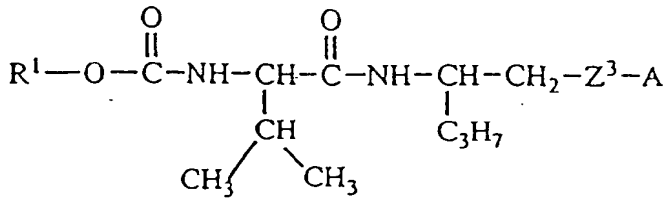
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
712	C_3H_7-i	O		
713	C_3H_7-i	O		
714	C_3H_7-i	O		
715		O		
716		O		
717	C_3H_7-i	O		
718	C_3H_7-i	O		
719		O		
720		O		
721	C_3H_7-i	S		
722	C_3H_7-i	S		
723	C_3H_7-i	S		
724		S		
725		S		
726		S		

Table 10



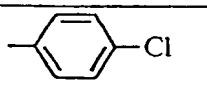
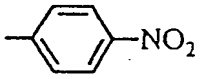
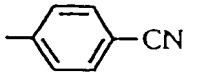
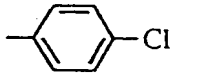
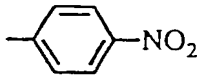
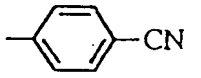
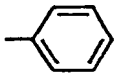
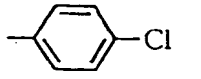
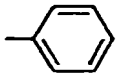
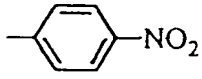
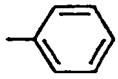
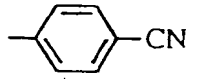
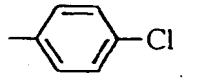
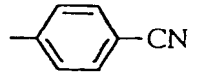
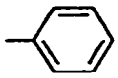
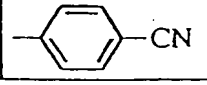
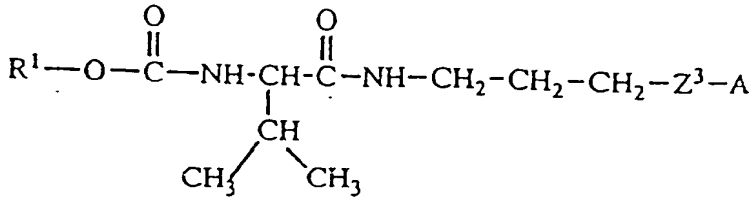
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
727	C_4H_9-t	O		
728	C_4H_9-t	O		
729	C_4H_9-t	O		
730	C_3H_7-i	O		
731	C_3H_7-i	O		
732	C_3H_7-i	O		
733		O		
734		O		
735		O		
736	C_3H_7-i	S		
737	C_3H_7-i	S		
738		S		

Table 11



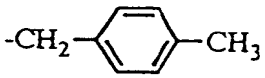
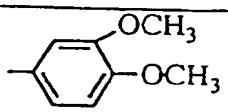
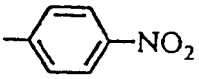
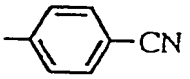
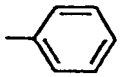
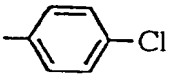
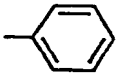
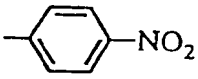
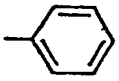
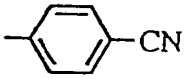
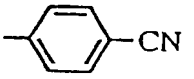
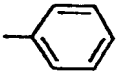
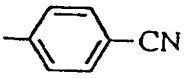
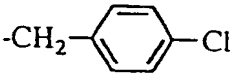
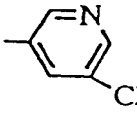
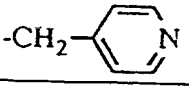
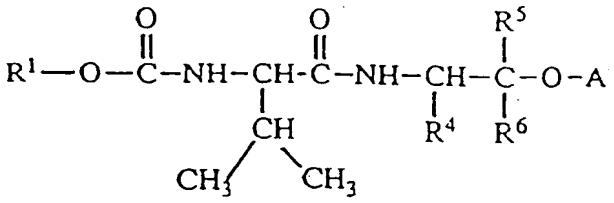
Compound No.	R^1	Z^3	A	Melting Point (°C) or Reflactive Index (n_D^{20})
739		O		
740	C_3H_7-i	O		
741	C_3H_7-i	O		150-151
742		O		
743		O		
744		O		
745	C_3H_7-i	S		
746		S		
747	C_3H_7-i	O		
748	C_3H_7-i	O		
749	C_3H_7-i	O		

Table 12



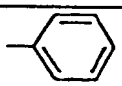
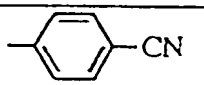
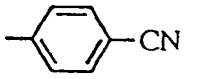
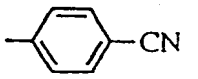
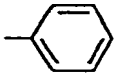
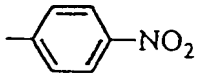
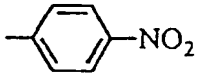
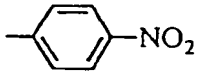
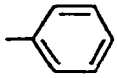
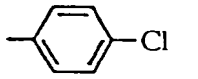
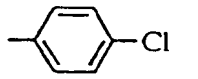
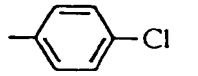
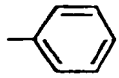
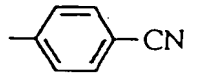
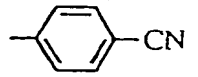
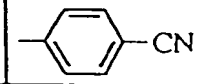
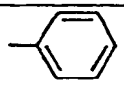
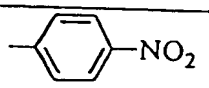
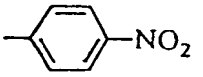
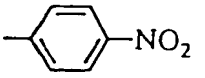
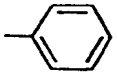
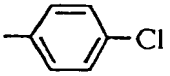
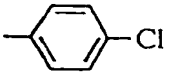
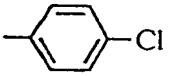
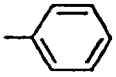
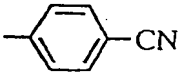
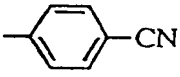
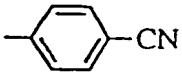
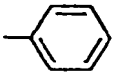
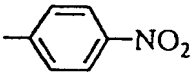
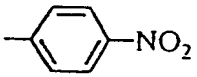
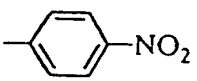
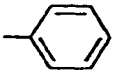
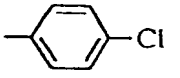
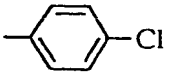
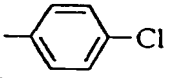
Compound No.	R^1	R^4	R^5	R^6	A	Melting Point (°C) or Reflactive Index (n_D^{20})
750		H	H	CH_3		not determined
751	C_4H_9-t	H	H	CH_3		46-50
752	C_3H_7-i	H	H	CH_3		
753		H	H	CH_3		
754	C_4H_9-t	H	H	CH_3		
755	C_3H_7-i	H	H	CH_3		
756		H	H	CH_3		
757	C_4H_9-t	H	H	CH_3		
758	C_3H_7-i	H	H	CH_3		
759		H	CH_3	CH_3		
760	C_4H_9-t	H	CH_3	CH_3		
761	C_3H_7-i	H	CH_3	CH_3		

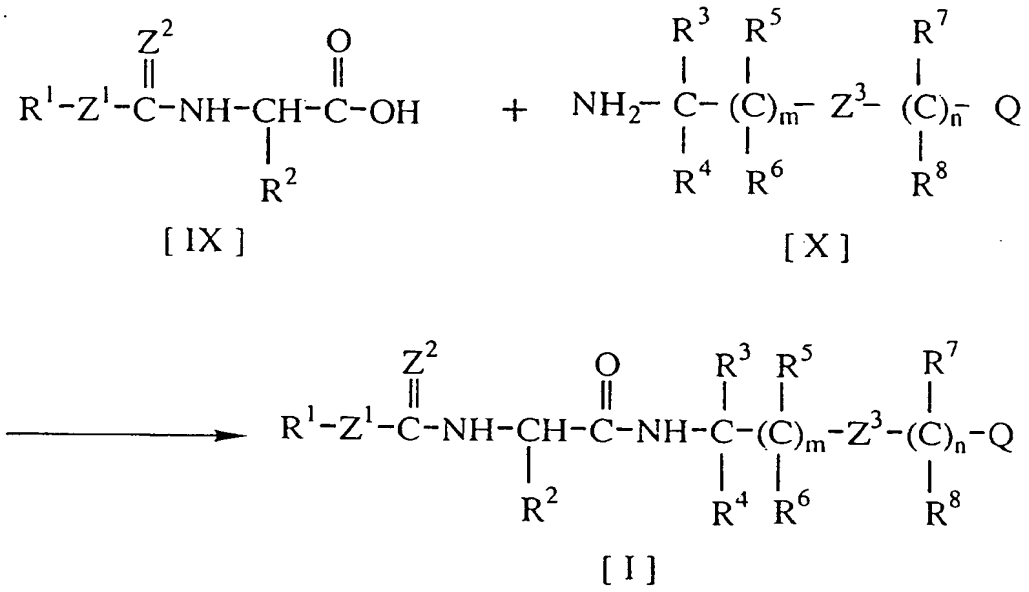
Table 12 (continued)

Compound No.	R^1	R^4	R^5	R^6	A	Melting Point (°C) or Reflactive Index (n_D^{20})
762		H	CH_3	CH_3		
763	C_4H_9-t	H	CH_3	CH_3		
764	C_3H_7-i	H	CH_3	CH_3		
765		H	CH_3	CH_3		
766	C_4H_9-t	H	CH_3	CH_3		
767	C_3H_7-i	H	CH_3	CH_3		
768		C_3H_7-i	H	H		150-152
769	C_4H_9-t	C_3H_7-i	H	H		
770	C_3H_7-i	C_3H_7-i	H	H		154-157
771		C_3H_7-i	H	H		
772	C_4H_9-t	C_3H_7-i	H	H		
773	C_3H_7-i	C_3H_7-i	H	H		
774		C_3H_7-i	H	H		
775	C_4H_9-t	C_3H_7-i	H	H		
776	C_3H_7-i	C_3H_7-i	H	H		

The compounds represented by Formula [I] according to the present invention can be prepared, for example, in the following manner.

Preparation Process A

(Reaction Scheme 1)



[IX] [X] [I]

wherein R^1, R^2, R^3, R^4, R^5, R^6, R^7, R^8, Z^1, Z^2, Z^3, Q, m, and n have the same meanings as defined above.

The compounds represented by Formula [I] according to the present invention can be prepared by reacting an amino acid derivative represented by Formula [IX] or the amino acid derivative wherein the carboxyl group is activated, with an amine represented by Formula [X] in the presence of a base and/or a catalyst, if necessary.

In the present reaction, as the amino acid derivative represented by Formula [IX] with an activated carboxyl group, there can be mentioned, for example, an acid halide such as an acid chloride, an acid anhydride derived from the two molecules of the amino acid derivatives represented by Formula [IX], a mixed acid anhydride derived from the amino acid derivative represented by Formula [IX] and other acid or an O-alkyl carbonic acid, and an activated ester such as p-nitrophenyl ester, 2-tetrahydropyranyl ester, and 2-pyridyl ester and the like. These amino acid derivatives with activated carboxyl groups can be synthesized according to conventional methods [for example, see *Methoden der Organischen Chemie*, Vol. 15, No. 2, from page 2; *Georg Thieme Verlag Stuttgart*: 1974; *Chemische Berichte*, Vol. 38, page 605 (1905); *Journal of the American Chemical Society*, Vol. 74; page 676 (1952); and *Journal of the American Chemical Society*, Vol. 86, page 1839 (1964)].

In addition, it is also possible to perform the present reaction using a condensing agent such as N, N'-dicyclohexylcarbodiimide, carbonyldiimidazole, 2-chloro-1,3-dimethylimidazolium chloride, or the like.

The present reaction can be performed in a conventional solvent: this solvent can be any solvent that does not hinder the reaction, for example, hydrocarbons such as pentane, hexane, heptane, cyclohexane, petroleum ether, ligroin, benzene, toluene, xylene and the like, halogenated hydrocarbons such as dichloromethane, dichloroethane, chloroform, carbon tetrachloride, chlorobenzene, dichlorobenzene and the like, ethers such as diethyl ether, diisopropyl ether, ethylene glycol dimethyl ether, tetrahydrofuran, dioxane and the like, ketones such as acetone, methyl ethyl ketone, methyl isopropyl ketone, methyl isobutyl ketone and the like, esters such as methyl acetate, ethyl acetate and the like, nitriles such as acetonitrile, propionitrile, benzonitrile and the like, aprotic polar solvents such as dimethylsulfoxide, dimethylformamide, sulfolane and the like, and mixed solvents combining solvents selected from the aforementioned.

The base can be any type of base generally used in this type of reaction. For example, there can be mentioned hydroxides of alkaline metals such as sodium hydroxide, potassium hydroxide and the like, hydroxides of alkaline earth metals such as calcium hydroxide and the like, carbonates of alkaline metals

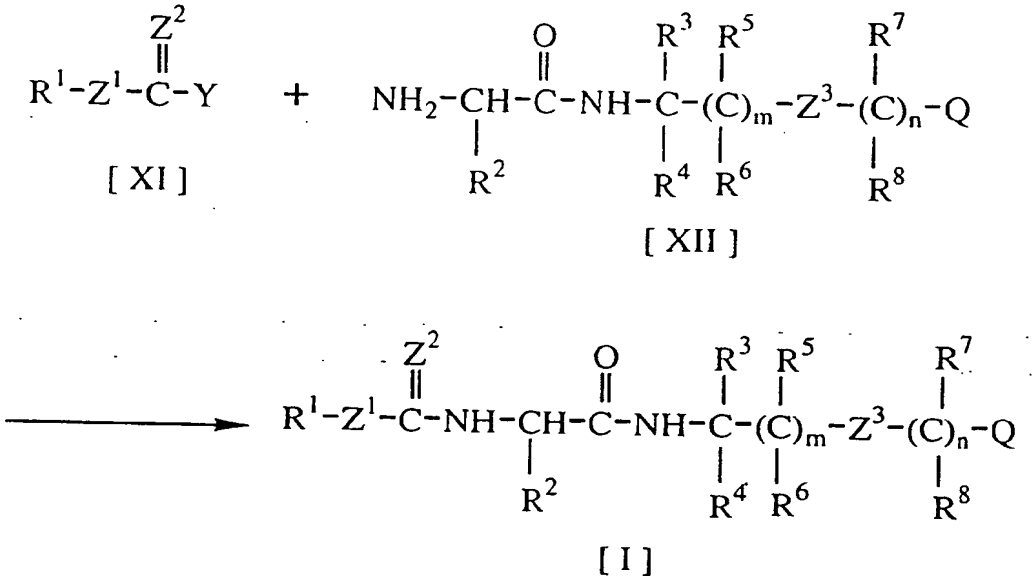
such as sodium carbonate, potassium carbonate and the like, bicarbonates of alkaline metals such as sodium bicarbonate, potassium bicarbonate and the like, organic bases such as triethylamine, trimethylamine, dimethylaniline, pyridine, N-methylmorpholine, N-methylpiperidine, 1,5-diazabicyclo [4.3.0] non-5-ene (DBN), 1,8-diazabicyclo[5.4.0] undec-7-ene (DBU), and the like, and preferably tertiary amines such as triethylamine, pyridine, N-methylpiperidine and the like.

As the catalyst, there can be mentioned 4-dimethylaminopyridine, 1-hydroxybenzotriazole, dimethylformamide and the like. The present reaction is carried out at a temperature of -75 °C to 100 °C, preferably -60 °C to 40 °C. The reaction time is preferably 1 to 20 hours.

Furthermore, compounds represented by formula [IX] as the starting material can generally be synthesized by conventional methods [for example, see *Methoden der Organischen Chemie*, Vol. 15, No. 2, from page 2; *Georg Thieme Verlag Stuttgart*: 1974; *Chemistry of the Amino Acids*, vol. 2, page 891; *John Wiley & Sons, N.Y.* (1964); and *Journal of the American Chemical Society*, Vol. 79, page 4686 (1957)]. Various manufacturing methods for compounds [X] can also be considered such as those methods stated in Japanese patent application First Publication No. Sho 63-146876, *Tetrahedron Letters*, page 21, 1973, and Japanese Patent Application, First Publication, No. Hei 5-271206).

Preparation Process B

(Reaction Scheme 2)



[XI] + [XII] → [I]

wherein R^1, R^2, R^3, R^4, R^5, R^6, R^7, R^8, Z^1, Z^2, Z^3, Q, m, and n have the same meanings as defined above, and Y represents a halogen atom, a 4,6-dimenylpyrimidinylthio group, an $R^1OC(O)O$- group, or an -ON = C(CN)Ph group (in which Ph represents a phenyl group).

Compounds of the present invention represented by Formula [I] can be manufactured by means of reacting the compound represented by Formula [XI] with an amine represented by Formula [XII] or the salt of the amine derivative with an inorganic acid such as hydrochloride and the like, or a salt of the amine derivative with an organic acid such as tosylate and the like, in the presence of a base when required.

The present reaction can be performed in a conventional solvent: this solvent can be any solvent that does not hinder the reaction, for example, hydrocarbons such as pentane, hexane, heptane, cyclohexane, petroleum ether, ligroin, benzene, toluene, xylene and the like, halogenated hydrocarbons such as dichloromethane, dichloroethane, chloroform, carbon tetrachloride, chlorobenzene, dichlorobenzene and the like, ethers such as diethyl ether, diisopropyl ether, ethylene glycol dimethyl ether, tetrahydrofuran, dioxane and the like, ketones such as acetone, methyl ethyl ketone, methyl isopropyl ketone, methyl isobutyl ketone and the like, esters such as methyl acetate, ethyl acetate and the like, nitriles such as acetonitrile, propionitrile, benzonitrile and the like, aprotic polar solvents such as dimethylsulfoxide, dimethylformamide,

sulfolane and the like, water, and mixed solvents combining solvents selected from the aforementioned.

The base can be any type of base generally used in this type of reaction. For example, there can be mentioned hydroxides of alkaline metals such as sodium hydroxide, potassium hydroxide and the like, hydroxides of alkaline earth metals such as calcium hydroxide and the like, carbonates of alkaline metals such as sodium carbonate, potassium carbonate and the like, bicarbonates of alkaline metals such as sodium bicarbonate, potassium bicarbonate and the like, organic bases such as triethylamine, trimethylamine, dimethylaniline, N-methylmorpholine, pyridine, N-methylpiperidine, 1,5-diazabicyclo [4.3.0] non-5-ene (DBN), 1,8-diazabicyclo [5.4.0] undec-7-ene (DBU), and the like, and preferably tertiary amines such as triethylamine, pyridine, N-methylpiperidine and the like. The present reaction is carried out at a temperature of -20 °C to 100 °C, preferably 0 °C to 40 °C. The reaction time is preferably 30 minutes to 20 hours.

Compounds represented by Formula [XII] as the starting material represent novel compounds, and can be manufactured, for example, by means of treating carbamates of compounds [I] synthesized by the procedure of preparation process A using a conventional process for removing the amino protecting group of the amino acid such as catalytic reduction, or by treating with acids such as liquid hydrofluoric acid, sulfonic acids, hydrochloric acid, hydrobromic acid, formic acid and the like.

In the following, synthesis examples of amino-acid amide derivatives, which are novel intermediates of the compounds of the present invention represented by Formulae [X] and [XII], are provided as reference examples.

[Reference Example I]

Synthesis of 2-(4-cyanophenoxy)-1-methylethylamine (Intermediate Compound No. 1)

293 g of ammonium acetate and 16.7 g of sodium cyanoborohydride were added to a solution containing 66.5 g of 4-cyanophenoxyacetone dissolved in 1500 mL of methanol, and the resultant mixture was stirred for 30 hours at room temperature. The reaction mixture was then concentrated under reduced pressure, and acidified with concentrated hydrochloric acid. 500 mL of diethyl ether and 300 mL of water were then added thereto. Subsequently, the resultant water layer was made basic with a 5% aqueous solution of sodium hydroxide, the solution was extracted with 1000 mL of diethyl ether, and then washed with water. The organic layer was then dried over anhydrous sodium sulfate, and the diethyl ether was removed under reduced pressure. The obtained residue was distilled under reduced pressure to yield 13.0 g of the desired product (19%). Boiling point: 132 °C / 0.26 mmHg.

[Reference Example 2]

Synthesis of 2-(4-chloro-2-methylphenoxy)-1-methylethylamine (Intermediate Compound No. 2)

120 g of ammonium acetate and 9.8 g of sodium cyanoborohydride were added to a solution containing 31 g of (4-chloro-2-methylphenoxy)acetone dissolved in 700 mL of methanol, and the resultant mixture was stirred for 20 hours at room temperature. After the reaction mixture was concentrated under reduced pressure, 180 mL of concentrated hydrochloric acid and 100 mL of water were added to the residue. The whole mixture was stirred for 1 hour, and then extracted with 300 mL of diethyl ether. The water layer was alkalified using a 5% aqueous solution of sodium hydroxide, and then extracted with 500 mL of ethyl acetate. The organic layer was washed with water, dried over anhydrous magnesium sulfate, and then concentrated under reduced pressure. A fraction possessing a low boiling point was removed from the obtained oily products to afford 25 g (yield 81 %) of the desired product. Refractive index: 1.5360.

[Reference Example 3]

Synthesis of 2-(4-chlorophenoxy)-1-methylpropylamine (Intermediate Compound No. 3)

82 g of ammonium acetate and 6.7 g of sodium cyanoborohydride were added to a solution containing 21 g of 3-(4-chlorophenoxy)-2-butanone dissolved in 500 mL of methanol, and the reaction mixture was stirred for 20 hours at room temperature. The reaction mixture was then concentrated under reduced pressure, and 180 mL of concentrated hydrochloric acid and 100 mL of water were added to the residue. The whole mixture was extracted with 300 mL of diethyl ether. The obtained water layer was alkalified using a 5% aqueous solution of sodium hydroxide, and then extracted with 500 mL of ethyl acetate. The organic

layer was washed with water, dried over anhydrous magnesium sulfate, and then concentrated under reduced pressure. A fraction possessing a low boiling point was removed from the obtained oily products to afford 18 g (yield 86%) of the desired product. Refractive index: 1.5360.

[Reference Example 4]

Synthesis of 1-methyl-2-(2-methylphenoxy)ethylamine (Intermediate Compound No. 4)

A solution containing 36 g of 2-(2-methylphenoxy)acetone oxime O-methyl ether dissolved in 150 mL of dimethoxyethane was added dropwise to a suspension containing 13 g of sodium borohydride in 500 mL of dimethoxyethane at room temperature. After the mixture was stirred for 15 minutes at room temperature, a solution containing 66 g of trifluoroborane diethyl ether complex dissolved in 100 mL of dimethoxyethane was added dropwise to the mixture at room temperature. The reaction mixture was stirred for 30 minutes at room temperature and then refluxed for 3 hours. The resultant mixture was allowed to sit and cool naturally to room temperature and then acidified using a 10% hydrochloric acid. The dimethoxyethane layer was concentrated and combined with the water layer. The mixture was alkalified using sodium carbonate, and then extracted with dichloromethane, followed by washing with water. The organic layer was dried over anhydrous magnesium sulfate, and then the dichloromethane was removed under reduced pressure. The residue was distilled under reduced pressure to obtain 6.4 g (yield 21%) of the desired product. Boiling point: 65 °C / 0.08 mmHg.

[Reference Example 5]

Synthesis of 2-(4-cyanophenoxy)-1-methylethylamine (Intermediate Compound No. 1)

50.0 g of 2-amino-1-propanol was added dropwise to a stirred mixture of 29.3 g of 60% sodium hydride and 300 mL of N,N-dimethylformamide at 0 °C. After the reaction mixture was stirred for 30 minutes at 0 °C, a solution containing 121.2 g of 4-bromobenzonitrile dissolved in N,N-dimethylformamide was added dropwise to the reaction mixture. The resultant mixture was stirred for 20 hours at room temperature. After completion of the reaction, the resultant mixture was poured into water and extracted with ethyl acetate. The organic layer was washed with water and then dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The residue was distilled under reduced pressure to obtain 48.0 g of the desired product (yield 41%). Boiling point: 132 °C / 0.26 mmHg.

[Reference Example 6]

Synthesis of (-)-2-(4-cyanophenoxy)-1-methylethylamine (Intermediate Compound No. 5)

25.0 g of R-(-)-2-amino-1-propanol was added dropwise to a stirred mixture of 14.0 g of 60% sodium hydride and 200 mL of N,N-dimethylformamide at a temperature of 5 °C to 10 °C. After the reaction mixture was stirred for 30 minutes, a solution containing 45.0 g of 4-chlorobenzonitrile dissolved in N,N-dimethylformamide was added dropwise to the reaction mixture. The reaction mixture was stirred for 20 hours at room temperature. After completion of the reaction, the reaction mixture was poured into water and extracted with ethyl acetate. The organic layer was washed with water and then dried over anhydrous magnesium sulfate. The ethyl acetate was removed under reduced pressure. The residue was distilled under reduced pressure to obtain 33.0 g of the desired product (yield: 56%). Boiling point: 60 - 66 °C / 0.08 mmHg, $[\alpha]D^{20}$ -15.7° (C 1.0 CH_3OH).

[Reference Example 7]

Synthesis of 1-methyl-2-(2-pyrimidyloxy)ethylamine (Intermediate Compound No. 6)

2.0 g of 2-amino-1-propanol was added dropwise to a stirred mixture of 1.3 g of 60% sodium hydride and 30 mL of N,N-dimethylformamide at room temperature. After the reaction mixture was stirred for 30 minutes, a solution containing 3.7 g of 2-chloropyrimidine dissolved in N,N-dimethylformamide was added dropwise to the reaction mixture. The mixture was stirred for 2 hours at 100 °C. After completion of the reaction, the reaction mixture was cooled. The solids were filtered off. The solvent in the filtrate was removed under reduced pressure. The residue was purified by column chromatography on silica gel to

obtain 2.1 g of the desired product (yield: 50%). Refractive index: 1.5481.

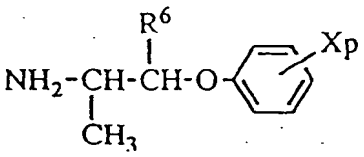
[Reference Example 8]

Synthesis of 1-methyl-2-(4-pyridyloxy)ethylamine (Intermediate Compound No. 7)

6.2 g of 2-amino-1-propanol was added dropwise to a stirred mixture of 4.0 g of 60% sodium hydride and 50 mL of N,N-dimethylformamide at 5 °C - 10 °C. After the reaction mixture was stirred for 30 minutes, 12.5 g of 4-chloropyridine hydrochloride in limited amounts was added to the reaction mixture. The mixture was stirred for 20 hours at room temperature. After completion of the reaction, the solids were filtered off. The solvent in the filtrate was removed under reduced pressure. The residue was purified by column chromatography on silica gel to obtain 3.8 g of the desired product (yield: 30%). Refractive index: 1.5469.

Specific examples of intermediate [X] obtained through the operations of Reference Examples 1 to 8 are shown in Table 13.

Table 13



Intermidiate Compound No.	R^6	Xp	Reflactive Index (n_D^{20}) or Boiling Point (°C/mmHg)
8	H	2-OCH_3	96.5/0.15
9	H	3-OCH_3	1.5158
10	H	4-OCH_3	95/0.10
11	H	2-CN	1.5566
12	H	3-CN	1.5409
13	H	2-F	70/0.22
14	H	3-F	74/0.15
15	H	2-NO_2	1.5582
16	H	2,4-Cl_2	1.5475
17	H	3,4-Cl_2	107/0.16
18	H	3,5-Cl_2	100/0.12
19	H	3,4-$(OCH_3)_2$	1.5361
20	H	3,5-$(OCH_3)_2$	12.5/0.10
21	CH_3	4-CN	1.5480
22	CH_3	4-NO_2	1.6263

[Reference Example 9]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide hydrochloride (Intermediate Compound No. 23)

Hydrogen chloride gas was introduced into a solution containing 3.7g of N^2-tert-butoxycarbonyl-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide dissolved in 100 mL of methylene chloride for 1 hour at room temperature. After completion of the reaction, the methylene chloride was removed under reduced pressure, thus obtaining a crude crystal. The crude crystal was washed with acetone to afford 3.1 g of the desired product (yield: 100%). Melting point: 59 - 63 °C.

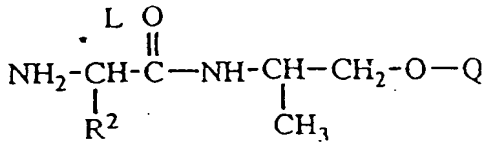
[Reference Example 10]

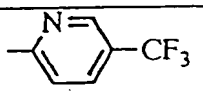
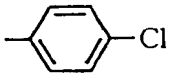
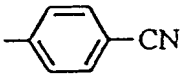
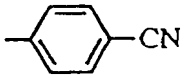
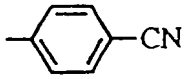
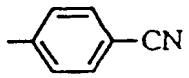
Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-isoleucinamide (Intermediate Compound No. 24)

Hydrogen chloride gas was introduced into a solution containing 15.0 g of N^2-tert-butoxycarbonyl-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-isoleucinamide dissolved in 300 mL of methylene chloride for 1 hour at room temperature. After completion of the reaction, the methylene chloride was removed under reduced pressure, thus obtaining a crude crystal. 200 ml of a saturated aqueous solution of sodium bicarbonate and 200 ml of methylene chloride were added to the crude crystal, and the mixture was stirred for 30 minutes and extracted with methylene chloride. The organic layer was washed with water, and dried over anhydrous sodium sulfate. The methylene chloride was removed under reduced pressure. The obtained crude product was washed with acetone to afford 10.0 g of the desired product (yield: 90%). Melting point: 64 - 67 °C.

Specific examples of Intermediate [XII] obtained through the operations of Reference Examples 9 and 10 are shown in Table 14.

Table 14



Intermidiate Compound No.	R^2	Q	Reflactive Index (n_D^{20}) or Boiling Point (°C/mmHg)
25	C_3H_7-i		73-75
26	C_2H_5		43-44
27	C_2H_5		1.5391
28	C_3H_7		1.5299
29	C_4H_9-t		1.5251
30	C_3H_7-i		1.5250

[The Best Mode for Carrying Out the Invention]

The methods for producing the compounds according to the present invention as well as the use of the compounds will be described in detail in the following Synthesis Examples.

[Synthesis Example 1]

Synthesis of N^2-*tert*-butoxycarbonyl-N^1-[1-methyl-2-(4-nitrophenoxy)ethyl]-L-valinamide (Compound No. 16)

0.5 g of N-methylpiperidine was added to a solution containing 1.1 g of N-*tert*-butoxycarbonyl-L-valine dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 1 g of 1-methyl-2-(4-nitrophenoxy)ethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the dichloromethane layer was successively washed with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue, which was a crude crystal, was purified by column chromatography on silica gel, thus obtaining 0.7 g of the desired product in the form of a yellow powder (yield: 55%).

[Synthesis Example 2]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-isopropenyloxycarbonyl-L-valinamide (Compound No. 77)

0.6 g of N-methylmorpholine, and subsequently 0.4 g of isopropyl chloroformate were added to a solution containing 0.9 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]L-valinamide hydrochloride dissolved in 50 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the dichloromethane layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.23 g of the desired product in the form of colorless grains (yield: 13%).

[Synthesis Example 3]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-phenoxycarbonyl-L-valinamide (Compound No. 107)

1.3 g of N-methylpiperidine was added to a solution containing 3 g of N-phenoxycarbonyl-L-valine dissolved in 50 mL of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 1.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 2.2 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature, with stirring, and stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue, which was a crude crystal, was purified by column chromatography on silica gel, thus obtaining 1.1 g of the desired product in the form of a white powder (yield: 22%).

[Synthesis Example 4]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-isoleucinamide (Compound No. 228)

1.3 g of N-methylpiperidine was added to a solution containing 3 g of N-tert-butoxycarbonyl-L-isoleucine dissolved in 60 mL of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 1.8 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 2.3 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.6 g of the desired product in the form of a white powder (yield: 12%).

[Synthesis Example 5]

Synthesis of N^2-tert-butoxycarbonyl-N^1-(2-phenylthioethyl)-L-valinamide (Compound No. 551)

1 g of N-methylpiperidine was added to a solution containing 2.1 g of N-tert-butoxycarbonyl-L-valine dissolved in 40 mL of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 1.3 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 1.5 g of 2-phenylthioethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under

reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.4 g of the desired product in the form of cream yellow grains (yield: 12%).

[Synthesis Example 6]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[1-methyl-2-(4-nitrophenoxy)propyl]-L-valinamide (Compound No. 606)

0.5 g of N-methylpiperidine was added to a solution containing 1 g of N-tert-butoxycarbonyl-L-valine dissolved in 40 mL of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at the same temperature, 0.7 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 1 g of 1-methyl-2-(4-nitrophenoxy)propylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue, which was an oily substance, was purified by column chromatography on silica gel, thus obtaining 1.1 g of the desired product in the form of yellow viscous liquid (yield: 56%).

[Synthesis Example 7]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[2-(3,5-dimethoxyphenoxy)-1-methylethyl]-L-valinamide (Compound No. 22)

0.5 g of N-methylpiperidine was added to a solution containing 1.0 g of N-tert-butoxycarbonyl-L-valine dissolved in 100 mL of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 1 g of 2-(3,5-cyanophenoxy)-1-methylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue, which was a crude crystal, was purified by column chromatography on silica gel, thus obtaining 1.3 g of the desired product in the form of white powder (yield: 64%).

[Synthesis Example 8]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[1-methyl-2-(2,4,6-trichlorophenoxy)ethyl]-L-valinamide (Compound No. 25)

1.7 g of N-methylpiperidine was added to a solution containing 3.8 g of N-tert-butoxycarbonyl-L-valine dissolved in 80 mL of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at the same temperature, 2.4 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 4.5 g of 1-methyl-2-(2,4,6-trichlorophenoxy)ethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 4.6 g of the desired product in the form of a colorless needle crystal (yield: 58%).

[Synthesis Example 9]

Synthesis of N^2-isopropoxycarbonyl-N^1-[1-methyl-2-(4-nitrophenoxy)ethyl]-L-valinamide (Compound No. 45)

1.2 g of N-methylpiperidine was added to a solution containing 2.5 g of N-isopropoxycarbonyl-L-valine dissolved in 100 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 1.7 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 2.2 g of 2-(4-nitrophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.3 g of the desired product in the form of a yellow vitrified substance (yield: 6%).

^{1}H-NMR: ($CDCl_3$, δ)
1.16 - 1.33 (6H, m)
1.43 - 1.36 (9H, m)
2.56 (1H, m)
4.01 (2H, m)
4.00 - 5.33 (3H, m)
6.17 (1H, d)
6.87 (2H, d)
8.06 (2H, d)

[Synthesis Example 10]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-cyclohexyloxycarbonyl-L-valinamide (Compound No. 97)

0.8 g of N-methylpiperidine was added to a solution containing 2.0 g of N-cyclohexyloxycarbonyl-L-valine dissolved in 150 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 1.1 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 1.5 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.5 g of the desired product in the form of light brown powder (yield: 16%).

[Synthesis Example 11]

Synthesis of N^1-[1-methyl-2-(4-trifluoromethylphenoxy)ethyl]-N^2-phenoxycarbonyl-L-valinamide (Compound No. 114)

1.6 g of N-methylpiperidine was added to a solution containing 4.0 g of N-phenoxycarbonyl-L-valine dissolved in 80 ml of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at the same temperature, 2.2 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 3.5 g of 1-methyl-2-(4-trifluoromethylphenoxy)ethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 2.8 g of the desired product in the form of a white crystal (yield: 40%).

[Synthesis Example 12]

Synthesis of N^1-[1-methyl-2-(4-trifluoromethoxyphenoxy)ethyl]-N^2-phenoxycarbonyl-L-valinamide (Compound No. 115)

1.7 g of N-methylpiperidine was added to a solution containing 4.0 g of N-phenoxycarbonyl-L-valine dissolved in 80 ml of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at the same temperature, 2.3 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C.

4.0 g of 1-methyl-2-(4-trifluoromethoxyphenoxy)ethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 3.4 g of the desired product in the form of a white crystal (yield: 45%).

[Synthesis Example 13]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-phenoxycarbonyl-L-valinamide (Compound Nos. 116 and 117)

1.8 g of N-methylpiperidine was added to a solution containing 4.2 g of N-phenoxycarbonyl-L-valine dissolved in 100 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.4 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 3.1 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.0 g of a white powder. 0.6 g of the obtained white powder was purified using high pressure liquid chromatography (hereinafter, referred to as "HPLC") (YMC-063-15, hexane / ethyl acetate = 55 / 45) to separate two fractions. The ingredient of the first fraction possessing a short retention time was 0.3 g of a white powder (yield: 7%) possessing 145 to 147 °C of melting point and the ingredient of the second fraction possessing a long retention time was 0.3 g of a white powder (yield: 7%) possessing a melting point of 166 to 170 °C of melting point.

[Synthesis Example 14]

Synthesis of N^2-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-(3-methoxyphenoxycarbonyl)-L-valinamide (Compound No. 166)

1.0 g of N-methylmorpholine was added to a solution containing 1.5 g of N^1 -[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide hydrochloride dissolved in 100 ml of methylene chloride, at -20 °C. After 0.9 g of 3-methoxyphenyl chloroformate was added to the mixture at -20 °C, the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 2 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.25 g of the desired product in the form of a white plated crystal (yield: 12%).

[Synthesis Example 15]

Synthesis of N^2-(2-chloroethoxycarbonyl)-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide (Compound No. 184)

0.5 g of N-methylpiperidine was added to a solution containing 1.1 g of N-(2-chloroethoxycarbonyl)-L-valine dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at

the same temperature, 0.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 0.9 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue, which was an oily substance, was purified by column chromatography on silica gel, thus obtaining 1.0 g of the desired product in the form of colorless grains (yield: 52%).

[Synthesis Example 16]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-(4-methylbenzyloxycarbonyl)-L-valinamide (Compound No. 195)

0.6 g of N-methylpiperidine was added to a solution containing 1.5 g of N-(4-methylbenzyloxycarbonyl)-L-valine dissolved in 100 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.8 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 1.0 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue, which was a crude crystal, was purified by column chromatography on silica gel, thus obtaining 0.6 g of the desired product in the form of light white powder (yield: 28%).

[Synthesis Example 17]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-phenoxythiocarbonyl-L-valinamide (Compound No. 208)

0.4 g of N-methylmorpholine was added to a suspension containing 1.1 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamidesuspended in 40 ml of methylene chloride, at -15 °C. After 0.7 g of phenyl chlorothionoformate was added to the mixture at -15 °C, the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.2 g of the desired product in the form of a yellow glutinous substance (yield: 75%).

^{1}H-NMR: ($CDCl_3$, δ)
1.05 (6H, m)
1.35 (3H, m)
2.30 (1H, m)
4.00 (2H, m)
4.44 (1H, m)
4.54 (1H, m)
6.16, 6.25 (1H, d)
7.26 (9H, m)
7.51 (1H, br)

[Synthesis Example 18]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-(phenylthio)thiocarbonyl-L-valinamide (Compound No. 211)

0.5 g of N-methylmorpholine was added to a suspension containing 1.4 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamidesuspended in 40 ml of methylene chloride, at -15 °C. After 0.9 g of phenyl

chlorodithioformate was added to the mixture at -15 °C, the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.4 g of the desired product in the form of a yellow glutinous substance (yield: 66%).

^{1}H-NMR: ($CDCl_3$, δ)
0.83 (6H, m)
1.30, 1.32 (3H, d)
2.13 (1H, m)
3.96 (2H, m)
4.35 (1H, m)
4.78 (1H, dd)
6.04, 6.13 (1H, d)
6.93, 6.98 (2H, d)
7.15, 7.22 (1H, d)
7.57 (7H, m)

[Synthesis Example 19]

Synthesis of N^1-(1-methyl-2-phenylthioethyl)-N^2-phenoxycarbonyl-L-valinamide (Compound No. 212)

1.3 g of N-methylmorpholine was added to a suspension containing 3.0 g of N^1-(1-methyl-2-phenylthioethyl)-L-valinamide hydrochloride suspended in 80 ml of methylene chloride, at -15 °C. After 1.9 g of phenyl chloroformate was added to the mixture at - 15 °C, the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 2.3 g of the desired product in the form of a white crystal (yield: 54%).

[Synthesis Example 20]

Synthesis of N^1-[2-(4-chloroanilino)-1-methylethyl]-N^2-isopropoxycarbonyl-L-valinamide (Compound No. 221)

1.9 g of N-methylpiperidine was added to a solution containing 3.8 g of N-isopropoxycarbonyl-L-valine dissolved in 80 ml of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at the same temperature, 2.6 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 3.5 g of 2-(4-chloroanilino)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 3.3 g of the desired product in the form of a white crystal (yield: 47%).

[Synthesis Example 21]

Synthesis of 2-tert-butoxycarbonylamino-N-[2-(4-chlorophenoxy)-1-methylethyl]-(2S)-butyramide (Compound No. 233)

2.0 g of N-methylpiperidine was added to a solution containing 4.1 g of (2S)-2-tert-butoxycarbonylaminobutyric acid dissolved in 60 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 3.7 g of 2-(4-chlorophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the

reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 5.6 g of the desired product in the form of a colorless glutinous substance (yield: 76%).

[Synthesis Example 22]

Synthesis of 2-tert-butoxycarbonylamino-N-[2-(4-cyanophenoxy)-1-methylethyl]-(2S)-butyramide (Compound No. 235)

0.5 g of N-methylpiperidine was added to a solution containing 1.0 g of (2S)-2-tert-butoxycarbonylaminobutyric acid dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.7 g of isobutyl chloroformate was added to the mixture at -20 °C, and stirred for 1 hour at -20 °C. 0.9 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.0 g of the desired product in the form of a glutinous substance (yield: 54%).

^{1}H-NMR: ($CDCl_3$, δ)
0.94 (3H, t)
1.20 - 1.50 (12H, m)
1.69 (2H, m)
3.83 - 4.56 (4H, m)
5.30 (1H, d)
6.60 (1H, m)
6.90 (2H, d)
7.50 (2H, d)

[Synthesis Example 23]

Synthesis of N^1-[2-(4-chlorobenzyloxy)-1-methylethyl]-N^2-isopropoxycarbonyl-L-valinamide (Compound No. 246)

0.5 g of N-methylpiperidine was added to a solution containing 1 g of N-isopropoxycarbonyl-L-valine dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at the same temperature, 0.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 1 g of 2-(4-chlorobenzyloxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained oily residue was purified by column chromatography on silica gel, thus obtaining 0.9 g of the desired product in the form of a colorless plated crystal (yield: 48%).

[Synthesis Example 24]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[1-methyl-2-(4-methylthiophenoxy)ethyl]-L-valinamide (Compound No. 327)

3.4 g of N-methylpiperidine was added to a solution containing 7.5 g of N-tert-butoxycarbonyl-L-valine dissolved in 100 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 4.7 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 6.8 g of 1-methyl-2-(4-methylthiophenoxy)ethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was

stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The obtained oily residue was purified by column chromatography on silica gel, thus obtaining 6.2 g of the desired product in the form of a colorless prism-shaped crystal (yield: 46%).

[Synthesis Example 25]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[1-methyl-2-(4-methylsulfinylphenoxy)ethyl]-L-valinamide (Compound No. 328)

1.5 g of m-chloroperbenzoic acid was added to a solution containing 3.0 g of N^2-tert-butoxycarbonyl-N^1-[1-methyl-2-(4-methylthiophenoxy)ethyl]-L-valinamide dissolved in 60 ml of methylene chloride, at 0 °C. After the mixture was stirred for 5 hours at room temperature, the reaction mixture was filtered. The filtrate was washed successively with a saturated aqueous solution of potassium carbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The obtained oily residue was purified by column chromatography on silica gel, thus obtaining 1.7 g of the desired product in the form of a colorless crystal (yield: 56%).

[Synthesis Example 26]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[1-methyl-2-(4-methylsulfonylphenoxy)ethyl]-L-valinamide (Compound No. 329)

2.1 g of m-chloroperbenzoic acid was added to a solution containing 2.0 g of N^2-tert-butoxycarbonyl-N^1-[1-methyl-2-(4-methylthiophenoxy)ethyl]-L-valinamide dissolved in 50 ml of methylene chloride, at 0 °C. After the mixture was stirred for 8 hours at a reflux temperature, the reaction mixture was allowed to sit and cool naturally to room temperature and filtered. The filtrate was washed successively with a saturated aqueous solution of potassium carbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The obtained residue was purified by column chromatography on silica gel, thus obtaining 1.3 g of the desired product in the form of a colorless prism-shaped crystal (yield: 60%).

[Synthesis Example 27]

Synthesis of N^1-[2-(4-fluorophenylsulfinyl)-1-methylethyl]-N^2-isopropoxycarbonyl-L-valinamide (Compound No. 354)

1.3 g of m-chloroperbenzoic acid was added to a solution containing 2.5 g of N^1-[2-(4-fluorophenylthio)-1-methylethyl]-N^2-isopropoxycarbonyl-L-valinamide dissolved in 50 ml of methylene chloride, at 0 °C. After the mixture was stirred for 5 hours at room temperature, the reaction mixture was filtered. The filtrate was washed successively with a saturated aqueous solution of potassium carbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The obtained residue was purified by column chromatography on silica gel, thus obtaining 1.8 g of the desired product in the form of a colorless prism-shaped crystal (yield: 69%).

[Synthesis Example 28]

Synthesis of N^1-[2-(4-fluorophenylsulfonyl)-1-methylethyl]-N^2-isopropoxycarbonyl-L-valinamide (Compound No. 355)

3.4 g of m-chloroperbenzoic acid was added to a solution containing 2.2 g of N^1-[2-(4-fluorophenylthio)-1-methylethyl]-N^2-isopropoxycarbonyl-L-valinamide dissolved in 50 ml of methylene chloride, at 0 °C. After the mixture was stirred for 8 hours at a reflux temperature, the reaction mixture was allowed to sit and cool to room temperature, and then filtered. The filtrate was washed successively with a saturated aqueous solution of potassium carbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The obtained residue was purified by column chromatography on silica gel, thus obtaining 2.0 g of the desired product in the form of a white crystal

(yield: 83%).

[Synthesis Example 29]

Synthesis of N^2-isopropoxycarbonyl-N^1-[1-methyl-2-(2-methylphenylthio)ethyl]-L-valinamide (Compound No. 367)

1.9 g of N-methylpiperidine was added to a solution containing 3.9 g-of N-isopropoxycarbonyl-L-valine dissolved in 80 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.6 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 3.5 g of 1-methyl-2-(2-methylphenylthio)ethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 3.6 g of the desired product in the form of a white crystal (yield: 51%).

[Synthesis Example 30]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl)-N^2-(3-tetrahydrofuranyl)oxycarbonyl-L-valinamide (Compound No. 376)

1.0 g of N-methylmorpholine, and subsequently 0.7 g of 3-tetrahydrofuranyl chloroformate were added to a suspension containing 1.5 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide hydrochloride suspended in 100 ml of methylene chloride at -20 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 2 hours at room temperature. Water was subsequently added to the reaction mixture. After the dichloromethane layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.1 g of the desired product in the form of white powder (yield: 61%).

[Synthesis Example 31]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-(3-methylcyclohexyloxycarbonyl)-L-valinamide (Compound No. 379)

0.4 g of N-methylmorpholine, and subsequently 0.8 g of 3-methylcyclohexyl chloroformate were added to a suspension containing 1.0 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide suspended in 50 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.2 g of the desired product in the form of a white crystal (yield: 80%).

[Synthesis Example 32]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-propargyloxycarbonyl-L-valinamide (Compound No. 381)

0.2 g of N-methylmorpholine, and subsequently 0.2 g of propargyl chloroformate were added to a suspension containing 0.5 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide suspended in 30 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.5 g of the desired product in the form of white

powder (yield: 78%).

[Synthesis Example 33]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-(2-methoxy-1-methylethyl)oxycarbonyl-L-valinamide (Compound No. 383)

1.0g of N-methylmorpholine, and subsequently 0.7 g of 2-methoxy-1-methylethylchloroformate were added to a suspension containing 1.5 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamidesuspended in 150 ml of methylene chloride at -20 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 2 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.37 g of the desired product in the form of a white plated crystal (yield: 20%).

[Synthesis Example 34]

Synthesis of N^1-[2-(4-fluoro-N-methylanilino)-1-methylethyl]-N^2-phenoxycarbonyl-L-valinamide (Compound No. 391)

1.6 g of N-methylpiperidine was added to a solution containing 3.9 g of N-phenoxycarbonyl-L-valine dissolved in 80 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.2 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 3.0 g of 2-(4-fluoro-N-methylanilino)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.2 g of the desired product in the form of a white crystal (yield: 19%).

[Synthesis Example 35]

Synthesis of N^2-(4-chlorophenoxycarbonyl)-N^1-[2-(4-cyanophenoxy-1-methylethyl]-L-valinamide (Compound Nos. 395 and 396)

1.7 g of N-methylpiperidine was added to a solution containing 4.7 g of N-(4-chlorophenoxycarbonyl)-L-valine dissolved in 250 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.3 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 3.0 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 0.4 g of the desired product in the form of white powder. In addition, the powder was purified by HPLC (YMC-063-15, hexane / ethyl acetate = 55 / 45) to separate two fractions. One fraction possessing a short retention time was 0.17 g of white powder possessing 137 - 140 °C of melting point (yield: 2 %), and another fraction possessing a long retention time was 0.17 g of white powder possessing a melting point of 174 - 179 °C (yield: 2 %).

[Synthesis Example 36]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-(2-nitrophenoxycarbonyl-L-valinamide (Compound No. 400)

1.3g of N-methylmorpholine, and subsequently 2.5 g of 2-nitrophenyl chloroformate were added to a suspension containing 3.4g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide suspended in 100 ml of methylene chloride at -20 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 2 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.0 g of the desired product in the form of a yellow plated crystal (yield: 18 %).

[Synthesis Example 37]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-(4-fluorophenoxycarbonyl)-L-valinamide (Compound No. 401)

1.2 g of N-methylpiperidine was added to a solution containing 3.0 g of N-(4-fluorophenoxycarbonyl)-L-valine dissolved in 80 ml of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at the same temperature, 1.6 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 2.3 g of (-)-2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 1.1 g of the desired product in the form of a white crystal (yield: 23 %).

[Synthesis Example 38]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-(3,4-dimethylphenoxycarbonyl)-L-valinamide (Compound No. 403)

0.6 g of N-methylmorpholine, and subsequently 1.2 g of 3,4-dimethylphenyl chloroformate were added to a suspension containing 1.5 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide suspended in 50 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.7 g of the desired product in the form of a white crystal (yield: 74 %).

[Synthesis Example 39]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[2-(2-pyridyloxy)-1-methylethyl]-L-valinamide (Compound No. 409)

2.0 g of N-methylpiperidine was added to a solution containing 4.3 g of N-tert-butoxycarbonyl-L-valine dissolved in 80 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 3.3 g of 2-(2-pyridyloxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 2.0 g of the desired product in the form of colorless grains (yield: 28 %).

[Synthesis Example 40]

Synthesis of N^1-[2-(5-chloro-2-pyridyloxy)-1-methylethyl]-N^2-isopropyloxycarbonyl-L-valinamide (Compound No. 412)

0.8 g of N-methylmorpholine, and subsequently 0.5 g of isopropyl chloroformate were added to a suspension containing 1.4 g of N^1-[2-(5-chloro-2-pyridyloxy)-1-methylethyl]-L-valinamide hydrochloride suspended in 50 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.6 g of the desired product in the form of colorless grains (yield: 38 %).

[Synthesis Example 41]

Synthesis of N^1-[2-(5-chloro-2-pyridyloxy)-1-methylethyl]-N^2-phenoxycarbonyl-L-valinamide (Compound No. 413)

0.8 g of N-methylmorpholine, and subsequently 0.7 g of phenyl chloroformate were added to a suspension containing 1.4 g of N^1-[2-(5-chloro-2-pyridyloxy)-1-methylethyl]-L-valinamide hydrochloride suspended in 50 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.6 g of the desired product in the form of colorless grains (yield: 34 %).

[Synthesis Example 42]

Synthesis of N^1-[2-(4-fluoro-N-methylanilino)-1-methylethyl]-N^2-phenoxycarbonyl-L-isoleucinamide (Compound No. 422)

1.9 g of N-methylpiperidine was added to a solution containing 4.8 g of N-phenoxycarbonyl-L-isoleucine dissolved in 80 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.6 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 3.5 g of 2-(4-fluoro-N-methylanilino)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel thus obtaining 1.1 g of the desired product in the form of a white crystal (yield: 13 %).

[Synthesis Example 43]

Synthesis of N^2-(ethylthio)carbonyl-N^1-[1-methyl-2-(4-nitrophenoxy)ethyl]-L-valinamide (Compound No. 432)

0.3 g of N-methylmorpholine, and subsequently 0.4 g of ethyl chlorothioformate were added to a suspension containing 0.9 g of N^1-[1-methyl-2-(4-nitrophenoxy)ethyl]-L-valinamide suspended in 50 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.0 g of the desired product in the form of yellow grains (yield: 79 %).

[Synthesis Example 44]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-leucinamide (Compound No. 455)

1.5 g of N-methylpiperidine was added to a solution containing 3.4 g of N-tert-butoxycarbonyl-L-leucine dissolved in 60 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.0 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 2.6 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 5.1 g of the desired product in the form of a colorless glutinous substance (yield: 86 %).

^{1}H-NMR: ($CDCl_3$, δ)
0.92 (6H, m)
1.28, 1.32 (3H, d)
1.39, 1.43 (9H, s)
1.46, 1.65 (2H, m)
1.65 (1H, m)
3.98 (2H, m)
4.06 (1H, m)
4.35 (1H, m)
4.91 (1H, br)
6.46 (1H, br)
6.97 (2H, d)
7.57 (2H, dd)

[Synthesis Example 45]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-tert-leucinamide (Compound No. 457)

1.7 g of N-methylpiperidine was added to a solution containing 4 g of N-tert-butoxycarbonyl-L-tert-leucine dissolved in 50 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.4 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 3.1 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 3.9 g of the desired product in the form of a colorless amorphous substance (yield: 58 %).

[Synthesis Example 46]

Synthesis of 2-tert-butoxycarbonylamino-3-methyl-N-[2-(4-cyanophenoxy)-1-methylethyl]-3-butenic acid amide (Compound No. 460)

0.5 g of N-methylpiperidine was added to a solution containing 1.1 g of 2-tert-butoxycarbonylamino-3-methyl-3-butenic acid dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 1.9 g of 2-(4-cyanophenoxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the

reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 0.3 g of the desired product in the form of a colorless glutinous substance (yield: 32 %).

[Synthesis Example 47]

Synthesis of N-[2-(4-cyanophenoxy)-1-methylethyl]-2-isopropoxycarbonylaminocyclopentylacetic acid amide (Compound No. 462)

0.4 g of N-methylmorpholine, and subsequently 0.5 g of isopropyl chloroformate were added to a suspension containing 1.2 g of 2-amino-N-[2-(4-cyanophenoxy)-1-methylethyl]cyclopentylacetic acid amide suspended in 40 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.4 g of the desired product in the form of a colorless plated crystal (yield: 90 %).

[Synthesis Example 48]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-phenoxycarbonyl-L-norvalinamide (Compound No. 465)

0.5 g of N-methylmorpholine, and subsequently 0.8 g of phenyl chloroformate were added to a suspension containing 1.4 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-norvalinamide suspended in 40 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.1 g of the desired product in the form of a colorless plated crystal (yield: 57 %).

[Synthesis Example 49]

Synthesis of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-N^2-phenoxycarbonyl-L-leucinamide (Compound No. 466)

0.5 g of N-methylmorpholine, and subsequently 0.8 g of phenyl chloroformate were added to a suspension containing 1.5 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-leucinamide suspended in 40 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.5 g of the desired product in the form of colorless powder (yield: 73 %).

[Synthesis Example 50]

Synthesis of 2-(4-chlorophenoxycarbonylamino)-N-[2-(4-cyanophenoxy)-1-methylethyl]cyclopentylacetic acid amide (Compound No. 471)

0.4 g of N-methylmorpholine, and subsequently 0.8 g of 4-chlorophenyl chloroformate were added to a suspension containing 1.2 g of 2-amino-N-[2-(4-cyanophenoxy)-1-methylethyl]cyclopentylacetic acid amide suspended in 40 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried

over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.6 g of the desired product in the form of colorless grains (yield: 30 %).

[Synthesis Example 51]

Synthesis of N^2-benzyloxycarbonyl-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-(4-chlorophenyl)glycinamide (Compound No. 475)

0.4 g of N-methylmorpholine, and subsequently 0.6 g of benzyl chloroformate were added to a suspension containing 1.3 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-(4-chlorophenyl)glycinamide suspended in 40 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.2 g of the desired product in the form of colorless grains (yield: 70 %).

[Synthesis Example 52]

Synthesis of N^2-(1-cyano-1-methylethoxycarbonyl)-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide (Compound No. 476)

0.5 g of N-methylmorpholine, and subsequently 0.4 g of 1-cyano-1-methylethyl chloroformate were added to a suspension containing 0.7 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide hydrochloride suspended in 50 ml of methylene chloride at -20 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 3 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.6 g of the desired product in the form of colorless grains (yield: 71 %).

[Synthesis Example 53]

Synthesis of N^2-(2-chlorocyclohexyloxycarbonyl)-N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide (Compound No. 477)

0.4 g of N-methylmorpholine, and subsequently 0.9 g of 2-chlorocyclohexyl chloroformate were added to a suspension containing 1.0 g of N^1-[2-(4-cyanophenoxy)-1-methylethyl]-L-valinamide hydrochloride suspended in 50 ml of methylene chloride at -20 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.1 g of the desired product in the form of a white crystal (yield: 71 %).

[Synthesis Example 54]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[2-(3-chloro-5-trifluoromethyl-2-pyridyloxy)-1-methylethyl]-L-valinamide (Compound No. 479)

2.0 g of N-methylpiperidine was added to a solution containing 5.6 g of N-tert-butoxycarbonyl-L-valine dissolved in 100 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 2.7 g of isobutyl chloroformate was added to the mixture at -40 °C, and stirred for 1 hour at -20 °C. 1.5 g of 2-(3-chloro-5-trifluoromethyl-2-pyridyloxy)-1-methylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature, with stirring, and stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium

bicarbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 7.0 g of the desired product in the form of colorless grains (yield: 77 %).

[Synthesis Example 55]

Synthesis of N^1-[1-(5-chloro-6-ethyl-4-pyrimidinyloxy)-2-propyl]-N^2-isopropoxycarbonyl-L-valinamide (Compound No. 481)

0.34 g of N-methylpiperidine was added to a solution containing 0.7 g of N-isopropoxycarbonyl-L-valine dissolved in 50 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.47 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 0.74 g of 1-(5-chloro-6-ethyl-4-pyrimidinyloxy)-2-propylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 0.6 g of the desired product in the form of a white prism-shaped crystal (yield: 43 %).

[Synthesis Example 56]

Synthesis of N-tert-butoxycarbonyl-L-valyl-N-(4-chlorophenyl)-N-methyl-DL-alaninamide (Compound No. 490)

0.9 g of N-methylpiperidine was added to a solution containing 2.0 g of N-tert-butoxycarbonyl-L-valine dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 1.3 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 2.0 g of N^1-(4-chlorophenyl)-N^1-methyl-DL-alaninamide was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 3.4 g of the desired product in the form of a colorless needle crystal (yield: 87 %).

[Synthesis Example 57]

Synthesis of N-isopropoxycarbonyl-L-isoleucyl-N-(4-cyanophenyl)-D-alaninamide (Compound No. 506)

0.26 g of N-methylpiperidine was added to a solution containing 0.57 g of N-isopropoxycarbonyl-L-isoleucine dissolved in 60 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.36 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 0.5 g of N^1-(4-cyanophenyl)-D-alaninamide was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.5 g of the desired product in the form of white powder (yield: 49 %).

[Synthesis Example 58]

Synthesis of N-cyclohexyloxycarbonyl-L-valyl-N-(4-cyanophenyl)-D-alaninamide (Compound No. 509)

0.6 g of N-methylmorpholine, and subsequently 0.6 g of cyclopentyl chloroformate were added to a suspension containing 1.0 g of L-valyl-N-(4-cyanophenyl)alaninamide, hydrochloride suspended in 50 ml of methylene chloride at -20 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.6 g of the desired product in the form of a white crystal (yield: 49 %).

[Synthesis Example 59]

Synthesis of N-phenoxycarbonyl-L-valyl-N-(4-chlorobenzyl)-DL-alaninamide (Compound No. 516)

0.55 g of N-methylmorpholine, and subsequently 0.43 g of phenyl chloroformate were added to a suspension containing 0.95 g of L-valyl-N-(4-chlorobenzyl)-DL-alaninamide, hydrochloride suspended in 50 ml of methylene chloride at -15 °C. The mixture was allowed to sit and warm naturally to room temperature and stirred for 15 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed with water, the organic layer was dried over anhydrous magnesium sulfate and then the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.9 g of the desired product in the form of white powder (yield: 75 %).

[Synthesis Example 60]

Synthesis of N-phenoxycarbonyl-L-valyl-DL-alanine phenyl ester (Compound No. 522)

0.24 g of N-methylpiperidine was added to a solution containing 0.57 g of N-phenoxycarbonyl-L-valine dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.33 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 0.5 g of DL-alanine phenyl ester was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.2 g of the desired product in the form of white powder (yield: 20 %).

[Synthesis Example 61]

Synthesis of N^1-(4-cyanophenyl)-N^2-(2-phenoxycarbonylamino)-(2S)-butyryl-D-alaninamide (Compound No. 524)

0.45 g of N-methylpiperidine was added to a solution containing 1.0 g of (2S)-2-phenoxycarbonylaminobutyric acid dissolved in 50 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.61 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 0.85 g of N^1-(4-cyanophenyl)-D-alaninamide was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained crude crystal was purified by column chromatography on silica gel, thus obtaining 0.8 g of the desired product in the form of white powder (yield: 45 %).

[Synthesis Example 62]

Synthesis of N-isopropoxycarbonyl-L-valyl-N-(4-cyanophenyl)glycinamide (Compound No. 526)

0.3 g of N-methylpiperidine was added to a solution containing 0.6 g of N-isopropoxycarbonyl-L-valine dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.4 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 0.5 g of N^1-(4-cyanophenyl)glycinamide was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 0.5 g of the desired product in the form of colorless powder (yield: 49 %).

[Synthesis Example 63]

Synthesis of N^2-tert-butoxycarbonyl-N^1-(1,2-dimethyl-2-phenoxyethyl)-L-valinamide (Compound No. 602)

0.6 g of N-methylpiperidine was added to a solution containing 1.3 g of N-tert-butoxycarbonyl-L-valine dissolved in 40 ml of methylene chloride, at -20 °C. After the mixture was stirred for 15 minutes at the same temperature, 0.8 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 1 g of 1,2-dimethyl-2-phenoxyethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous magnesium sulfate and the methylene chloride was removed under reduced pressure. The obtained oily substance was purified by column chromatography on silica gel, thus obtaining 1.3 g of the desired product in the form of a white glutinous substance (yield: 57 %).

^{1}H-NMR: ($CDCl_3$, δ)
0.8 - 1.02 (6H, m)
1.18 - 1.45 (15H, m)
2.10 (1H, m)
3.65 - 4.45 (3H, m)
5.18 (1H, m)
6.38 (1H, m)
6.72 - 7.35 (5H, m)

[Synthesis Example 64]

Synthesis of N^2-tert-butoxycarbonyl-N^1-[2-(4-cyanophenoxy)-1,2-dimethylethyl]-L-valinamide (Compound No. 607)

0.5 g of N-methylpiperidine was added to a solution containing 1.1 g of N-tert-butoxycarbonyl-L-valine dissolved in 60 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.7 g of isobutyl chloroformate was added to the mixture, and stirred for 1 hour at -20 °C. 1.0 g of 2-(4-cyanophenoxy)-1,2-dimethylethylamine was added to this mixture at -60 °C, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 20 hours at room temperature. Water was subsequently added to the reaction mixture. After the methylene chloride layer was washed successively with a 5% aqueous solution of sodium bicarbonate and water, the organic layer was dried over anhydrous sodium sulfate and the methylene chloride was removed under reduced pressure. The residue was purified by column chromatography on silica gel, thus obtaining 1.2 g of the desired product in the form of a colorless glassy substance (yield: 61 %).

^{1}H-NMR: ($CDCl_3$, δ)
0.79- 1.03 (6H, m)
1.15 - 1.46 (15H, m)
2.03 (1H, m)
3.63 - 4.72 (3H, m)

5.06 (1H, m)
6.30 (1H, m)
6.83 - 7.60 (4H, m)

[Synthesis Example 65]

Synthesis of N^1-[2-(4-cyanophenoxy)propyl]-N^2-phenoxycarbonyl-L-valinamide (Compound No. 750)

0.16 g of N-methylpiperidine was added to a suspension containing 0.25 g of N^1-[2-(4-cyanophenoxy)-propyl]-L-valinamide hydrochloride suspended in 20 ml of methylene chloride, at -20 °C. After the mixture was stirred for 10 minutes at the same temperature, 0.13 g of phenyl chloroformate was added drop by drop to the mixture, and then the reaction mixture was allowed to sit and warm naturally to room temperature while being stirred, and was stirred for 3 hours at room temperature. After the methylene chloride was removed under reduced pressure, the residue was purified by column chromatography on silica gel, thus obtaining 0.2 g of the desired product in the form of a white glutinous substance (yield: 63 %).

^{1}H-NMR: ($CDCl_3$, δ)
1.00 (6H, m)
1.23 (3H, d)
2.13 (1H, m)
3.31 (1H, m)
4.00 (2H, m)
4.49 (1H, m)
5.93 (1H, d)
6.52 (1H, m)
6.80 - 7.56 (9H, m)

The agricultural or horticultural fungicide according to the present invention is a composition containing an amino acid amide derivative represented by Formula [I] as an active ingredient. In the case where the compounds according to the present invention are employed as an agricultural or horticultural fungicide, the compounds acting as the active ingredient can be formulated appropriately, depending on the purpose. The active ingredient is usually diluted in an inert liquid or a solid carrier, and a surfactant or the like is added thereto, if necessary. The mixture is then formulated in a known-manner into, for example, a fine powder, a wettable powder, an emulsifiable concentrate, granules, or the like.

Suitable examples of carriers employed in the formulation are solid carriers such as talc, bentonite, clay, kaolin, diatomaceous earth, white carbon, vermiculite, slaked lime, siliceous sand, ammonium sulfate, urea, or the like; and liquid carriers such as isopropyl alcohol, xylene, cyclohexanone, methylnaphthalene, and the like. Illustrative examples of the surfactants and dispersants include salts of dinaphthylmethanesulfonic acid, sulfate esters of alcohol, alkylarylsulfonic acid, and ligninesulfonic acid, polyoxyethylene glycol ethers, polyoxyethylene alkyl aryl ethers, polyoxyethylenesorbitan monoalkylates, and the like. Suitable examples of auxiliary agents include carboxymethylcellulose, and the like. These preparations can be applied directly, or after diluting the preparation to a suitable concentration.

The agricultural or horticultural fungicide according to the present invention can be employed for a number of purposes: for example, treating seeds, spraying of stem and leaf portions, injection into irrigation water, and applying to the soil. The proportion of the active ingredient is selected as needed. When formulated into a fine powder or granules, 0.1% by weight to 20% by weight of the active ingredient is preferred. For an emulsifiable concentrate or wettable powder, 5% by weight to 80% by weight of the active ingredient is adequate.

The rate of application of the agricultural or horticultural fungicide according to the present invention may vary depending on the type of active compound employed, the kind of the pest or disease to be controlled, the nature of occurrence of the pest or disease, the degree of damage, environmental conditions, the form of preparation to be used, and the like. When the agricultural or horticultural fungicide of the present invention is applied directly in the form of fine powder or granules, it is recommended that the rate of application of the active ingredient be suitably chosen within the range of 0.1 g to 5 kg per 10 ares, preferably, in the range of 1 g to 1 kg per 10 ares. In addition, when the fungicide of the present invention is in the form of a liquid such as an emulsifiable concentrate or a wettable powder, it is recommended that the ratio for application of the active ingredient be suitably chosen within the range of 0.1 ppm to 10,000 ppm, and preferably within the range of 10 ppm to 3,000 ppm.

The compounds according to the present invention in the formulation described above can control plant diseases caused by fungi in the *Oomycetes*, *Ascomycetes*, *Deuteromycetes*, and *Basidiomycetes* or other pathogenic fungi. The fungi include, but are not limited to, *Pseudoperonospora* such as cucumber downy mildew (*Pseudoperonospora cubensis*), *Phytophthora* such as tomato late blight (*Phytophthora infestans*), and *Plasmopara* such as grape downy mildew (*Plasmopara viticola*).

The agricultural or horticultural fungicide according to the present invention may be employed alone or in combination with other fungicides, insecticides, herbicides, plant growth modifiers, fertilizers or the like.

Next, the representative formulations are illustrated with reference to the following Formulation Examples, wherein all "%" represent percent by weight".

Formulation Example 1:

Fine powder

2 % of Compound No. 15, 5% of diatomaceous earth, and 93% of clay were uniformly mixed and ground into a fine powder.

Formulation Example 2:

Wettable powder

50 % of Compound No. 16, 45% of diatomaceous earth, 2% of sodium dinaphthylmethanedisulfonate, and 3% of sodium ligninsulfonate were uniformly mixed and ground into a wettable powder.

Formulation Example 3:

Emulsifiable concentrate

30 % of Compound No. 19, 20% of cyclohexanone, 11% of polyoxyethylene alkylaryl ether, 4% of calcium alkylbenzenesulfonate, and 35% of methylnaphthalene were uniformly dissolved, thus obtaining an emulsifiable concentrate.

Formulation Example 4:

Granules

5 % of Compound No. 101, 2% of sodium lauryl alcohol sulfonate, 5% of sodium lignin sulfonate, 2% of carbomethylcellulose, and 86% of clay were mixed and ground. 20% of water was added to the ground mixture. The resulting mixture was kneaded and formed into granules of 14 mesh to 32 mesh by means of an extrusion granulator, and then dried into the desired granules.

Effects of the Invention

The agricultural or horticultural fungicides according to the present invention exhibit high ability to control the growth or spread of cucumber downy mildew (*Pseudoperonospora cubensis*), tomato late blight (*Phytophthora infestans*), and grape downy mildew (*Plasmopara viticola*), and are effective for potato late blight (*Phytophthora infestans*). In addition, the agricultural or horticultural fungicides according to the present invention not only exhibit the ability to prevent fungal infection, but also exhibit the ability to eliminate pathogenic fungi after it has invaded a host plant. Furthermore, the agricultural or horticultural fungicides of the present invention are also characterized in that they are not harmful chemicals and exhibit excellent characteristics such as systemic action, residual activity, and persistence after rain-fall.

The effects of the compounds according to the present invention are now illustrated with reference to the following Test Examples. Comparative Compound X and Comparative Compound Y employed in the Test Examples are the compounds disclosed as synthesis intermediates for drugs in Japanese Patent Application, First Publication, No. Sho 62-89696. These Comparative Compounds were employed after being formulated in the same manner as the compounds of the present invention to be tested.

Comparative Compound X: N^2-tert-butoxycarbonyl-N^1-(2-phenoxyethyl)-D-alaninamide

Comparative Compound Y: N^2-tert-butoxycarbonyl-N^1-(2-phenylthioethyl)-D-alaninamide

Test Example 1:

Test on the Effect of Preventing Infection by Cucumber Downy Mildew (*Pseudoperonospora cubensis*)

Cucumber seeds (variety: "Sagami hanjiro") were sown at a rate of 10 seeds each in a square PVC (polyvinyl chloride) pot, wherein each side is 9 cm wide. The seeds were allowed to grow in a greenhouse, for 7 days, to the cotyledonous stage. A wettable powder prepared as in Formulation Example 2 was diluted with water to a concentration of 500 ppm of the active ingredient, and the aqueous preparation obtained was then applied at a rate of 10 ml per pot to the cucumber seedlings at their cotyledonous stage. After drying in the air, the plant was inoculated with a spore suspension of cucumber downy mildew (*Pseudoperonospora cubensis*) fungi using a spray and then placed in a moist chamber at 22 °C for 24 hours, and then placed in a greenhouse. On the seventh day after the inoculation, the extent of lesion was rated in accordance with the standards of evaluation as shown in Table 15 in order to secure the preventive effects of the compounds according to the present invention. The results of the test are given in Table 16.

Table 15

Standard of evaluation:	Affected area
Class A:	No lesions were observed
Class B:	Affected area is less than 25%
Class C:	Affected area is 25% or more and less than 50%
Class D:	Affected area is 50% or more

Table 16

Compound No.	Evaluation	Compound No.	Evaluation	Compound No.	Evaluation
1	B	124	B	246	B
2	A	129	A	323	A
4	A	134	A	326	A
6	A	135	A	327	A
7	A	154	A	328	A
8	B	157	A	329	B
10	A	160	A	331	B
13	A	163	A	333	B
14	B	166	A	335	A
16	A	169	A	336	A
17	A	184	A	338	A
18	B	193	A	339	A
19	A	195	B	340	A
23	A	204	B	341	A
24	B	205	A	342	A
26	B	208	A	343	A
27	A	211	A	344	B
29	A	212	A	345	A
33	A	213	A	347	A
42	A	214	A	348	B
45	A	215	A	349	A
54	B	216	A	350	B
63	A	217	A	351	A
77	A	219	A	352	A
88	A	220	A	353	A
98	A	221	A	354	A
101	A	227	A	355	A
104	A	228	A	356	A
107	A	230	A	357	B
108	A	231	A	358	A
112	A	232	A	359	A
114	A	235	A	360	A
115	A	236	A	361	B
116	A	238	A	362	B

Table 16 (continued)

Compound No.	Evaluation
363	A
364	B
365	A
366	A
367	A
368	A
369	A
370	A
371	A
372	A
373	A
374	A
376	A
377	A
378	A
379	A
380	A
381	A
382	A
383	A
385	A
386	A
387	B
388	A
389	A
390	A
391	A
392	A
393	A
394	A
395	A
397	A
399	A

Compound No.	Evaluation
401	A
402	A
403	A
405	A
408	A
410	A
411	B
412	A
413	A
414	A
416	A
417	A
418	A
419	A
421	A
422	A
423	A
424	A
425	A
426	A
427	A
429	A
430	A
431	A
432	A
439	A
440	A
451	A
452	A
453	A
455	A
456	A
462	A

Compound No.	Evaluation
465	A
466	A
467	A
468	A
47-1	B
477	A
482	A
486	A
492	A
493	A
495	A
496	B
499	B
502	A
506	A
508	A
509	A
510	A
511	A
512	A
513	A
517	A
519	A
523	A
525	A
605	A
606	A
607	A
708	A
768	A
770	A
Comparative Example X	D
Comparative Example Y	D

Test Example 2:

Test on the Effect of Treating Infection by Cucumber Downy Mildew (*Pseudoperonospora cubensis*)

Cucumber seeds (variety: "Sagami hanjiro") were sown at a rate of 10 seeds each in a square PVC (polyvinyl chloride) pot, wherein each side is 9 cm wide. The seeds were allowed to grow in a greenhouse, for 7 days, to the cotyledonous stage. The seedlings were inoculated with a spore suspension of cucumber downy mildew (*Pseudoperonospora cubensis*) fungi and then placed in a moist chamber at 22°C for 24 hours. After drying in the air, a wettable powder prepared as in Formulation Example 2 was diluted with water to a concentration of 500 ppm of the active ingredient, and the aqueous preparation obtained was then applied at a rate of 10 ml per pot to the cucumber seedlings. The seedlings were then placed in a green house. On the seventh day after the inoculation, the extent of lesions was rated in accordance with the standards of evaluation shown in Table 15 in order to secure the effect of treating with the compounds according to the present invention. The results of the test are given in Table 17.

Table 17

Compound No.	Evaluation	Compound No.	Evaluation	Compound No.	Evaluation
4	B	129	B	333	A
10	A	134	A	335	A
13	A	135	A	336	A
16	A	154	A	340	B
19	B	157	A	341	B
29	A	160	A	342	A
33	A	163	A	345	A
42	A	184	B	348	B
45	A	212	A	349	A
54	B	213	A	351	A
63	A	215	A	352	A
77	A	216	B	353	A
88	B	219	A	354	A
104	A	220	B	355	A
107	A	221	A	356	A
108	B	228	B	358	A
114	B	230	B	360	B
115	A	231	A	365	A
116	A	232	A	367	B
124	A	238	A	368	A

Table 17 (continued)

Compound No.	Evaluation
369	A
371	A
374	A
376	A
378	A
381	B
382	A
383	A
385	A
386	A
388	B
394	A
395	A
397	A
399	A
401	A
402	B
405	B
414	A
416	A

Compound No.	Evaluation
417	A
418	A
419	B
423	A
424	A
425	A
427	A
429	A
439	A
451	A
452	A
453	A
455	A
456	A
462	A
465	A
466	A
467	B
468	A
477	B

Compound No.	Evaluation
486	B
492	A
495	A
499	B
502	A
506	A
508	A
509	A
513	A
517	A
519	B
523	A
606	A
607	B
708	A
768	B
770	B
Comparative Example X	D
Comparative Example Y	D

Test Example 3:

Test on the Effect of Preventing Infection by Tomato Late Blight (*Phytophthora infestans*)

One tomato seedling (variety: "Ponterosa") was transplanted into each porcelain pot (diameter: 12 cm) and grown in a greenhouse. A wettable powder prepared as in Formulation Example 2 was diluted with water to a concentration of 500 ppm of the active ingredient, and the aqueous preparation obtained was then applied at a rate of 20 ml per pot to the tomato seedlings at their 6- or 7-leaf stage. After drying in the air, the plant was inoculated with a zoosporangium suspension of tomato late blight (*Phytophthora infestans*) fungi and then placed in a moist chamber at 22 • C. On the fourth day after the inoculation, index of incidence was determined based on the size of the affected area as shown in Table 18. The degree of damage was calculated according to Equation (1) and the index of incidence and the ability to prevent the disease (controlling activity) was calculated according to Equation (2). The results are shown in Table 19.

Table 18

Incidence Index	Affected Area
0	No lesions
1	Less than 5%
2	5% or more and less than 33.3%
3	33.3% or more and less than 66.6%
4	66.6% or more

Equation (1)

$$\text{Degree of Damage (\%)} = \frac{\Sigma\,(\text{Incidence Index} \times \text{Number of Proper Leaves})}{4 \times \text{Number of Leaves Examined}} \times 100$$

Equation (2)

$$\text{Controlling Activity (\%)} = \left(1 - \frac{\text{Degree of Damage}}{\text{Degree of Damage in Untreated Plot}}\right) \times 100$$

Table 19

Compound No.	Controlling Activity (%)	Compound No.	Controlling Activity (%)	Compound No.	Controlling Activity (%)
2	100	104	100	220	100
4	100	107	100	221	100
6	100	108	100	228	100
7	100	112	100	231	100
10	100	115	100	232	100
13	100	116	100	235	100
16	100	129	100	238	100
17	100	134	100	323	100
19	100	135	100	326	100
23	100	154	100	336	100
27	100	157	100	345	100
29	100	160	100	352	100
33	100	163	100	356	100
42	100	166	100	359	100
45	100	169	100	360	100
63	100	184	100	364	100
77	100	193	100	365	100
88	100	213	100	369	100
98	100	215	100	371	100
101	100	217	100	372	100

Table 19 (continued)

Compound No.	Controlling Activity (%)	Compound No.	Controlling Activity (%)	Compound No.	Controlling Activity (%)
373	100	403	100	477	100
374	100	404	100	482	100
378	100	405	100	492	100
379	100	408	100	495	100
380	100	414	100	502	100
381	100	417	100	508	100
382	100	418	100	509	100
386	100	423	100	513	100
388	100	424	100	519	100
390	100	427	100	523	100
391	100	430	100	605	100
393	100	439	100	606	100
394	100	440	100	607	100
395	100	451	100	708	100
397	100	462	100	768	100
399	100	465	100	770	100
401	100	466	100	Comparative Example X	0
402	100	467	100	Comparative Example Y	0

Test Example 4:

Test on the Effect of Preventing Infection by Grape Downy Mildew (*Plasmopara viticola*)

Grape rooted cuttings (variety: "Kyoho"), each grown from a cutting and pruned, was grown in a porcelain pot (diameter: 12 cm) and maintained in a greenhouse. A wettable powder prepared as in Formulation Example 2 was diluted with water to a concentration of 500 ppm of the active ingredient, and the aqueous preparation obtained was then applied at a rate of 20 ml per pot to the grape seedlings at their 4- or 5-leaf stage. After drying in the air, the plant was inoculated with a zoosporangium suspension of grape downy mildew (*Plasmopara viticola*) fungi and then placed in a moist chamber at 22 °C for 24 hours. On the seventh day in the greenhouse after the inoculation, the plant was again placed in a moist chamber at 22 °C for 24 hours to cultivate conidiospores. The incidence area where conidiospores grew on each leaf was examined and the incidence index determined according to the standards shown in Table 18. The degree of damage was calculated according to Equation (1) and the incidence index and the ability to prevent the disease (controlling activity) was calculated according to Equation (2). The results of the test are shown in Table 20.

Table 20

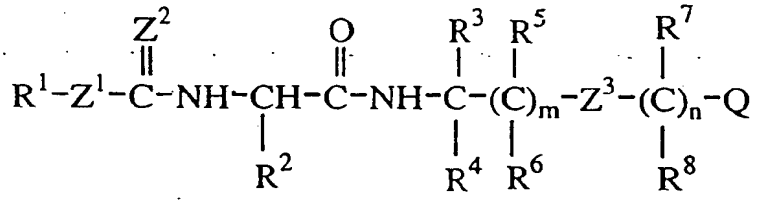
Compound No.	Controlling Activity (%)	Compound No.	Controlling Activity (%)	Compound No.	Controlling Activity (%)
2	100	104	100	220	100
4	100	107	100	221	100
6	100	108	100	228	100
7	100	112	100	231	100
10	100	115	100	232	100
13	100	116	100	235	100
16	100	129	100	238	100
17	100	134	100	323	100
19	100	135	100	326	100
23	100	154	100	336	100
27	100	157	100	345	100
29	100	160	100	352	100
33	100	163	100	356	100
42	100	166	100	359	100
45	100	169	100	360	100
63	100	184	100	364	100
77	100	193	100	365	100
88	100	213	100	369	100
98	100	215	100	371	100
101	100	217	100	372	100

Table 20 (continued)

Compound No.	Controlling Activity (%)	Compound No.	Controlling Activity (%)	Compound No.	Controlling Activity (%)
373	100	403	100	477	100
374	100	404	100	482	100
378	100	405	100	492	100
379	100	408	100	495	100
380	100	414	100	502	100
381	100	417	100	508	100
382	100	418	100	509	100
386	100	423	100	513	100
388	100	424	100	519	100
390	100	427	100	523	100
391	100	430	100	605	100
393	100	439	100	606	100
394	100	440	100	607	100
395	100	451	100	708	100
397	100	462	100	768	100
399	100	465	100	770	100
401	100	466	100	Comparative Example X	0
402	100	467	100	Comparative Example Y	0

Claims

1. An amino-acid amide derivative represented by the formula:



[I]

wherein R^1 represents

a lower alkyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom, an alkoxy group, and a cyano group),

a lower alkenyl group,

a lower alkynyl group,

a cycloalkyl group (optionally having at least one same or different substituent selected from the group consisting of methyl group and a halogen atom),

a cycloalkylalkyl group,

a cycloalkenyl group,

an alkylene oxide group,

an aralkyl group (optionally having at least one same or different substituent selected from the group consisting of a methyl group, a cyano group, and a nitro group),

a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom,

a lower alkyl group which may be substituted with a same or different halogen atom,

a lower alkoxy group which may be substituted with a same or different halogen atom,

a cyano group, and

a nitro group), or

a heterocyclic group,

R^2 represents an ethyl group, an *n*-propyl group, an isopropyl group, an isobutyl group, a *sec*-butyl group, a *tert*-butyl group, an alkenyl group, a cycloalkyl group, a phenyl group (optionally having at least one substituent of halogen atom),

R^3 represents a hydrogen atom or a lower alkyl group,

R^4 represents a hydrogen atom, a lower alkyl group, or a cyano group,

R^5, R^6, and R^7 independently represent a hydrogen atom or a lower alkyl group,

R^8 represents a hydrogen atom, a lower alkyl group, an aralkyl group, a phenyl group, an alkoxycarbonyl group, or a cyano group,

Z^1 and Z^2 independently represent an oxygen atom or a sulfur atom,

Z^3 represents

an oxygen atom,

a sulfur atom,

a group N-R^{10} (wherein R^{10} represents a hydrogen atom, a methyl group, a methylcarbonyl group, a phenylcarbonyl group, a methoxycarbonyl group, or a methoxymethyl group),

a sulfinyl group,

a sulfonyl group,

a group COO,

a group $CONR^{11}$ (wherein R^{11} represents a hydrogen atom or a lower alkyl group),

Q represents

a phenyl group [optionally having at least one same or different substituent selected from the group consisting of

a halogen atom,

a lower alkyl group which may be substituted with at least one same or different halogen atom,

a lower alkoxy group which may be substituted with a same or different halogen atom,

a cyano group,

a nitro group,

a lower alkoxycarbonyl group,

a methylsulfonyl group,

a methylsulfinyl group,

a methylthio group which may be substituted with a halogen atom,

a dimethylamino group,

a phenylsulfonyl group,

an acyl group, and

a phenyl group],

an alkylene oxide group,

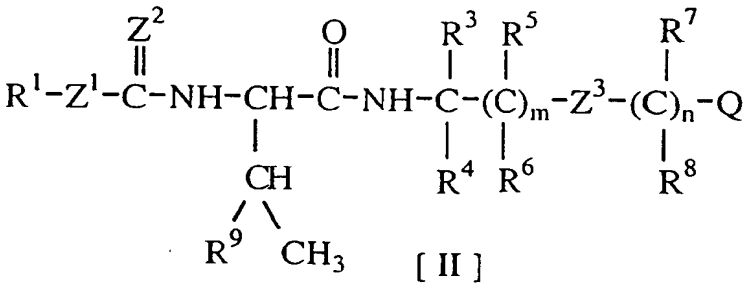
a heterocyclic group (optionally having a substituent selected from the group consisting of a halogen atom, an alkyl group, a trifluoromethyl group, and a nitro group), or

a condensed heterocyclic group optionally having a substituent selected from the group consisting of a halogen atom and a nitro group,

m represents an integer from 0 to 2, and

n represents 0 or 1.

2. An amino-acid amide derivative as recited in claim 1, which is represented by the formula:



[II]

wherein R^1 represents

a lower alkyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom and an alkoxy group),

a lower alkenyl group,

a lower alkynyl group,

a cycloalkyl group (optionally having at least one substituent of methyl group),

a cycloalkenyl group,

an alkylene oxide group,

an aralkyl group (optionally having at least one same or different substituent selected from the group consisting of a methyl group, a cyano group, and a nitro group),

a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom, a methyl group, a methoxy group, a cyano group, a trifluoromethyl group, a trifluoromethoxy group, and a nitro group),

or a heterocyclic group,

R^3 represents a hydrogen atom or a lower alkyl group,

R^4 represents a hydrogen atom, a lower alkyl group, or a cyano group,

R^5, R^6, and R^7 independently represent a hydrogen atom or a lower alkyl group,

R^8 represents a hydrogen atom, a lower alkyl group, an aralkyl group, a phenyl group, an alkoxycarbonyl group, or a cyano group,

R^9 represents a hydrogen atom, a methyl group or an ethyl group,

Z^1 and Z^2 independently represent an oxygen atom or a sulfur atom,

Z^3 represents an oxygen atom, a sulfur atom, a group N-R^{10} (wherein R^{10} represents a hydrogen atom, a methyl group, a methylcarbonyl group, a phenylcarbonyl group, a methoxycarbonyl group, or a methoxymethyl group), a sulfinyl group, or a sulfonyl group,

Q represents

a phenyl group [optionally having at least one same or different substituent selected from the group consisting of

a halogen atom,

a lower alkyl group which may be substituted with at least one halogen atom,

a lower alkoxy group which may be substituted with at least one halogen atom,

a cyano group,

a nitro group,

a lower alkoxycarbonyl group,

a methylsulfonyl group,

a methylsulfinyl group,

a methylthio group which may be substituted with a halogen atom,

a dimethylamino group,

a phenylsulfonyl group,

an acyl group, and

a phenyl group],

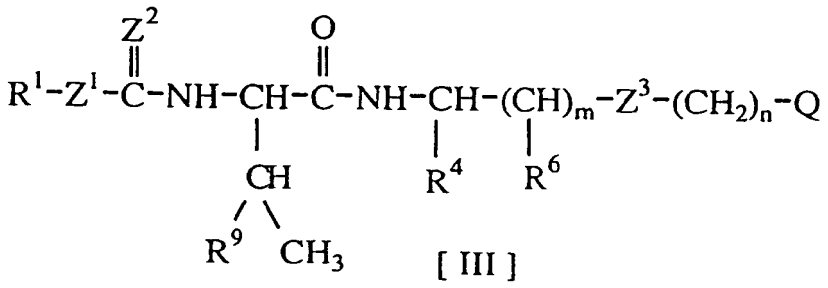
a heterocyclic group (optionally having a substituent selected from the group consisting of a halogen atom and a nitro group), or

a condensed heterocyclic group (optionally having a substituent selected from the group consisting of a halogen atom and a nitro group),

m represents an integer from 0 to 2, and

n represents 0 or 1.

3. An amino-acid amide derivative as recited in Claim 1, which is represented by the formula:



[III]

wherein R^1 represents a C_1 ~ C_6 alkyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom and an alkoxy group), a C_2 ~ C_6 alkenyl group, a C_2 ~ C_6 alkynyl group, a C_3 - C_8 cycloalkyl group (optionally having at least one substituent of methyl group), a C_2 ~ C_8 alkylene oxide group, a C_7 ~ C_8 aralkyl group (optionally having at least one substituent of methyl group), or a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom, a methyl group, a methoxy group, a trifluoromethyl group, a trifluoromethoxy group, and a nitro group),

R^4 represents a hydrogen atom, a C_1 ~ C_3 alkyl group, or a cyano group,

R^6 represents a hydrogen atom or a C_1 ~ C_3 alkyl group,

R^9 represents a hydrogen atom, a methyl group, or an ethyl group,

Z^1 and Z^2 independently represent an oxygen atom or a sulfur atom,

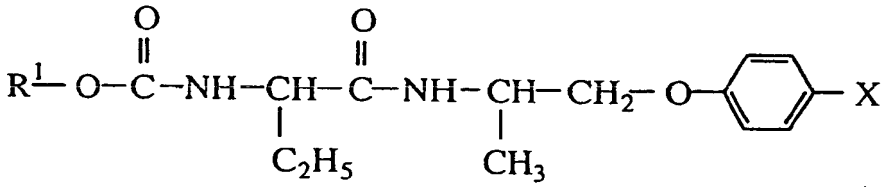
Z^3 represents an oxygen atom, a sulfur atom, a group N-R^{10} (wherein R^{10} represents a hydrogen atom, a methyl group, a methylcarbonyl group, or a phenylcarbonyl group), a sulfinyl group, or a sulfonyl group,

Q represents a phenyl group [optionally having at least one same or different substituent selected from the group consisting of a halogen atom, a C_1 ~ C_3 alkyl group which may be substituted with at least one same or different halogen atom, a C_1 ~ C_3 alkoxy group which may be substituted with a same or different halogen atom, a cyano group, a nitro group, a methylsulfonyl group, a methylsulfinyl group, and a methylthio group], a pyrimidinyl group, or a pyridyl group which may be substituted with a halogen atom,

m represents 1 or 2, and

n represents 0 or 1.

4. An amino-acid amide derivative as recited in Claim 1, which is represented by the formula:

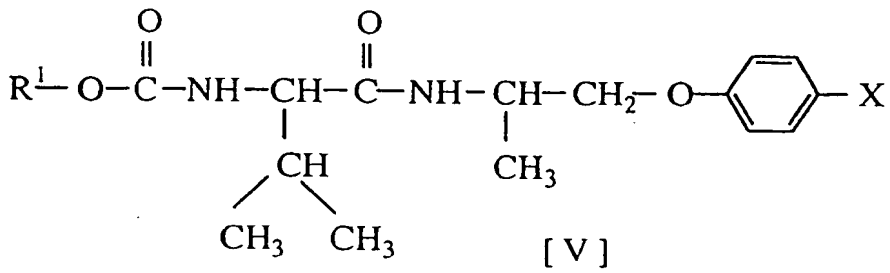


[IV]

wherein R^1 represents an isopropyl group, a *tert*-butyl group, a cyclopentyl group, or a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom, a methyl group, a methoxy group, and a nitro group), and

X represents a halogen atom, a cyano group, or a nitro group.

5. An amino-acid amide derivative as recited in Claim 1, which is represented by the formula:

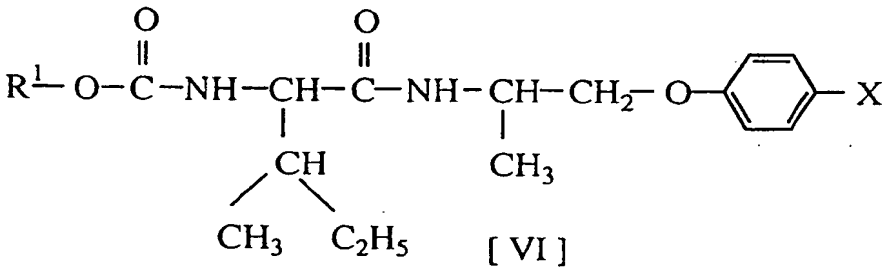


[V]

wherein R^1 represents an isopropyl group, a *tert*-butyl group, a cyclopentyl group, a cyclohexyl group which may be substituted with a methyl group, or a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom, a methyl group, a methoxy group, a trifluoromethyl group, a trifluoromethoxy group, and a nitro group), and

X represents a halogen atom, a cyano group, or a nitro group.

6. An amino-acid amide derivative as recited in Claim 1, which is represented by the formula:

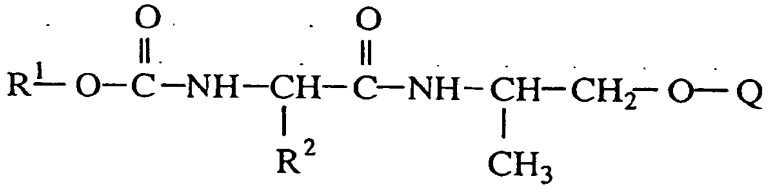


[VI]

wherein R^1 represents an isopropyl group, a *tert*-butyl group, a cyclopentyl group, or a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom, a methyl group, a methoxy group, and a nitro group), and

X represents a halogen atom, a cyano group, or a nitro group.

7. An amino-acid amide derivative as recited in Claim 1, which is represented by the formula:



[VII]

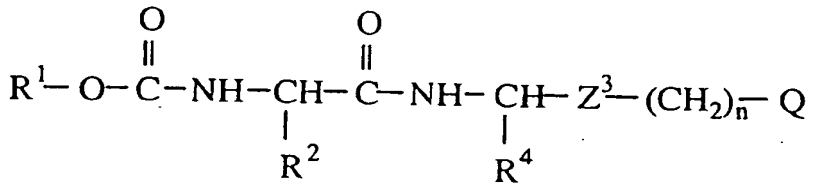
wherein R^1 represents a $C_1 \sim C_6$ alkyl group (optionally having at least one substituent cyano group), a $C_3 \sim C_8$ cycloalkyl group (optionally having at least one substituent halogen atom), a $C_4 \sim C_8$ cycloalkyl $C_1 \sim C_3$ alkyl group, a benzyl group or a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom and a difluoromethoxy group),

R^2 represents an *n*-propyl group, an isopropyl group, an isobutyl group, a *tert*-butyl group, an isopropenyl group, a $C_3 \sim C_8$ cycloalkyl group, or a phenyl group (optionally having at least one

substituent halogen atom), and

Q represents a phenyl group (optionally having at least one substituent cyano group), a pyridyl group (optionally having at least one substituent of a same or different halogen atom or trifluoromethyl group), or a pyrimidinyl group (optionally having at least one same or different substituent of a halogen atom or $C_1 \sim C_3$ alkyl group).

8. An amino-acid amide derivative as recited in Claim 1, which is represented by the formula:



[VIII]

wherein R^1 represents a $C_1 \sim C_6$ alkyl group, a $C_3 \sim C_8$ cycloalkyl group, or a phenyl group (optionally having at least one substituent halogen atom),

R^2 represents an ethyl group, an isopropyl group, or a *sec*-butyl group,

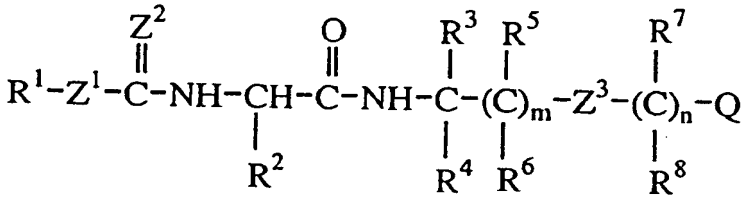
R^4 represents a hydrogen atom or a $C_1 \sim C_3$ alkyl group,

Z^3 represents a group COO, a group $CONR^{12}$ (wherein R^{12} represents a hydrogen atom or a $C_1 \sim C_3$ alkyl group),

Q represents a phenyl group (optionally having at least one same or different substituent selected from the group consisting of a halogen atom, a $C_1 \sim C_3$ alkyl group, a $C_1 \sim C_3$ alkoxy group, and a cyano group), and

n represents 0 or 1.

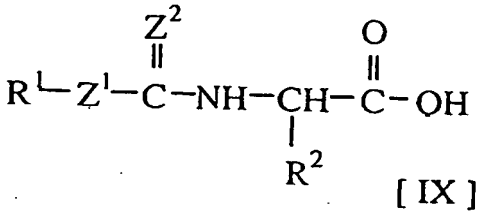
9. A process for preparing an amino-acid amide derivative represented by the formula:



[I]

wherein R^1, R^2, R^3, R^4, R^5, R^6, R^7, R^8, Z^1, Z^2, Z^3, Q, m, and n have the same meanings as defined in Claim 1,

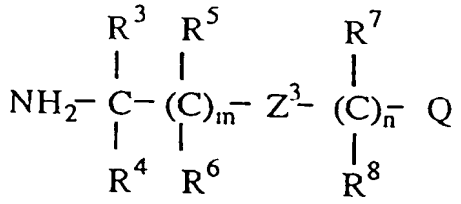
comprising the step of: reacting a compound represented by the formula:



[IX]

wherein R^1, R^2, Z^1, and Z^2 have the same meanings as defined in Claim 1,

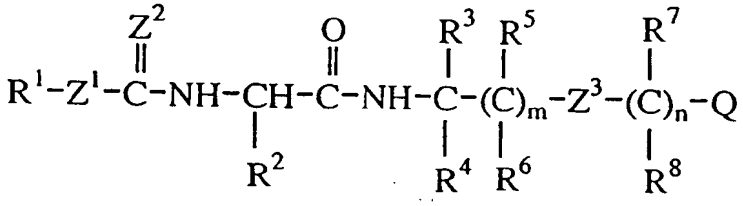
with a compound represented by the formula:



[X]

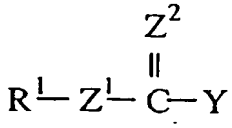
wherein R^3, R^4, R^5, R^6, R^7, R^8, Z^3, Q, m, and n have the same meanings as defined in Claim 1.

10. A process for preparing an amino-acid amide derivative represented by the formula:



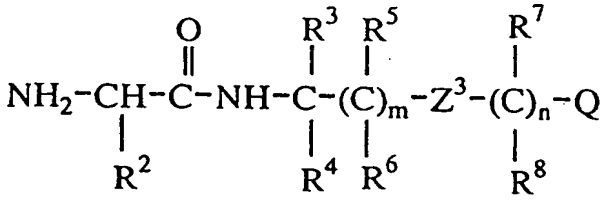
[I]

wherein R^1, R^2, R^3, R^4, R^5, R^6, R^7, R^8, Z^1, Z^2, Z^3, Q, m, and n have the same meanings as defined in Claim 1,
comprising the step of: reacting a compound represented by the formula:



[XI]

wherein Y represents a halogen atom, a 4,6-dimethylpyrimidinylthio group, a group $R^1OC(O)O$-, or a group -ON=C(CN)Ph (wherein Ph represents a phenyl group), and R^1, Z^1, and Z^2 have the same meanings as defined in Claim 1,
with a compound represented by the formula:



[XII]

wherein R^2, R^3, R^4, R^5, R^6, R^7, R^8, Z^3, Q, m, and n have the same meanings as defined in Claim 1.

11. An agricultural or horticultural fungicidal composition which includes an effective amount of an amino-acid amide derivative as recited in Claim 1.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/JP94/00708

A. CLASSIFICATION OF SUBJECT MATTER
Int. Cl[5] C07C271/22, 271/32, 271/54, 317/14, 327/20, C07D213/28, 239/32, 307/04, 307/40, 307/77, 333/14, 333/50, A01N47/12
According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
Int. Cl[5] C07C271/22, 271/32, 271/54, 271/06, 327/20, 317/14, C07D213/28, 239/32, 307/04, 307/40, 307/77, 333/14, 333/50,

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)
CAS ONLINE

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	JP, A, 4-338372 (Bayer AG.), November 25, 1992 (25. 11. 92), Pages 1 to 3 & EP, A2, 496,239	1-11
A	JP, A, 4-283554 (Bayer AG.), October 8, 1992 (08. 10. 92), Pages 1 to 3 & EP, A1, 485,794	1-11
A	JP, A, 4-308507 (American Cyanamid Co.), October 30, 1992 (30. 10. 92), Pages 1 to 3 & EP, A1, 493,683	1-11
A	JP, A, 4-230653 (Bayer AG.), August 19, 1992 (19. 08. 92), Pages 1 to 2 & EP, A, 477,639	1-11
A	JP, A, 4-230652 (Bayer AG.), August 19, 1992 (19. 08. 92), Pages 1 to 2 & EP, A1, 550,788	1-11

☐ Further documents are listed in the continuation of Box C. ☐ See patent family annex.

* Special categories of cited documents:
"A" document defining the general state of the art which is not considered to be of particular relevance
"E" earlier document but published on or after the international filing date
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
"O" document referring to an oral disclosure, use, exhibition or other means
"P" document published prior to the international filing date but later than the priority date claimed
"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"&" document member of the same patent family

Date of the actual completion of the international search	Date of mailing of the international search report
July 7, 1994 (07. 07. 94)	July 26, 1994 (26. 07. 94)
Name and mailing address of the ISA/ Japanese Patent Office Facsimile No.	Authorized officer Telephone No.

Form PCT/ISA/210 (second sheet) (July 1992)

INTERNATIONAL SEARCH REPORT

International application No.
PCT/JP94/00708

B(Continuation). FIELDS SEARCHED

A01N47/12, 47/10

Form PCT/ISA/210 (extra sheet) (July 1992)